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## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ ANC/19	<b>Name:</b> Analytical Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester, the students will be delivered two written tests each of maximum 25 points. To be allowed for the oral part of the examination, the students will have to gather at least 25 points from both tests (i.e. 50% of the total possible count). The maximum number of points obtainable at the oral part of the exam is 50. The final classification is obtained from the sum of both parts of the examination – written and oral. For the final classification to be A one has to obtain 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> Upon completing the Course the students acquire theoretical knowledge about the basics qualitative and quantitative analysis and is able to apply selected analytical methods for the analysis of inorganic and organic substances.	
<b>Brief syllabus:</b> 1. Introduction – the concept of analytical reactions, electrolytic dissociation, water as solvent. 2. Chemical equilibrium – the concept of equilibrium, equilibrium constants, strong and weak electrolytes, relation between thermodynamics and equilibrium constants. 3. Acidobasic reactions – theory of acids and bases, calculating the pH of strong and weak acids, bases and salts, buffers. 4. Precipitation reactions – calculating the solubility of moderately soluble substances, decrease of solubility by own ions, effect of foreign ions on the solubility. 5. Redox reactions – equilibrium of redox reactions, determination of equilibrium constants, factors controlling the redox equilibria. 6. Complex reactions as analytical reactions, catalytical induces reactions. 7. Reactions of organic reagents. 8. The process of chemical analysis group reactions of cations and anions, selective reactions of cations and anions. 9. Qualitative analysis of organic substances – qualitative elemental analysis (C, H, N, S, halogens and metals). 10. Qualitative analysis of organic substances – proof of functional groups. 11. Overview of selected spectral methods.	

12. Chemometrical evaluation of analytical results and calibration functions. Interpretation and presentation of results.  
13. Conclusion.

**Literature:**

Karlíček R., a kol. (2009): Analytická chemie pro farmaceuty. Karolinum, ISBN 97 8802 46 1453 3

Majer J., (1989) : Analytická chemia. - 1. vyd. - Martin : Osveta n.p., - 368 s.

Holzbecher Z., Churáček J., (1987) : Analytická chemia. - 1. vyd. – Praha, SNTL - Nakladatelství technické literatury, - 663 s.

Barcza L., (2006): A mennyiségi kémiai analízis gyakorlati kézikönyve. Medicina Kiadó, ISBN: 963 2429 61 3

Barcza L., (2007): Kvantitatív analitikai kémia. Budapest, Semmelweis Kiadó, ISBN 978 963 9656 73 4

Barcza L., Buvári Á., (2009): A minőségi kémiai analízis. Medicina Könyvkiadó, ISBN 978 9 6 322 6186 7

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 40

A	B	C	D	E	FX
15.0	22.5	22.5	10.0	20.0	10.0

**Teacher:** doc. Ing. Ondrej Hegedűs, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ ARC/15	<b>Name:</b> Inorganic Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester, the students will be delivered two written tests each of maximum 25 points. To be allowed for the oral part of the examination, the students will have to gather at least 25 points from both tests (i.e. 50% of the total possible count). The maximum number of points obtainable at the oral part of the exam is 50. The final classification is obtained from the sum of both parts of the examination – written and oral. For the final classification to be A one has to obtain 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> The students master the basic laws and principles in Inorganic Chemistry and are able to apply the systematic knowledge about non-metallic and metallic elements including their compounds.	
<b>Brief syllabus:</b> Periodic system of elements and the electron structure of their valence shells, Chemistry of non-transitional, transitional and internally transitional elements. Coordinational compounds. <ol style="list-style-type: none"> <li>1. The periodic system of elements and the electron structure of their valence shells.</li> <li>2. Compounds in general, lattice and bond types, characteristics and categories of compounds – hydrides, halogenides, oxides, peroxides, superoxides, oxoacids, sulphides, nitrides, phosphides, carbides, silicides, borides, cyanides, cyanates.</li> <li>3. Hydrogen, bond types, occurrence, preparation, its compounds and isotopes.</li> <li>4. General properties of metals (including transition metals).</li> <li>5. Coordination compounds.</li> <li>6. Alkali metals – elements of group I of the periodic system, bond types, compounds, the subgroup of copper.</li> <li>7. Alkaline earth metals – elements of group II of the periodic system, bond types, compounds, the subgroup of zinc.</li> <li>8. Hybridization.</li> <li>9. Elements of group III of the periodic system, bond types, compounds, the subgroup of scandium, hybridization types.</li> <li>10. Elements of group IV of the periodic system, bond types, compounds, the subgroup of titanium.</li> <li>11. Elements of group V of the periodic system, bond types, compounds, the subgroup of vanadium.</li> </ol>	

12. Elements of group VI of the periodic system, bond types, compounds, the subgroup of chromium.
13. Elements of group VII of the periodic system, bond types, compounds, the subgroup of manganese.
14. Elements of group VIII of the periodic system and their compounds.

**Literature:**

- Greenwood N. N., Earnshaw A., (1993): Chemie prvků I a II. ISBN 80-85427-38-9
- Krätsmár - Šmogrovič J. a kol., (2007): Všeobecná a anorganická chémie. Osveta, ISBN 80 806 3245 8
- Fajnor V., (1998) : Všeobecná a anorganická chémie. - 1. vyd. – Bratislava, Univerzita Komenského - 266 s. - ISBN 80-223-1257-6
- Gažo J., Kohout J., Serátor M., (1981) : Všeobecná a anorganická chémie. Bratislava, ALFA - 804 s.
- Lukeš I., (2009): Systematická anorganická chémie. - 1. vyd. – Praha, Nakladatelství Karolinum - 230 s. ISBN 978-80-246-1614-8
- Zikmund M.,(1995): Anorganická chémie. Bratislava : Univerzita Komenského, ISBN 80-223-0919-2
- Bánhidi L., (1989): Szervetlen kémia. Budapest, Tankönyvkiadó, ISBN 96 318 2192 7
- Fehér D., (1987): Szervetlen kémia. Budapest, Tankönyvkiadó, ISBN 96 318 0282 5

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 47

A	B	C	D	E	FX
36.17	17.02	12.77	17.02	10.64	6.38

**Teacher:** doc. RNDr. Róbert Gyepes, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ BC1/15	<b>Name:</b> Biochemistry I.
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester 2 writing tests are compulsory: the maximum points are $2 \times 25 = 50$ . The minimum eligibility requirement for the oral exam is overall 25 points from the two writing tests. The maximum points at the oral exam are 50. The final evaluation comprises both the writing test and oral exam (maximum points $50 + 50 = 100$ ). Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> During the pedagogical education the students will study the basic biochemical processes of the living systems	
<b>Brief syllabus:</b> 1. Alcohols and oxo compounds (aldehydes, ketones). Physical and chemical properties, structure, synthesis, and reactions. 2. Carboxylic and nucleic acids and heterocyclic compounds. Synthesis and structure determination. 3. Amino acids: properties, structure, and optical activity. The isoelectric point. Characterization of the proteinogenic amino acids. Essential amino acids. 4. Peptides. Formation and structure of the peptide bond. Biologically important peptides. 5. Proteins. Structure and characterization of proteins. Biological roles of proteins. 6. Enzymes. Structure of the enzymes; the active center. Biological roles of enzymes. 7. The mechanism of the enzyme action. The Michaelis–Menten equation. The Michaelis-constant. Characterization of the inhibitors. 8. Coenzymes. 9. Lipids. Hydrolysable and non-hydrolysable lipids. Structure and biological roles. 10. Chemical composition of the cell membrane. Types of membrane-transport processes. 11. Writing test	
<b>Literature:</b> Ferenčík, M. a kol. Biochémia. Bratislava : Slovak Academic Press, 2000. Karlubík, M.: Biochémia. Nitra: VŠP, 1990. Kiss T., Bevezetés a bioszervetlen kémiába. Nemzeti Tankönyvkiadó Zrt. ISBN: 978 963 195 999 4	

Lásztity, Radomír: Biokémia. Nemzeti Tankönyvkiadó, 1995. ISBN 9631865657  
Škárka, B.: Biochémia. Alfa Bratislava, 1987  
Vodrážka, Z. a kol.: Biochemie, Akademia, 2007. ISBN 8020006001

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 14

A	B	C	D	E	FX
50.0	35.71	0.0	0.0	14.29	0.0

**Teacher:** Mgr. Alexandra Hengerics Szabó, PhD.

**Date of last update:** 12.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ BC2/15	<b>Name:</b> Biochemistry II.
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester 2 writing tests are compulsory: the maximum points are $2 \times 25 = 50$ . The minimum eligibility requirement for the oral exam is overall 25 points from the two writing tests. The maximum points at the oral exam are 50. The final evaluation comprises both the writing test and oral exam (maximum points $50 + 50 = 100$ ). Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> By studying this subject, the students will study the basic biochemical processes of the living systems. Student will get a basic overview on those chemical processes which take place in living organisms. He/she will be able to recognise the interdisciplinary relationship between chemistry and biology.	
<b>Brief syllabus:</b> 1. Categories and biological roles of the carbohydrates. Monosaccharides. Constitution and configuration. The optical activity. The Fischer projection, the Tollens lactol formation, and the Haworth formula. Oxidation and reduction of monosaccharides. Oligo- and polysaccharides. 2. The nucleic acids. Nucleosides and nucleotides. Categories of the nucleic acids. The primary and secondary structures of the nucleotides. The DNA double helix. 3. The biochemical processes in the living systems. The characterization and importance of the redox reactions. Bioenergetics. The citric acid cycle. 4. The oxidative phosphorylation. 5. Written test 6. Metabolism of the saccharides. Anabolism of the saccharides: photosynthesis, steps of the photosynthesis. 7. Catabolism of the saccharides: glycolysis under aerobic and anaerobic conditions. 8. Metabolism and hydrolysis of the lipids. Degradation of the fatty acids. Biosynthesis of the fatty acids and the lipids. 9. The natural nitrogen cycle. Metabolism of the proteins —anabolism and catabolism. The urea (ornithine) cycle. 10. The regulation mechanisms in the living systems.	

11. Written test					
<b>Literature:</b> Ferenčík, M. a kol. Biochémia. Bratislava : Slovak Academic Press, 2000. Karlubík, M.: Biochémia. Nitra: VŠP, 1990. Kiss T., Bevezetés a bioszervetlenkémiába. Nemzeti Tankönyvkiadó Zrt. ISBN: 978 963 195 999 4 Lásztity, Radomír: Biokémia. Nemzeti Tankönyvkiadó, 1995. ISBN 9631865657 Škárka, B.: Biochémia. Alfa Bratislava, 1987 Vodrážka, Z. a kol.: Biochemie, Akademia, 2007. ISBN 8020006001					
<b>Language, knowledge of which is necessary to complete a course:</b>					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 11					
A	B	C	D	E	FX
54.55	9.09	9.09	27.27	0.0	0.0
<b>Teacher:</b> Mgr. Alexandra Hengerics Szabó, PhD.					
<b>Date of last update:</b> 12.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ CHV/15	<b>Name:</b> Calculations in Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester the students will be issued a test of maximum 50 points, while another amount of 50 points can be acquired for his/her homework. For a successful completion of the course one has to gather at least 50 point, i.e. 50% of the total points possible. For the final classification to be A one has to acquire 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> Completing the Course the students acquire skills in selected chemical calculations and get acquainted with the mathematical apparatus needed in chemical calculations, which can be later applied in their further pedagogical career upon tackling common laboratory operations.	
<b>Brief syllabus:</b> 1. Calculations based on chemical equations. 2. Calculation of pureness and yield of chemical reactions. 3. Gas laws. Ideal gases. 4. Chemical reactions and redox processes. Balancing redox reactions. 5. Electrochemistry – Faraday’s laws, chemical equilibrium of redox systems. 6. Thermochemistry – enthalpy of formation, reaction enthalpy, thermochemical laws. 7. Equilibrium in electrolytes – dissociation of acids and bases. 8. Equilibrium in electrolytes – water dissociation and the hydrogen exponent. 9. Buffers. 10. Hydrolysis of salts. 11. Written test. Conclusion.	
<b>Literature:</b> Krätzmár-Šmogrovič, J. a kol., (2007): Všeobecná a anorganická chémia. Osveta, ISBN 80 806 3245 8 Fajnor V.,(1992) Laboratórna technika, názvoslovie a chemické výpočty. Vysokoškolské skriptá, UK Bratislava, ISBN 80 223 0436 0 Sokolík J., (2012) Názvoslovie a príprava vybraných anorganických látok, UK Bratislava, ISBN 978 80 223 2913 2	

Kotočová A, Valigura D.(1993): Všeobecná chémia- Návody na laboratórne cvičenia. Bratislava: STU, ISBN 80 227 0560 8  
 Csányi C., (2002): Kémiai példatár és tesztgyűjtemény megoldásokkal. Budapest, ISBN 96 31 6211 2 X  
 Kiss Zs., (2004): Összefoglaló feladatgyűjtemény kémiából - Megoldások. Budapest, Nemzeti Tankönyvkiadó, ISBN 963 19 5394 7  
 Mayer J., (2002): Módszertani stratégiák 4. Országos Közoktatási Intézet, ISBN 9636825033

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 43

A	B	C	D	E	FX
23.26	20.93	20.93	11.63	16.28	6.98

**Teacher:** Mgr. Katarína Szarka, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ DCH/15	<b>Name:</b> History of Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester 1 writing test is compulsory: the maximum points are 50. Moreover, another 50 points are available from the (maximum points 50 + 50 = 100). Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> During their studies, students will be acquainted, in chronological order, with the development of chemical science during our history. They will be able to apply this knowledge in practice during their chemical class in the future.	
<b>Brief syllabus:</b> 1. Introductory 2. Born of chemistry as a science 3. Chemistry in the age of the ancient Greek and Roman Empire 4. The age of alchemy 5. Chemistry, as a branch of science 6. Development of the chemical science in the 17th century. The flogiston theory. 7. Birth of the modern chemistry 8. Development of the chemistry in the 19th century 9. Birth and development of chemical industry 10. Discovery of radioactivity. Its importance and impact on the development of chemistry in the 20th century. 11. Famous chemists and their discoveries 12. Nobel Laurates in chemistry 13. Writing test	
<b>Literature:</b> Linkešová, M., (2010): Kapitoly z histórie chémie 2. prepracované vydanie. – Trnava, Pedagogická fakulta Trnavskej univerzity v Trnave, 145s. - ISBN 978-80-8082-399-3, dostupné online: <a href="http://katchem.truni.sk/prilohy/Kapitoly%20z%20historie%20chemie.pdf">http://katchem.truni.sk/prilohy/Kapitoly%20z%20historie%20chemie.pdf</a>	

Cídlová,H. et al , (2011) : Historie chemie. Studijní materiál je určen pro studenty volitelného předmětu Historie chemie. Je součástí řešení projektu FR VŠ 464/2011. dostupné online: <http://www.ped.muni.cz/wchem/sm/hc/hist/default.htm>

Balázs, L., (1996): A kémia története I-II. Budapest, Nemzeti Tankönyvkiadó,1075s., - ISBN 963-18-7344-7.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 32

A	B	C	D	E	FX
84.38	6.25	6.25	0.0	0.0	3.13

**Teacher:** Dr. habil. PaedDr. György Juhász, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ ENC/15	<b>Name:</b> Environmental Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester the students will be issued a test of maximum 50 points, while another amount of 50 points can be acquired for his/her homework. For a successful completion of the course one has to gather at least 50 point, i.e. 50% of the total points possible. For the final classification to be A one has to acquire 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> With succesful fullfilment the students will known the basic phrases from the area of ecology and protection of environment. Besides they acquire theoretical bases, and able to understand the relationship between chemistry and environment, they will be able to solve practical problems within the theme.	
<b>Brief syllabus:</b> 1. Introduction – biosphere, man and his environment 2. Each basic school and secondary school subjects, especially the role of chemistry in student’s enviromental education 3. Atmosphere and the air pollution 4. Water and the water pollution 5. The soil and soil protection 6. The wastewater treatment, reducing of the aerospace pollution 7. Radioecology – nuclear power stations and the environment. 8. Waste – waste management, recycling. 9. Environmental monitoring. 10. Environmental chemistry experiments – water. 11. Environmental chemistry experiments – air. 12. Environmental chemistry experiments – soil. 13. The current state of enviromental education and its‘ perspectives. The concept of natural environment and the characterization of the actual state of environment in Slovakia. The pollution of aerospace, water and soil. Radioactivity and the protection of environment – Application of the acquired knowledge in education of chemistry in elementary school and secondary school.	
<b>Literature:</b>	

With successful fulfillment the students will know the basic phrases from the area of ecology and protection of environment. Besides they acquire theoretical bases, and able to understand the relationship between chemistry and environment, they will be able to solve practical problems within the theme.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 32

A	B	C	D	E	FX
50.0	28.13	18.75	3.13	0.0	0.0

**Teacher:** Mgr. Andrea Vargová, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ FC1/15	<b>Name:</b> Physical Chemistry I.
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 / 1 <b>For the study period:</b> 26 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester there will be two written assessments for 25-25 points, in order to access the oral test, students have to achieve at least 25 points from the two assessments together, i.e. 50% of the total. For the oral examination the student can get 50 points. The final evaluation result depends on the oral exam and on the written assessments (50%-50%). To achieve evaluation A 90-100% is needed, for evaluation B 80-89% is needed, for evaluation C 70-79% is needed, for evaluation D 60-69% is needed, and for evaluation E 50-59% is needed from the total number of points.	
<b>Results of education:</b> By completing this course, students acquire basic knowledge about the structure and description of the Solids, Liquids, and Gases. Based on the laws of thermodynamics, students describe and explain the phenomena accompanying the physico-chemical and chemical processes. They will be able to explain laws, and they acquire the necessary skills to characterize and analyze the characteristics of mixtures. Students will be able to apply the acquired theoretical knowledge on the practical lessons of physical chemistry.	
<b>Brief syllabus:</b> 1. Equations of State and the Ideal Gas Law, State Functions and Path Functions Kinetic Theory of Gase 2. Physical Meaning of the Boltzmann Distribution Law, Boltzmann and Maxwell Distribution, Real Gases. 3. The Law of Corresponding States , Liquids, Surface Tension and Viscosity, Solids 4. Thermodynamics, Heat, Work, Internal Energy, Expansion and Compression of an Ideal Gas 5. First Law of Thermodynamics, Enthalpy, Heat Capacity, Adiabatic Changes 6. Thermochemistry 7. Written assessment. 8. II. Law of Thermodynamics, Entropy, Carnot cycle. 9. The Gibbs Energy and the Helmholtz Energy, Fugacity and the Equilibrium Constant for Real Gases. 10. Ideal and Real Solutions, The Chemical Potential, The Gibbs and Duhem Equation 11. Phases Equilibrium, Gibbs' Phase Rule, The Clapeyron Equation	

12. Raoult's and Henry's Law, Phase Diagrams  
 13. Colligative Properties, Phase Diagrams of Condensed Systems.  
 14. Written assessment.

**Literature:**

Atkins, P.W.: Fizikai kémia I-III. a tankönyvi feladatok megoldására. Tankönyvkiadó, 1991. ISBN 9631843505  
 Atkins, P. W.: Fizikai kémia I. Egyensúly. Budapest: Nemzeti Tankönyvkiadó, 2002. ISBN: 9631933148  
 Atkins, P. W.: Fizikai kémia II. Szerkezet. Budapest: Nemzeti Tankönyvkiadó, 2002. ISBN: 963192145X  
 Biskupič S., Kellő V., Staško A., Vavra J., (1991) : Fyzikálna chémia I. - 1. vyd. - Bratislava ALFA - 296 s. - ISBN 80-05-00931-3  
 Brdička R., (1977): Základy fyzikální chemie. Praha, ACADEMIA  
 Číperka J., (1990): Fyzikálna chémia. Bratislava: Osveta, ISBN 80 217 0134 x  
 Ulický L., Vavra J., (1992) : Návody do cvičenia z fyzikálnej chémie. - 1. vyd. – Bratislava, SVŠT v Bratislave - 216 s.  
 Ulický L., a kol., (1972) : Štruktúra tuhej fázy. - 1. vyd. – Bratislava, SVŠT v Bratislave- 130 s.  
 Ulický L., Fyzikálna chémia I., FPV UCM, 1999

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 33

A	B	C	D	E	FX
6.06	30.3	24.24	21.21	18.18	0.0

**Teacher:** prof. Róbert Mészáros, DSc.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ FC2/15	<b>Name:</b> Physical Chemistry II.
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 / 1 <b>For the study period:</b> 26 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester there will be two written assessments for 25-25 points, in order to access the oral test, students have to achieve at least 25 points from the two assessments together, i.e. 50% of the total. For the oral examination the student can get 50 points. The final evaluation result depends on the oral exam and on the written assessments (50%-50%). To achieve evaluation A 90-100% is needed, for evaluation B 80-89% is needed, for evaluation C 70-79% is needed, for evaluation D 60-69% is needed, and for evaluation E 50-59% is needed from the total number of points.	
<b>Results of education:</b> By completing this course, students acquire basic knowledge about the chemical balance in the chemical and electrochemical systems. They can control the conduction of electricity in electrolyte solutions, they are able to explain. In addition to the above mentioned things, students are able to understand the reaction rates of simple and complex chemical reactions, and the basic principles of colloid chemistry.	
<b>Brief syllabus:</b> 1. Chemical Equilibrium, The Equilibrium Constant for a Mixture of Ideal Gases, The Variation of $K_p$ with Temperature and pressure, Le Chatelier's Rule. 2. Electrolyte Solutions, Thermodynamics of Ion Formation and Solvation 3. Chemical Equilibrium in Electrolyte Solutions, Ostwald's Rule 4. Hydrolysis of salts, Buffer solutions 5. Conduct Electricity in electrolytic solutions, Faraday's Law, Conductivity, 6. Written assessment 7. Electrochemical Cells, Batteries, The, Electrodes and Electrode potential 8. Chemical Kinetics, Rate Laws, Reaction rates 9. Zero - First-, Second-, Third- Order Reactions 10. Determination of Reaction Order, Reaction Mechanisms 11. Temperature Dependence of Rate Constants, Activated Complex Theory, The Collision Theory of Reaction rates. 12. Catalysis, Photochemistry, Diffusion, 13. Colloids, Solutions, and Mixtures, Adsorption	

14. Written assessment.

**Literature:**

Ulický L., a kol.(1999): Fyzikálna chémia I., FPV UCM  
Atkins P.W., (1991) : Fizikai kémia I-III. a tankönyvi feladatok megoldására. Tankönyvkiadó, ISBN 96 318 4350 5  
Atkins P. W., (2002): Fizikai kémia I. Egyensúly. Budapest: Nemzeti Tankönyvkiadó, ISBN: 96 319 3314 8  
Atkins P. W.,(2002): Fizikai kémia II. Szerkezet. Budapest: Nemzeti Tankönyvkiadó, ISBN: 96 319 2145 X  
Atkins P.W.,(1999): Fyzikálna chémia, STU Bratislava, 6. vyd. ISBN 80 227 1238 8  
Biskupič S., Kellö V., Staško A., Vavra J., (1991) : Fyzikálna chémia I. - 1. vyd. - Bratislava ALFA - 296 s. - ISBN 80-05-00931-3  
Brdička R., (1977): Základy fyzikální chemie. Praha, ACADEMIA  
Čipera J., (1990): Fyzikálna chémia. Bratislava: Osveta, ISBN 80 217 0134 x  
Ulický L., a kol. (1972) : Štruktúra tuhej fázy. - 1. vyd. – Bratislava, SVŠT v Bratislave- 130 s.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 32

A	B	C	D	E	FX
9.38	18.75	31.25	28.13	12.5	0.0

**Teacher:** prof. Róbert Mészáros, DSc.

**Date of last update:** 12.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ FPC/19	<b>Name:</b> Physics for Chemists
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester two written clearance by 25 points will be held and at the end of the semester an oral exam will be held, where the student can get 50 points, while condition of access to oral test will achieve the two checks totaling at least 25 points To obtain grade A it is necessary to get altogether at least 90 points, for grade B at least 80 points, for grade C to get at least 70 points, for grade D to get at least 60 points and for grade E at least 50 points.	
<b>Results of education:</b> By completing the course students acquire basic knowledge of physics in mechanics, thermodynamics, electromagnetism and nuclear physics	
<b>Brief syllabus:</b> 1. Introduction. The purpose and content of physics. Relationship of physics to other sciences. Physical quantities. Units of physical quantities. 2. Measurement and measurement errors. 3. Motion. The concept of mass point. Relativity of motion. Track and trajectory. Motion in one-dimensional space. Track and speed of motion. Medium speed. Instant speed. Acceleration. Uniform motion on a straight line. Straightforward uneven movement. Uniformly accelerated motion. Free fall. 4. Horizontal and projectile throw- discharge. Uniform circular motion. Dynamics. 5. Newton's laws of motion. Strength. I. Newton's laws of motion. Newton II. laws of motion. Newton III. laws of motion. The force of gravity, weight, normal force. Applications. The balance of bodies. 6. The second current written proof of knowledges. 7. The friction, circular motion and other applications. Mechanical energy and its conservation. Mass and energy. Power and efficiency. Power. Atmospheric pressure. Archimedes principle. Fluid flow. Surface effects in liquids. 8. Thermodynamics. Heat, temperature, thermodynamic equilibrium. The equation of state. 1st and 2nd law of thermodynamics. Heat engine and its effectiveness. Applications. 9. The transmission of heat, diffusion. Electromagnetism - basic concepts, electric field, potential, voltage, work, energy.	

10. Electrical circuits, electric current, resistor, capacitor. Power.
11. The magnetic field and its basic features.
12. Electromagnetic induction, alternating current, transformer.
13. Solar energy, its origin, collectors, converting to electricity and heat.
14. Optics. Maxwell's equations. The interaction of matter with light.
15. Special relativity.
16. The second current written proof of knowledges.

**Literature:**

Krempaský J., (1992): Fyzika-Základný kurz pre technické univerzity. Bratislava, ALFA, ISBN 80-05-01063-X  
 Červeňová M.,(1998): Príklady na prijímacie skúšky. STU Bratislava, ISBN 80 227 1029 6  
 Krempaský J., (1992): Fyzika - Základný kurz pre technické univerzity. Bratislava, Alfa. ISBN 80-05-01063-X.  
 Paál T.,(2001): Fizika. Budapest, Nemzeti Tankönyvkiadó, ISBN 00 0954 3  
 Feynman R. P.,(1969) : Mai fizika 1 - A modern természettudomány alapjai - A mechanika törvénye. Budapest, Műszaki könyvkiadó, ISBN 00 0827 9  
 Feynman R. P., (1970): Mai fizika 4 - Statisztikus mechanika. Termodinamika. Hullámtan. Szimmetriák a fizika törvényeiben. Budapest, Műszaki Könyvkiadó, ISBN 00 0815 4  
 Székely L., (2010): Albert Einstein válogatott írásai - 3. vyd. - Budapest : Typotex Kiadó, - 444 s. - ISBN 978 963 279 158 6

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 32

A	B	C	D	E	FX
34.38	12.5	6.25	18.75	28.13	0.0

**Teacher:** Mgr. Ladislav Jaruska, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ KSP/15	<b>Name:</b> Selected Chapters from School Chemical Experiments
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester 1 writing test is compulsory: the maximum points are 50. Moreover, another 50 points are available from the mid-term and final projects (maximum points 50 + 50 = 100). The minimum requirement for the successful accomplishment of the course is overall 50 points, i.e. 50% of 100 points. Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> After successfully completing this course, students will be able to perform and explain the demo experiments, moreover will be able to apply them in his/her teacher career in the future.	
<b>Brief syllabus:</b> 1. Introduction. 2. Demonstration experiments with flame 3. Preparation of hydrogen gas; its physical and chemical properties 4. Preparation of oxygen gas; its physical and chemical properties 5. Oxides of sulfur — preparation, and study on their properties by using demo experiments 6. Oxides of carbon — preparation, and study on their properties by using demo experiments 7. Demonstration of colorful acid–base reactions 8. Demonstration of factors having influence on the rate of the chemical reactions 9. Teacher’s demo experiments for the qualitative analysis of selected inorganic compounds 10. Teacher’s demo experiments for the qualitative analysis of selected organic compounds 11. Student’s independent demo experiments of their choice 12. Final writing test	
<b>Literature:</b> Balázs, L., (1986): Kémiai kísérletek. Budapest: Móra Ferenc Könyvkiadó, 158s. - ISBN 963 11 5085 2. Kuracina, R. et al., (2009): Chemické pokusy hravo a zaujímavost. Trnava: AlumniPress, 89s. ISBN 978-80-8096-097-1. Dostupné online: <a href="http://www.prirodnejavy.eu/sub/brozura2.pdf">http://www.prirodnejavy.eu/sub/brozura2.pdf</a> Perczel, S., (1984): Kémiai kísérlet-gyűjtemény. Budapest: Tankönyvkiadó, 173s. - ISBN 9631778223.	

<p>Podhorányi, Gy.(1984): Kémiai kísérletgyűjtemény. Budapest: Nemzeti Tankönyvkiadó, 85s.- ISBN 9631873412.</p> <p>Straka,M.,(1997): Kouzelnické pokusy z chemie. Informační a metodické centrum. 34s. dostupné online: <a href="http://vestenie.wbl.sk/Pokusy.pdf">http://vestenie.wbl.sk/Pokusy.pdf</a></p>					
<b>Language, knowledge of which is necessary to complete a course:</b>					
<b>Notes:</b>					
<b>Evaluation of subjects</b>					
Total number of evaluated students: 14					
A	B	C	D	E	FX
57.14	21.43	7.14	7.14	0.0	7.14
<b>Teacher:</b> Mgr. Andrea Vargová, PhD.					
<b>Date of last update:</b> 12.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ KSV/15	<b>Name:</b> Selected Chapters from Chemistry Calculuses
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester, the students will be delivered a test of maximum 50 points, while he/she can gather another 50 points with homeworks assigned during the semester. For the successful termination of the course, one has to gather at least 50 points (i.e. 50% of the maximum count of points). For the final classification to be A one has to obtain 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> Attending the course the students get acquainted with basic chemistry calculuses needed for primary and secondary education. Students are able to implement their knowledge and process them into the education process. They are able to make worksheet, form problem and tasks in chemistry education. Students are capable to process chemistry exercise and problem, analyze its didactical aspects and make assessment tools to them.	
<b>Brief syllabus:</b> 1. Introduction. Physical-chemical quantities, base quantities (ISQ), units. 2. Amount of substance, size of an ensemble of elementary entities, relative atomic and molecular mass, volume, relationships between physical quantities. 3. Solutions, mass-, volume- and mole fraction. 4. Molar concentration, calculuses to make solutions. 5. Chemistry calculuses by reaction rates. 6. Balancing redox a non-redox reactions. 7. Thermochemical calculuses. 8. Creation of writing tests of chemistry calculuses to assess students knowlegde in primary- and secondary education. Tvorba písomných previerok chemických výpočtov. 9. Creation of worksheet to exercise chemistry calculuses in primary- and secondary education. 10. Creation the online exercises and tests for students in primary- and secondary education. 11. Writing test. 12. Sumary course evaluation.	
<b>Literature:</b> Krätsmár-Šmogrovič, J. a kol.(2007): Všeobecná a anorganická chémia. Osveta, ISBN 80 806 3245 8	

Fajnor V., (1998): Všeobecná a anorganická chémia. Vysokoškolské skriptá - 1. vyd. – UK Bratislava, 266 s. - ISBN 80-223-1257-6  
 Kiss Zs., (2004): Összefoglaló feladatgyűjtemény kémiából – Megoldások. Budapest, Nemzeti Tankönyvkiadó,. ISBN 963 19 5394 7  
 Kotočová A., Valigura D.,(1993): Všeobecná chémia- Návody na laboratorne cvičenia. Bratislava: Slovenská technická univerzita, ISBN 80 227 0560 8  
 Sík J., (1992): Kémiai számítások képletgyűjteménye. Budapest: Műszaki Könyvkiadó, ISBN 963 10 9419 7  
 Cieľové požiadavky na vedomosti a zručnosti maturantov z chémie – podľa aktuálneho vydania ŠPÚ on-line dostupné na [www.statpedu.sk](http://www.statpedu.sk)

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 16

A	B	C	D	E	FX
56.25	12.5	6.25	12.5	12.5	0.0

**Teacher:** Mgr. Katarína Szarka, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ MCL/15	<b>Name:</b> Management of School Chemistry Laboratories
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester, the students will be delivered two written tests each of maximum 25 points. To be allowed for the oral part of the examination, the students will have to gather at least 25 points from both tests (i.e. 50% of the total possible count). The maximum number of points obtainable at the oral part of the exam is 50. The final classification is obtained from the sum of both parts of the examination – written and oral. For the final classification to be A one has to obtain 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> Students get special basic technical knowledge to build up and furnish school chemistry laboratory. They get acquainted with equipments, devices, materials and chemicals needed to set laboratory going. The students know laboratory safety rules and guidelines and implements them into the chemistry laboratory practise in their pedagogical process.	
<b>Brief syllabus:</b> 1. Laboratory Safety Guidance, giving an assistance (first aid) in case of laboratory accident, rules of the fire protection during the laboratory work. 2. Laboratory equipments, devices, materials and chemicals. 3. Pressure vessel and maintenance and use. Chemicals and materials – their ordering, storage. 4. Laboratory glass and electrical equipments. 5. Storage of solid and liquid chemicals. Skladovanie tuhých a tekutých chemikálií. List of chemical stock. Chemical storage. Hazardous chemical substancies. The waste storage and liqidation 6. The 1st writing test. 7. Preparation, labeling, storage and manipulation with solution. 8. Operational regulations of the laboratory. 9. Legal aspects of the laboratory working. 10. Building up strategy of the school chemistry laboratory. 11. Internal auditing in laboratory and technical controll the laboratory operation. 12. The 2nd working test.	
<b>Literature:</b>	

Fajnor V., (1992): Laboratórna technika, názvoslovie a chemické výpočty. UK Bratislava, ISBN 80 223 0436 0  
 Sokolík J., a kol., (2012): Názvoslovie a príprava vybraných anorganických látok. UK Bratislava, ISBN 978 80 223 2913 2  
 Kotočová A., Valigura D., (1993): Všeobecná chémia- Návody na laboratórne cvičenia. STU Bratislava, ISBN 80 227 0560 8  
 Karlíček R., a kol., (2009) : Analytická chemie pro farmaceuty, Karolinum, - 279 s., ISBN 978 80 246 1453 3  
 Čermáková E., Feltl L., Němcová I. (1980) : Analytická chemie 2. - 1. vyd. – Praha, SNTL, Nakladatelství technické literatury,- 272 s.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 10

A	B	C	D	E	FX
70.0	10.0	10.0	10.0	0.0	0.0

**Teacher:** Mgr. Katarína Szarka, PhD., Mgr. Andrea Vargová, PhD.

**Date of last update:** 12.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ MOB/15	<b>Name:</b> Molecular Biology
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester a writing test is compulsory: the maximum points are 50. Further 50 points can be collected from project work. The minimum requirement for the successful accomplishment of the course is overall 50 points, i.e. 50% of 100 points. Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> By absolving this course, the students will obtain basic knowledge on the mechanism of DNA replication, transcription and translation. He/she will become familiar with the molecular basics of genetics, with the transfer of genetic information and its performance during the personal development.	
<b>Brief syllabus:</b> 1. History and progress of molecular biology 2. Nucleic acids. Structure of DNA. The double helix. DNA sequence 3. Physical and chemical properties of the DNA 4. Methods of DNA examination 5. Structure of RNA. Characterization of the different forms of RNA. Comparison of the DNA and RNA 6. Writing test 7. DNA replication 8. Translation 9. Transcription. The genetic code 10. Regulation of gene expression 11. DNA recombination. Practical use of the genetic recombination 12. DNA cloning. Methods and use of DNA sequencing 13. The size of the genome, and its organization 14. DNA polymorphism 15. Writing test	
<b>Literature:</b>	

Gálová Z., et al. (2007) : Molekulárna biológia. - 2. vyd. - Nitra : SPU - 165 s. - ISBN 978-80-8069-951-2  
 Golais F., (1986) : Molekulárna biológia a genetika vírusov. - Bratislava : UK v Bratislave, - 124. - ISBN 00 1062 7  
 Hrubý K., (1961) : Genetika. - 1. vyd. - Praha : Československé Akadémie Vied, - 647 s.  
 Vodrážka Z.(2007) : Biochemie. - 1. vyd. - Praha : Academia, - 190 s. - ISBN 978-80-200-0600-4.  
 Brechtlová M., Halčák L., (2007) : Lekárska biochémia - Seminárna a praktická časť. - 3. vyd. - Bratislava : Univerzita Komenského v Bratislave,- 168 s. - ISBN 978-80-223-2304-8  
 Mandl J.,et al. (2006) : Biokémia. - 1. vyd. - Budapest : Semmelweis Kiadó, - 176 s. - ISBN 963 9656 18 6.  
 Watson J.D., (1988) : Rekombinantní DNA. - 1. vyd. - Praha : Academia, - 294 s.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 13

A	B	C	D	E	FX
30.77	15.38	15.38	30.77	7.69	0.0

**Teacher:** Mgr. Andrea Vargová, PhD.

**Date of last update:** 12.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ MPC/15	<b>Name:</b> Mathematics for Chemists
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 / 0 <b>For the study period:</b> 26 / 0 <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester there will be two written assessments for 25-25 points, in order to access the oral test, students have to achieve at least 25 points from the two assessments together, i.e. 50% of the total. For the oral examination the student can get 50 points. The final evaluation result depends on the oral exam and on the written assessments (50%-50%). To achieve evaluation A 90-100% is needed, for evaluation B 80-89% is needed, for evaluation C 70-79% is needed, for evaluation D 60-69% is needed, and for evaluation E 50-59% is needed from the total number of points	
<b>Results of education:</b> By completing the course, students gain knowledge of linear algebra, mathematical analysis and statistics. Aside from this, they also acquire the skills to work with the mathematical apparatus.	
<b>Brief syllabus:</b> <ol style="list-style-type: none"> <li>1. Expressions, Transformation of Expressions, polynomes, the complex numbers.</li> <li>2. Vectors, Vector Spaces And Fields , Matrices, Determinants, Linear systems of equations.</li> <li>3. Algebraic equations. Groups of molecular symmetries,</li> <li>4. Real function of one variables – definition and properties, graphs, elementary functions.</li> <li>5. Limit of a function, continuity for real function.</li> <li>6. Differentiable Functions of One Variable – Definition of the Derivative, L’Hospital’s Rule, Use of Differential Calculus in Chemistry.</li> <li>7. Integral Calculus of Functions of One Variables - Definition of the Integral, methods of Integral Calculus, Rieman Integral, Newtonov – Leibniz formule, application of the integral Use of Integral in Chemistry.</li> <li>8. Written assessment.</li> <li>9. First order Differential Equations – with separable variables, homogenous, linear, equations with constant coefficients, Use of Differential equations in Chemistry.</li> <li>10. Basic Differential and Integral Calculus of real functions with multiple variables – definitions, properties of functions, partial derivatives, gradient, multiple Integral.</li> <li>11. Infinite Sequences and Series, Taylor’s Theorem,</li> <li>12. Statistical analysis of measurements.</li> <li>13. Graphical analysis of measurements.</li> </ol>	

14. Written assessment.

**Literature:**

Neubrunn T., (1992): Matematická analýza I . - 1. vyd. – Bratislava, Univerzita Komenského, 190 s. - ISBN 80-223-0055-1

Neubrunn T., (1992) : Matematická analýza II. - 1. vyd. - Bratislava, Univerzita Komenského, 166 s. - ISBN 80-223-0051-9

Krajňáková D., Míčka J., Macháčová L., (1988): Zbierka úloh z matematiky. Bratislava, Alfa, 538 s. - ISBN 0002566

Chajdiak J., (2002): Štatistika v Exceli . 1. vyd. – Bratislava, Statis,. 159 s. - ISBN 80-85659-27-1

Petres T., ( 2003): Statisztika. Szeged , JATEPress, 272 s. - ISBN 0242073

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 46

A	B	C	D	E	FX
8.7	17.39	13.04	28.26	26.09	6.52

**Teacher:** Dr. habil. PaedDr. György Juhász, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ OC1/15	<b>Name:</b> Organic Chemistry I.
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester 2 writing tests are compulsory: the maximum points are $2 \times 25 = 50$ . The minimum eligibility requirement for the oral exam is overall 25 points from the two writing tests. The maximum points at the oral exam are 50. The final evaluation comprises both the writing test and oral exam (maximum points $50 + 50 = 100$ ). Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> By successful absolution of this course, students will be familiar with the basic organic chemistry. They will get knowledge on the nomenclature of organic compounds, properties of the common organic compounds, the nature of the chemical reactions and some basic stereochemical terms and representations.	
<b>Brief syllabus:</b> 1. História organickej chémie a nomenklatura organických zlúčenín. 2. Stereogénny uhlík, absolútna konfigurácia, optické izoméry, nomenklatura chirálnych molekúl, racemická zmes. Stereochemia. Indukčný a mezoméryný efekt, konjugované $\pi$ – systémy. 3. Alkány, cykloalkány, bicykloalkány. Nomenklatura, štruktúra, fyzikálne a chemické vlastnosti. 4. Alkény, cykloalkény. Nomenklatura, štruktúra, fyzikálne a chemické vlastnosti. 5. Diény. Nomenklatura, štruktúra, fyzikálne a chemické vlastnosti. 6. Alkíny. Nomenklatura, štruktúra, fyzikálne a chemické vlastnosti. 7. Aromatické uhl'ovodíky. Nomenklatura aromatických uhl'ovodíkov. Aromatickosť. Nomenklatura, štruktúra, fyzikálne a chemické vlastnosti. 8. Reakcie aromatických uhl'ovodíkov. 9. Halogénuhl'ovodíky. Nomenklatura halogénuhl'ovodíkov. Väzba C – halogén — polarita väzby, dipólový moment, polarizovateľnosť molekúl. Fyzikálne a chemické vlastnosti. Reakcie halogénuhl'ovodíkov. Grignardove činidlá. 10. Aromatické halogénderiváty.	
<b>Literature:</b> Odporúčaná literatúra:	

Devínsky F., a kol.(2001) : Organická chémia pre farmaceutov. 1. vyd. – Bratislava, Osveta, - 750 s. ISBN 80-8063-056-9

Kováč J., Kováč Š.,(1977) : Organická chémia. 1 vyd. – Bratislava, Vydavateľstvo technickej a ekonomickej literatúry, 928 s.

Antus S., Mátyus P., (2010) : Szerves kémia I. Budapest, Nemzeti Tankönyvkiadó, ISBN: 978 963 195 716 7

Balogh Á., (1990): Szerves kémia. Budapest, Tankönyvkiadó, ISBN 96 318 2741 0

Halmos I., (1992): Szerves kémia. Budapest, Műszaki Könyvkiadó, ISBN 96 310 9743 9

Kajtár M., (2009): Változatok négy elemre - Szerves kémia 1-2. ELTE Eötvös Kiadó Kft., ISBN: ISBN 978 963 284 114 4.

McMurry J., (2007) : Organická chemie, ISBN 987-80-7080-637-1

Červinka O., (1980) : Organická chemie - 2. vyd. – Praha, SNTL, ALFA - 791 s.

Panchartek J., Štěřba V., Večeřa M., (1977) : Organická chemie II- Reakční mechanismy - 1. vyd. - Pardubice - 316 s.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 39

A	B	C	D	E	FX
41.03	35.9	12.82	5.13	0.0	5.13

**Teacher:** Mgr. Alexandra Hengerics Szabó, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ OC2/15	<b>Name:</b> Organic Chemistry II.
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester 2 writing tests are compulsory: the maximum points are $2 \times 25 = 50$ . The minimum eligibility requirement for the oral exam is overall 25 points from the two writing tests. The maximum points at the oral exam are 50. The final evaluation comprises both the writing test and oral exam (maximum points $50 + 50 = 100$ ). Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> By successfully finishing this course, students will learn the basic principles of organic chemistry. They will study the nomenclature of organic chemistry, the physical and chemical properties of the most important organic compounds, and the process of the basic organic reactions. In the future, they will be able to apply this basic knowledge for solving real practical problems.	
<b>Brief syllabus:</b> 1. Compounds with hydroxyl group. Alcohols and phenols. Reactivity of the hydroxyl group. Detection and identification of the hydroxyl derivatives 2. Ethers, thiols and sulfides 3. Compounds with carbonyl group. Aldehydes and ketones. 4. Carboxylic acids. Nomenclature, constitution. Physical and chemical properties 5. Functional derivatives of carboxylic acids. Acyl halides, anhydrides, esters, amides. 6. Written test 7. Carboxylic acid derivatives — acyl halides, amides 8. Nitrocompounds 9. Amines. Basicity of the amines. Reactions of the amines. Preparation and reactions of the diazonium salts 10. Heterocyclic compounds. Nomenclature, physical and chemical properties. 11. Polymers and plastics 12. Final writing test	
<b>Literature:</b> Bláha K., et al. (1985): Chemie organických sloučenin. Díl první. - 1. vyd. - Praha : SNTL Nakladatelství technické literatury, - 1131 s.	

Bláha K., et al. (1987) : Chemie organických sloučenin. Díl druhý - 1. vyd. - Praha : SNTL Nakladatelství technické literatury, - 1056 s.

Devínsky F., et al. (2001) : Organická chémia pre farmaceutov. 1. vyd. – Bratislava, Osveta, - 750 s. ISBN 80-8063-056-9

Kováč J., Kováč Š.,(1977) : Organická chémia. 1 vyd. – Bratislava, Vydavateľstvo technickej a ekonomickej literatury, 928 s.

Antus S., Mátyus P., (2010) : Szerves kémia I. Budapest, Nemzeti Tankönyvkiadó, ISBN: 978 963 195 716 7

Balogh Á., (1990): Szerves kémia. Budapest, Tankönyvkiadó, ISBN 96 318 2741 0

Halmos I., (1992): Szerves kémia. Budapest, Műszaki Könyvkiadó, ISBN 96 310 9743 9

Kajtár M.: Változatok négy elemre - Szerves kémia 1-2. ELTE Eötvös Kiadó Kft., ISBN: 9789 6328 4113 7

McMurry J., (2007) : Organická chemie, ISBN 987-80-7080-637-1

Červinka O., (1980) : Organická chemie - 2. vyd. – Praha, SNTL, ALFA - 791 s.

Panchartek J., et al. (1977) : Organická chemie II- Reakční mechanismy - 1. vyd. - Pardubice - 316 s.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 36

A	B	C	D	E	FX
30.56	44.44	19.44	2.78	2.78	0.0

**Teacher:** Mgr. Alexandra Hengerics Szabó, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ PC1/15	<b>Name:</b> Laboratory Course of Inorganic Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester the students will be issued two written tests each of maximum 30 points, while another amount of 40 points can be granted for his/her laboratory protocols. The final classification is obtained as the sum of points obtained for the written tests (60%) and from the classification of laboratory protocols (40%). For the final classification to be A one has to acquire 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> During this practical course students will conduct the syntheses of selected inorganic compounds. Methods and chemicals employed are selected to cover the main types of inorganic compounds and are selected to provide necessary theoretical and practical skills not only within the organized pedagogical process, but also in the form of individual studies.	
<b>Brief syllabus:</b> 1. Safety regulations and health protection in chemical laboratories. Laboratory guide. 2. Preparation of elements – powder copper. 3. Preparation of oxides – iron(III) oxide. 4. Preparation of acids – boric acid.. 5. Preparation of hydroxides – nickel(II) hydroxide. 6. Preparation of salts – sodium chloride. 7. Written test. 8. Preparation of salts – barium nitrate. 9. Preparation of salts – potassium-aluminium sulphate dodecahydrate. 10. Preparation of salts – cobalt(II) chloride hexahydrate. 11. Preparation of complex compounds – copper(tetraammin)sulphate monohydrate. 12. Preparation of complex compounds – cobalt(hexaammin)chloride. 13. Written test. 14. Substitute lesson for missed classes/tasks.	
<b>Literature:</b> Fajnor V., (1992): Laboratórna technika, názvoslovie a chemické výpočty. Vysokoškolské skriptá, UK Bratislava, ISBN 80 223 0436 0	

Sokolík J., a kol., (2012): Názvoslovie a príprava vybraných anorganických látok. UK Bratislava, ISBN 978 80 223 2913 2  
 Kotočová A., Valigura D., (1993): Všeobecná chémia- Návody na laboratórne cvičenia. Bratislava, Slovenská technická univerzita, ISBN 80 227 0560 8  
 Sokolík J., a kol., (1991): Laboratórne cvičenia a výpočty zo všeobecnej a anorganickej chémie. UK Bratislava, ISBN 80 223 0366 6  
 Sík J., (1992): Kémiai számítások képletgyűjteménye. Budapest, Műszaki Könyvkiadó, ISBN 00 0950 1  
 Kiss Zs., (2004): Összefoglaló feladatgyűjtemény – Kémiából – Megoldások. Budapest, Nemzeti Tankönyvkiadó, ISBN 963 19 5394 7

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 40

A	B	C	D	E	FX
50.0	12.5	22.5	12.5	2.5	0.0

**Teacher:** Mgr. Katarína Szarka, PhD., Mgr. Andrea Vargová, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ PC2/15	<b>Name:</b> Laboratory Course of Analytical Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester the students will be issued two written tests each of maximum 30 points, while another amount of 40 points can be granted for his/her laboratory protocols. The final classification is obtained as the sum of points obtained for the written tests (60%) and from the classification of laboratory protocols (40%). For the final classification to be A one has to acquire 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> Completing the Course the students acquire some specialized knowledge in Inorganic Chemistry involving skills for the proof of cations and anions together with utilizing these skills for the exploration of an unknown mixture. The aim of volumetric analysis is to acquire practical skills in preparing standard solutions, in conducting titrations with emphasis on analytical accuracy, and to master the required calculations for determining the analyte concentration in the sample examined.	
<b>Brief syllabus:</b> 1. The classical division of cations and anions. Chemical tests of group I, II and III cations. 2. Chemical tests of group IV and V. Separation of group I. and II. cations. 3. Chemical tests of group III and IV. Anion tests. 4. Application of the classical division of cation for separating cations in an unknown sample. 5. Introduction to volumetric analysis. Solution standardization in volumetric analysis. 6. Alkalimetry of weak acids. Quantitative determination of acetic acid in vinegar. 7. Acidimetry. Alkalinity determination of sodium hydroxide. 8. Complexometry. Water hardness determination by chelatometry. 9. Indirect chelatometric determinations. Indirect determination of sulphates. 10. Reverse chelatometric determinations. Determination of aluminium. 11. Precipitation titrations. Argentometry. Determination of chlorides by Mohr. 12. Redox titrations. Manganometry. Determination of iron in samples. 13. Redox titrations. Bromatometry. Determination of arsenic. 14. Substitute lesson for missed classes/tasks.	
<b>Literature:</b> Majer J., et al. (1988): Analytická chémie. Martin, Osveta, – 368 s.	

<p>Karlíček, R. a kol., (2009) : Analytická chemie pro farmaceuty, Karolinum, - 279 s., ISBN 978 80 246 1453 3</p> <p>Čermáková E., Feltl L., Němcová I., (1980) : Analytická chemie 2, Instrumentální analýza- pro SPŠ skupiny studijních odborů technická chemie. - 1. vyd. – Praha, SNTL, Nakladatelství technické literatury, -272 s.</p> <p>Churáček J., Kotrlý. S., (1983) : Analytická chemie II. - 1. vyd. - Pardubice, -190 s.</p> <p>Okáč A., (1961) : Analytická chemie kvalitativní .- 1. vyd. - Praha : Nakladatelství akademie věd, - 550s.</p> <p>Barcza, L. (2006): A mennyiségi kémiai analízis gyakorlati kézikönyve. Medicina Kiadó ISBN: 96 324 2961 3</p> <p>Barcza, L. (2007) : Kvantitatív analitikai kémia. Budapest: Semmelweis Kiadó,</p> <p>Barcza, L., Buvári, Á. (2009) : A minőségi kémiai analízis gyakorlati kézikönyve. Medicina Könyvkiadó, ISBN: 978 963 226 246 8.</p> <p>Barcza, L., Buvári, Á. (2008) : A minőségi kémiai analízis alapjai. Medicina, ISBN:978 963 226 186 7.</p> <p>Keller R. (Ed. ) (1998): Analytical Chemistry. Wiley-VCH, Weinheim</p>					
<b>Language, knowledge of which is necessary to complete a course:</b>					
<b>Notes:</b>					
<b>Evaluation of subjects</b>					
Total number of evaluated students: 36					
A	B	C	D	E	FX
33.33	38.89	13.89	8.33	2.78	2.78
<b>Teacher:</b> Mgr. Alexandra Hengerics Szabó, PhD.					
<b>Date of last update:</b> 26.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ PC3/15	<b>Name:</b> Laboratory Course of Organic Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester 2 writing tests are compulsory: the maximum points are $2 \times 30 = 60$ . Further 40 points can be collected for the protocols prepared during the lab work. The minimum requirement for the successful accomplishment of the course is overall 60 points, i.e. 60%. Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> Students will synthesize various selected organic compounds. By selecting the starting materials and the synthetic methods, students will learn the basic principles and gain practical expertise in basic synthetic organic chemistry.	
<b>Brief syllabus:</b> 1. Lab works — the main emphasis is on the laboratory preparation of various organic compounds 2. Saturated, linear and cyclic hydrocarbons 3. Aromatic hydrocarbons 4. Halogen derivatives 5. Hydroxy derivatives 6. Ethers and nitro compounds 7. Writing test 8. Aldehydes, ketones, and organosulfur compounds 9. Carboxylic acids and derivatives 10. Substituted carboxylic acid derivatives 11. Natural products 12. Quantitative determination of food additives 13. Final writing test 14. Compensation day for missed classes	
<b>Literature:</b> Čižmáriková, R., (2012): Laboratórne cvičenia z organickej chémie . - 1. vyd. - Bratislava : Univerzita Komenského, 2012. - 115 s. - ISBN 978-80-223-3143-2. Hrnčiar P., et al. (1988) : Organická chémia v príkladoch. - 1. vyd. - Bratislava : Prírodovedecká fakulta Univerzity Komenského, - 224 s.	

Orosz Gy.,(1998): Szerves kémiai praktikum. Nemzeti Tankönyvkiadó, ISBN: 96 318 8408 2  
Večeřa M., Gasparič J., (1973) : Důkaz a identifikace organických látek. - 2.přepracované vyd. -  
Praha : SNTL, Nakladatelství technické literatury, - 422 s.  
Eckchlager K., (1971) : Chyby chemických rozborů : Moderní metody v chemické laboratoři ,  
svazek 6. - 2.přepracované vyd. - Praha : SNTL, Nakladatelství technické literatury, - 191 s.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 34

A	B	C	D	E	FX
70.59	17.65	5.88	2.94	0.0	2.94

**Teacher:** Mgr. Andrea Vargová, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky,  
DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ PC4/15	<b>Name:</b> Laboratory Course of Physical Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester there will be two written assessments for 30 points. Students can obtain additional 40 points for continuously transmitted laboratory protocols. The final evaluation arises from the average points of the two assessments (60%) and the laboratory protocols (40%). To achieve evaluation A 90-100% is needed, for evaluation B 80-89% is needed, for evaluation C 70-79% is needed, for evaluation D 60-69% is needed, and for evaluation E 50-59% is needed from the total number of points.	
<b>Results of education:</b> Practical course from physical chemistry is an integral part of the teaching process of theoretical physical chemistry. It applies the basic principles and laws of physical chemistry in laboratory practice. Students acquire necessary laboratory skills and the ability to process the results of experiments. Laboratory practice covers all areas of physical chemistry: chemical thermodynamics, structure and properties of matter, electrochemistry and chemical kinetics.	
<b>Brief syllabus:</b> <ol style="list-style-type: none"><li>1. Safety and safeguard of health in chemical laboratory</li><li>2. Conduct Electricity in electrolytic solutions.</li><li>3. Electrolyte Solutions of Inorganic salts</li><li>4. Factors that Affect Reaction Rate.</li><li>5. Chemical Equilibrium,</li><li>6. Written assessment.</li><li>7. Conductivity - Conductometric Titrations</li><li>8. Spectrophotometry – determination of concentration of capsanthine.</li><li>9. HPLC - High Performance Liquid Chromatography – determination of concentration of vitamin C by HPLC.</li><li>10. Determination of dissociation equilibrium constant of weak acid.</li><li>11. Written assessment.</li><li>12. Replacement term of missed laboratory practices</li></ol>	
<b>Literature:</b> Kotek J.,(2007) : Laboratorní technika. Univerzita Karlova v Praze, Nakladatelství Karolinum, ISBN 978 80 246 1441 0	

Adamčík V., et al. (1989) : Fyzikálna chémia - Laboratórne cvičenia z fyzikálnej chémie. - 1. vyd. - Bratislava : alfa Vydavateľstvo technickej a ekonomickej literatúry, - 200 s. - ISBN 80-05-00424-9

Grančičová O., Vollárová O., (1984) : Cvičenia z fyzikálnej chémie : Vysokoškolské skriptá.- 2. vyd. - Bratislava : UK.

Ulický L., Vavra J., (1992) : Návody do cvičenia z fyzikálnej chémie. - 1. vyd. - Bratislava : Slovenská Vysoká Škola Technická v Bratislave.

Ševčík P., Adamčíková E., (1982) : Pokročilé cvičenie z fyzikálnej chémie.- 1. vyd. - Bratislava : UK.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 31

A	B	C	D	E	FX
16.13	22.58	48.39	6.45	6.45	0.0

**Teacher:** prof. Róbert Mészáros, DSc., Attila Kardos, PhD., Dr. habil. PaedDr. György Juhász, PhD., Dr. habil. Imre Varga, PhD.

**Date of last update:** 12.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ PC5/15	<b>Name:</b> Laboratory Course of Biochemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester 2 writing tests are compulsory: the maximum points are $2 \times 30 = 60$ . Further 40 points can be collected for the protocols prepared during the lab work. The minimum requirement for the successful absolution of the course is overall 60 points, i.e. 60%. The maximum points at the oral exam are 50. Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> Students will learn the fundamental biochemical methods and the experimental determination of several, biologically important materials. Students will obtain the ability and experiences in the lab work, they will be able to interpret the scientific results individually, and will have the practical ability to propose and manage independent research projects	
<b>Brief syllabus:</b> 1. Stoichiometric determination of dry material content and wet 2. Hydrolysis of sugars, carbohydrates, and szacharides 3. Amino acids — separation of amino acid mixtures by thin-layer chromatography 4. Proteins — precipitation of casein from milk samples 5. Separation and detection of non-natural dyes 6. Written test 7. Separation and detection of natural dyes 8. Qualitative determination of ascorbic acid 9. Semiquantitative determination of quality properties in urine samples by HPLC 10. Quantitative determination of creatinine in urine samples by HPLC 11. Enzyme activity studies — Studying the activity profile of saccharase (invertase) in view of some external effects 12. The effect of the concentration of some heavy metals on the growth of microorganisms 13. Final writing test 14. Compensation day for missed classes	
<b>Literature:</b> Grones J., et al. (1986): Cvičenie metód z biochémie : Vysokoškolské skriptá. - 1. vyd. – Bratislava, Univerzita Komenského, - 64 s.	

Karlubík M., (1990): Biochémia. Nitra: VŠP  
 Karlubík M., (1987) : Návody na cvičenia z biochémie. Nitra: VŠP  
 Michalík I., (1989) : Návody na cvičenia z biochémie rastlín. Nitra: VŠP  
 Hrnčiar P., (1988) : Organická chémia v príkladoch. - 1. vyd. - Bratislava : Prírodovedecká fakulta UK, - 224 s  
 Görbe A. et al. (2011): Biokémiai gyakorlatok . - 1. vyd. - Budapest : Medicina Könyvkiadó Zrt., - 95 s. - ISBN 978 963 226 320 5.

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 28

A	B	C	D	E	FX
21.43	60.71	14.29	3.57	0.0	0.0

**Teacher:** Mgr. Alexandra Hengerics Szabó, PhD., Mgr. Andrea Vargová, PhD.

**Date of last update:** 12.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KCH/CHdb/SSB/15		<b>Name:</b> State Exam			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> <b>Recommended extent of course ( in hours ):</b> <b>Per week: For the study period:</b> <b>Methods of study:</b> present					
<b>Number of credits:</b> 2					
<b>Recommended semester/trimester of study:</b> 5., 6..					
<b>Level of study:</b> I.					
<b>Prerequisites:</b> KCH/CHdb/MPC/15 and KCH/CHdb/ANC/19 and KCH/CHdb/VSC/15 and KCH/CHdb/ZLT/15 and KCH/CHdb/ARC/15 and KCH/CHdb/FPC/19 and KCH/CHdb/PC1/15 and KCH/CHdb/OC1/15 and KCH/CHdb/PC2/15 and KCH/CHdb/FC1/15 and KCH/CHdb/OC2/15 and KCH/CHdb/PC3/15 and KCH/CHdb/BC1/15 and KCH/CHdb/FC2/15 and KCH/CHdb/PC4/15 and KCH/CHdb/BC2/15 and KCH/CHdb/PC5/15					
<b>Conditions for passing the subject:</b> Passed exam and succesfull accomplishe of the obligatory subjects. Oral answer of student evaluated by the Commission for state exams. Final evaluation: A - 100- 90% B - 89 - 80%, C - 79-70%, D - 69-60%, E - 59 - 50%. Credits are not awarded to student, who do not achieve 50%.					
<b>Results of education:</b> Through the subjects of the specialization, the graduate of the study programme Teacher Training in Chemistry (combined) masters the basic content of the disciplines of the specialization.					
<b>Brief syllabus:</b> Through the subjects of the specialization, the graduate of the study programme Teacher Training in Chemistry (combined) masters the basic content of the disciplines of the specialization.					
<b>Literature:</b> The suggested literatures available within information paper of the obligatory subjects.					
<b>Language, knowledge of which is necessary to complete a course:</b>					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 28					
A	B	C	D	E	FX
32.14	10.71	17.86	21.43	17.86	0.0
<b>Teacher:</b>					
<b>Date of last update:</b> 12.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ VAC/15	<b>Name:</b> Selected Chapters from Inorganic Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester, the students will be delivered a test of maximum 50 points, while he/she can gather another 50 points with homeworks assigned during the semester. For the successful termination of the course, one has to gather at least 50 points (i.e. 50% of the maximum count of points). For the final classification to be A one has to obtain 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> Attending the course the student get some more detailed theoretical knowledge about the inorganic chemistry of elements and their compounds.	
<b>Brief syllabus:</b> 1. The periodic system of elements and the electron structure of their valence shells. 2. Compounds in general, lattice and bond types, characteristics and categories of compounds – hydrides, halogenides, oxides, peroxides, superoxides, oxoacids, sulphides, nitrides, fosfides, karbides, silicides, borides, cyanides, cyanates. 3. Hydrogen, bond types, occurrence, preparation, its compounds and isotopes. 4. Alkali metals – elements of group I of the periodic system, bond types, compounds, the subgroup of copper. 5. Elements of group II of the periodic system, bond types, compounds, the subgroup of zinc. 6. Coordination compounds. 7. Elements of group III of the periodic system, bond types, compounds, the subgroup of scandium, hybridization types. 8. Elements of group IV of the periodic system, bond types, compounds, the subgroup of titanium. 9. Elements of group V of the periodic system, bond types, compounds, the subgroup of vanadium. 10. Elements of group VI of the periodic system, bond types, compounds, the subgroup of chromium. 11. Elements of group VII of the periodic system, bond types, compounds, the subgroup of manganese. 12. Elements of group VIII of the periodic system and their compounds. Prvky III. skupiny periodického systému, ich zlúčeniny, väzby, podskupina skandia, typy hybridizácie 13. Written test.	



**Literature:**

Odporúčaná literatúra:

Greenwood N. N., Earnshaw A., (1993): Chemie prvků I a II. ISBN 80-85427-38-9

Krätsmár - Šmogrovič J. a kol., (2007): Všeobecná a anorganická chémia. Osveta, ISBN 80 806 3245 8

Fajnor V., (1998) : Všeobecná a anorganická chémia. - 1. vyd. – Bratislava, Univerzita Komenského - 266 s. - ISBN 80-223-1257-6

Gažo J., Kohout J., Serátor M., (1981) : Všeobecná a anorganická chémia. Bratislava, ALFA - 804 s.

Lukeš I., (2009): Systematická anorganická chémie. - 1. vyd. – Praha, Nakladatelství Karolinum - 230 s. ISBN 978-80-246-1614-8

Zikmund M.,(1995): Anorganická chémia. Bratislava : Univerzita Komenského, ISBN 80-223-0919-2

Bánhidi L., (1989): Szervetlen kémia. Budapest, Tankönyvkiadó, ISBN 96 318 2192 7

Fehér D., (1987): Szervetlen kémia. Budapest, Tankönyvkiadó, ISBN 96 318 0282 5

**Language, knowledge of which is necessary to complete a course:****Notes:****Evaluation of subjects**

Total number of evaluated students: 39

A	B	C	D	E	FX
58.97	10.26	5.13	12.82	5.13	7.69

**Teacher:** doc. RNDr. Róbert Gyepes, PhD.**Date of last update:** 26.06.2023**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ VAN/16	<b>Name:</b> Calculations in Analytical Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester, the students will be delivered two written tests. For the successful termination of the course, one has to gather at least 50 points (i.e. 50% of the maximum count of points). For the final classification to be A one has to obtain 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> Attending the course the students get acquainted with theoretical basics of chemistry calculus needed for analytical chemistry purpose.	
<b>Brief syllabus:</b> 1. Introduction – solutions, their concentration and determination of the solutions' concentration. 2. Chemical equilibrium – the concept of chemical equilibrium, equilibrium constant, strong and weak electrolytes. 3. Calculations the pH of strong and weak acids and their salts, buffers. 4. Precipitations – solubility, the impact of own ions for solubility. 5. Gravimetry, gravimetric factor, calculations of quantities from gravimetric analyses. 6. Titrimetry, standard solutions in titrimetric analyses, standardization of solutions, determination of the concentration of the analyte/titrant. 7. Standardization of the solutions, determination of their concentration. 8. Acido-basic analytical methods to determine the quantum of chemical substances. 9. Oxidation and reduction methods (permanganometry) used to determine the quantum of chemical substances. 10. Oxidation and reduction methods (iodometry) used to determine the quantum of chemical substances. 11. Complexation and its methods to determine the quantum of chemical substances. 12. Introduction into the chemometry. The basic poems and calculus.	
<b>Literature:</b> Karlíček R., a kol. (2009): Analytická chemie pro farmaceuty. Karolinum, ISBN 97 8802 46 1453 3 Majer J., (1989) : Analytická chemia. - 1. vyd. - Martin : Osveta n.p., - 368 s. Holzbecher Z., Churáček J., (1987) : Analytická chemia. - 1. vyd. – Praha, SNTL - Nakladatelství technické literatury, - 663 s. Barcza L., (2006): A mennyiségí kémiái analízis gyakorlatai kézikönyve.	

<p>Medicina Kiadó, ISBN: 963 2429 61 3 Barcza L., (2007): Kvantitatív analitikai kémia. Budapest, Semmelweis Kiadó, ISBN 978 963 9656 73 4 Barcza L., Buvári Á., (2009): A minőségi kémiai analízis. Medicina Könyvkiadó, ISBN 978 9 6 322 6186 7</p>					
<p><b>Language, knowledge of which is necessary to complete a course:</b> -</p>					
<p><b>Notes:</b></p>					
<p><b>Evaluation of subjects</b> Total number of evaluated students: 15</p>					
A	B	C	D	E	FX
13.33	33.33	20.0	26.67	6.67	0.0
<p><b>Teacher:</b> doc. Ing. Ondrej Hegedűs, PhD.</p>					
<p><b>Date of last update:</b> 26.06.2023</p>					
<p><b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.</p>					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ VBC/18	<b>Name:</b> Selected Chapters from Biochemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b>	
<b>Results of education:</b>	
<b>Brief syllabus:</b>	
<b>Literature:</b>	
<b>Language, knowledge of which is necessary to complete a course:</b>	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 8	
a	n
100.0	0.0
<b>Teacher:</b> Mgr. Andrea Vargová, PhD.	
<b>Date of last update:</b> 12.05.2023	
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.	

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ VFC/15	<b>Name:</b> Selected Chapters from Physical Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester there will be one written assessment for 50 points, students can gain additional 50 points for submitted works. The requirement for passing the course is to achieve at least 50 points, i.e. 50% of the total. To achieve evaluation A 90-100% is needed, for evaluation B 80-89% is needed, for evaluation C 70-79% is needed, for evaluation D 60-69% is needed, and for evaluation E 50-59% is needed from the total number of points	
<b>Results of education:</b> After completing the course, student is able to connect theoretical knowledge with practice through solving problems and examples from selected areas of physical chemistry.	
<b>Brief syllabus:</b> 1. Introduction, Physical Units and Properties. 2. Equations of State and the Ideal Gas Law, State Functions and Path Functions Kinetic Theory of Gase. 3. Thermodynamics. 4. Thermochemistry. 5. Multi-Component and Multi-Phases Systems. 6. Chemical Equilibrium. 7. Electrolyte Solutions, Thermodynamics of Ion Formation and Solvation 8. Conduct Electricity in electrolytic solutions, Faraday's Law, Conductivity,. 9. Electrochemical Cells, Batteries, The, Electrodes and Electrode potential. 10. Chemical Kinetics, Rate Laws, Reaction rates. 11. Written assessment 12. End of Course	
<b>Literature:</b> Atkins, P.W.: Fizikai kémia I-III. a tankönyvi feladatok megoldására. Tankönyvkiadó, 1991. ISBN 9631843505 Atkins, P. W.: Fizikai kémia I. Egyensúly. Budapest: Nemzeti Tankönyvkiadó, 2002. ISBN: 9631933148 Atkins, P. W.: Fizikai kémia II. Szerkezet. Budapest: Nemzeti Tankönyvkiadó, 2002. ISBN: 963192145X	

Biskupič S., Kellö V., Staško A., Vavra J., (1991) : Fyzikálna chémia I. - 1. vyd. - Bratislava ALFA - 296 s. - ISBN 80-05-00931-3  
 Brdička R., (1977): Základy fyzikální chemie. Praha, ACADEMIA  
 Čipera J., (1990): Fyzikálna chémia. Bratislava: Osveta, ISBN 80 217 0134 x  
 Ulický L., Vavra J., (1992) : Návody do cvičenia z fyzikálnej chémie. - 1. vyd. – Bratislava, SVŠT v Bratislave - 216 s.  
 Ulický L., a kol., (1972) : Štruktúra tuhej fázy. - 1. vyd. – Bratislava, SVŠT v Bratislave- 130 s.  
 Ulický L., Fyzikálna chémia I., FPV UCM, 1999

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 30

A	B	C	D	E	FX
13.33	23.33	23.33	26.67	13.33	0.0

**Teacher:**

**Date of last update:** 12.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ VKM/15	<b>Name:</b> Selected Chapters from Mathematics
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester there will be one written assessment for 50 points, students can gain additional 50 points for regularly submitted works. The requirement for passing the course is to achieve at least 50 points, i.e. 50% of the total. To achieve evaluation A 90-100% is needed, for evaluation B 80-89% is needed, for evaluation C 70-79% is needed, for evaluation D 60-69% is needed, and for evaluation E 50-59% is needed from the total number of points.	
<b>Results of education:</b> By completing the course, students gain knowledge of linear algebra, mathematical analysis and statistics, and they simultaneously gain skills for working with the mathematical apparatus as well.	
<b>Brief syllabus:</b> 1. Expressions, Transformation of Expressions, polynoms, the complex numbers. 2. Vectors, Vector Spaces And Fields , Matrices, Determinants, Linear systems of equations. 3. Algebraic equations. Groups of molecular symmetries, 4. Real function of one variables – definition and properties, graphs, elementary functions. 5. Limit of a function, continuity for real function. 6. Differentiable Functions of One Variable – Definition of the Derivative, L’Hospital’s Rule, Use of Differential Calculus in Chemistry. 7. Integral Calculus of Functions of One Variables - Definition of the Integral, methods of Integral Calculus, Rieman Integral, Newtonov – Leibniz formule, application of the integral Use of Integral in Chemistry. 8. Written assessment. 9. First order Differential Equations – with separable variables, homogenous, linear, equations with constant coefficients, Use of Differential equations in Chemistry. 10. Basic Differential and Integral Calculus of real functions with multiple variables – definitions, properties of functions, partial derivatives, gradient, multiple Integral. 11. Infinite Sequences and Series, Taylor’s Theorem, 12. Statistical analysis of measurements. 13. Graphical analysis of measurements.	
<b>Literature:</b>	

**Odporúčaná literatúra:**

Neubrunn T., (1992): Matematická analýza I . - 1. vyd. – Bratislava, Univerzita Komenského, 190 s. - ISBN 80-223-0055-1.

Neubrunn T., (1992) : Matematická analýza II. - 1. vyd. - Bratislava, Univerzita Komenského, 166 s. - ISBN 80-223-0051-9.

Krajňáková D., Míčka J., Macháčová L., (1988): Zbierka úloh z matematiky. Bratislava, Alfa, 538 s. - ISBN 0002566.

Chajdiak J., (2002): Štatistika v Exceli . 1. vyd. – Bratislava, Statis., 159 s. - ISBN 80-85659-27-1.

Petres T., ( 2003): Statisztika. Szeged , JATEPress, 272 s. - ISBN 0242073

**Language, knowledge of which is necessary to complete a course:****Notes:****Evaluation of subjects**

Total number of evaluated students: 43

A	B	C	D	E	FX
11.63	20.93	9.3	23.26	27.91	6.98

**Teacher:** Dr. habil. PaedDr. György Juhász, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ VKO/15	<b>Name:</b> Selected Chapters from Organic Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester a writing test is compulsory: the maximum points are 50. Further 50 points can be collected from project work. The minimum requirement for the successful accomplishment of the course is overall 50 points, i.e. 50% of 100 points. Grading system: grade A (90–100%), grade B (80–89%), grade C (70–79%), grade D (60–69%), grade E (50–59%), and grade F (49% and below).	
<b>Results of education:</b> After the successful accomplishment of his/her studies, students will become familiar with the nomenclature of organic compounds, and will be able to solve problems in the field of organic chemistry. He/she will be able to characterize the fundamental groups of organic chemistry, successfully arrange the reaction equations of organic compounds and interpret the basic principles of stereochemistry.	
<b>Brief syllabus:</b> 1. Chemical bonds in organic compounds. Stereochemistry 2. Chemical calculations 3. Nomenclature of hydrocarbons 4. Nomenclature of hydrocarbon derivatives 5. Writing tests 6. Alkanes and cycloalkanes. Free radical substitution (SR) 7. Alkenes and alkynes. Electrophilic addition (AdE) 8. Arenes. Aromaticity 9. Reaction of aromatic compounds. Aromatic electrophilic substitution (aromatic SE) 10. Organohalogenic compounds. Reaction of alkyl halides. Nucleophilic substitution (SN) and elimination (E) 11. Final writing test	
<b>Literature:</b> Čižmáriková, R. et al. (2012): Laboratórne cvičenia z organickej chémie. Bratislava: Univerzita Komenského, 116 s., ISBN 978-80-223-3143-2. Hrnčiar P., ( 1988 ) : Organická chémia v príkladoch. Bratislava, Univerzita Komenského	

Devínsky F., a kol.(2001) : Organická chémia pre farmaceutov. 1. vyd. – Bratislava, Osveta, - 750 s. ISBN 80-8063-056-9  
 Kováč J., Kováč Š.,(1977) : Organická chémia. 1 vyd. – Bratislava, Vydavateľstvo technickej a ekonomickej literatúry, 928 s.  
 Bláha K., et al. (1985): Chemie organických sloučenin. Díl první - 1. vyd. - Praha : SNTL Nakladatelství technické literatúry, - 1131 s.  
 Antus S., Mátyus P., (2010) : Szerves kémia I. Budapest, Nemzeti Tankönyvkiadó, ISBN: 978 963 195 716 7  
 McMurry J., (2007) : Organická chemie, ISBN 987-80-7080-637-1  
 Červinka O., (1980) : Organická chemie - 2. vyd. – Praha, SNTL, ALFA - 791 s.  
 Panchartek J., et al. (1977) : Organická chemie II- Reakční mechanismy. -Pardubice

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 37

A	B	C	D	E	FX
78.38	13.51	2.7	2.7	0.0	2.7

**Teacher:** Mgr. Alexandra Hengerics Szabó, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ VSC/15	<b>Name:</b> General Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 / 1 <b>For the study period:</b> 26 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester there will be two written assessments for 25-25 points, in order to access the oral test, students have to achieve at least 25 points from the two assessments together, i.e. 50% of the total. For the oral examination the student can get 50 points. The final evaluation result depends on the oral exam and on the written assessments (50%-50%). To achieve evaluation A 90-100% is needed, for evaluation B 80-89% is needed, for evaluation C 70-79% is needed, for evaluation D 60-69% is needed, and for evaluation E 50-59% is needed from the total number of points.	
<b>Results of education:</b> After the successful completion of the educational process, the student acquires basic principles of chemical patterns, identifies general chemical definitions and types of chemical bonds and reactions. The student understands the atomic structure, and is able to express the reaction rates and mechanism of chemical reactions. The student knows the properties of various solutions and the principles of electrochemistry. At the end the student will be able to integrate the acquired knowledge in further education.	
<b>Brief syllabus:</b> Introduction to Chemistry – History of Chemistry 2. Basic Chemical Principles and Definitions (elements, substances, molecules, Avogadro's Law). 3. Atomic Structure (discovery of electron, Rutherford and Bohr Atomic model). 4. The quantum mechanical model of the atom 5. The periodic law and Periodic Table 6. Written assessment 7. Chemical Bond, Classical Theory (Berzelius, Frankland) and Semi-Classical Theory of Chemical Bonds (Kössel and Lewis). 8. Theory of Molecular Orbitals, $\sigma$ - bonds in H <sub>2</sub> molecule, $\pi$ - bonds. 9. Types of Chemical Bonds (covalent, polar bonds, ionic bonds). 10. Chemical Reactions – rates of Chemical Reactions, Mechanism and rates, Rates Equations, Rates constant. 11. Catalysis and biocatalysis. Energetics of Chemical Reactions (#Gr, #Hr, #Sr). 12. Properties of electrolytic solutions, acids and bases.	

13. Basic Principles of Electrochemistry, electrolysis and electrochemical cells.

14. Written assessment.

**Literature:**

Kotočová A., (1993): Všeobecná chémia. Bratislava, Slovenská technická univerzita, ISBN 80 227 0560 8

Gažo J. a kol., (1981): Všeobecná a anorganická chémia. Bratislava, ALFA

Čársky P., (1985): Ab initio výpočty v chémii. Praha, SNTL, Nakladatelství technické literatury

Csányi Cs., (2002): Kémiai példatár és tesztgyűjtemény megoldásokkal. Budapest, ISBN 96 316 2112 X

Gyorbíró K., (1994): Általános kémia. Budapest, Műszaki Könyvkiadó, ISBN 00 0255 3

Kiss Zs., (2004): Összefoglaló feladatgyűjtemény kémiából - Megoldások. Budapest, Nemzeti Tankönyvkiadó, ISBN 963 19 5394 7

Rózsahegyi M.,(1996): Érettségi felvételi feladatok. Mozaik Oktatási Stúdió, ISBN 963 697 017 3

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 44

A	B	C	D	E	FX
11.36	15.91	22.73	29.55	15.91	4.55

**Teacher:** Dr. habil. PaedDr. György Juhász, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ ZCH/18	<b>Name:</b> Green Chemistry
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b>	
<b>Results of education:</b>	
<b>Brief syllabus:</b>	
<b>Literature:</b>	
<b>Language, knowledge of which is necessary to complete a course:</b>	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 5	
a	n
100.0	0.0
<b>Teacher:</b>	
<b>Date of last update:</b> 12.05.2023	
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.	

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KCH/CHdb/ ZCM/16		<b>Name:</b> Chemometrics and the basics of quality systems in laboratory			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present					
<b>Number of credits:</b> 1					
<b>Recommended semester/trimester of study:</b> 4.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> During the semester, the students will be delivered two written tests. For the successful termination of the course, one has to gather at least 50 points (i.e. 50% of the maximum count of points). For the final classification to be A one has to obtain 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.					
<b>Results of education:</b> Attending the course the students will able to make calculuses of analytical chemistry analyses and can make statistical calculations of the result and interpret them. The students get acquainted with the recent trends of laboratory quality management.					
<b>Brief syllabus:</b> bsbsjd					
<b>Literature:</b> Karlíček R., a kol. (2009): Analytická chemie pro farmaceuty. Karolinum, ISBN 97 8802 46 1453 3 Majer J., (1989) : Analytická chémie. - 1. vyd. - Martin : Osveta n.p., - 368 s. Holzbecher Z., Churáček J., (1987) : Analytická chemia. - 1. vyd. – Praha, SNTL - Nakladatelství technické literatury, - 663 s. Barcza L., (2006): A mennyiségi kémiai analízis gyakorlati kézikönyve. Medicina Kiadó, ISBN: 963 2429 61 3 Barcza L., (2007): Kvantitatív analitikai kémia. Budapest, Semmelweis Kiadó, ISBN 978 963 9656 73 4 Barcza L., Buvári Á., (2009): A minőségi kémiai analízis. Medicina Könyvkiadó, ISBN 978 9 6 322 6186 7					
<b>Language, knowledge of which is necessary to complete a course:</b> -					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 23					
A	B	C	D	E	FX
56.52	26.09	4.35	4.35	8.7	0.0
<b>Teacher:</b> doc. Ing. Ondrej Hegedús, PhD.					

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ ZCV/15	<b>Name:</b> The Basics of Chemistry Calculuses
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester the students will be issued a test of maximum 50 points, while another amount of 50 points can be acquired for his/her homework. For a successful completion of the course one has to gather at least 50 point, i.e. 50% of the total points possible. For the final classification to be A one has to acquire 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> Within the educational process the students acquire knowledge about the relation between fundamental physical quantities and become capable of using basic chemical calculations, needed for the most common laboratory tasks	
<b>Brief syllabus:</b> 1. Introduction. Physical quantities and measures. 2. Quantity of substances, particle count, amount of substance, mass, volume, relations between the measures of quantity. 3. Calculation of chemical formulae and chemical equations. 4. Solutions, mass fraction and molar fraction. 5. Concentration of solutions. 6. Written test. 7. Volume fraction. 8. Solubility and the product of solubility, 9. Composition of multicomponent systems, the density of solutions. 10. Preparation of solutions. 11. Mass balance in chemical systems. 12. Conclusion.	
<b>Literature:</b> Odporúčaná literatúra: Krätsmár-Šmogrovič, J. a kol.(2007): Všeobecná a anorganická chémia. Osveta, ISBN 80 806 3245 8 Fajnor V.,( 1992) Laboratórna technika, názvoslovie a chemické výpočty. Vysokoškolské skriptá, UK Bratislava, ISBN 80 223 0436 0	



Sokolík J., (2012) Názvoslovie a príprava vybraných anorganických látok, UK Bratislava, ISBN 978 80 223 2913 2  
 Fajnor V., (1998): Všeobecná a anorganická chémia. Vysokoškolské skriptá - 1. vyd. – UK Bratislava, 266 s. - ISBN 80-223-1257-6  
 Kiss Zs.,( 2004): Összefoglaló feladatgyűjtemény kémiából – Megoldások. Budapest, Nemzeti Tankönyvkiadó,. ISBN 963 19 5394 7  
 Kotočová A., Valigura D.,(1993): Všeobecná chémia- Návody na laboratórne cvičenia. Bratislava: Slovenská technická univerzita, ISBN 80 227 0560 8  
 Sík J., (1992): Kémiai számítások képletgyűjteménye. Budapest: Műszaki Könyvkiadó, ISBN 963 10 9419 7

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 41

A	B	C	D	E	FX
17.07	24.39	26.83	4.88	21.95	4.88

**Teacher:** Mgr. Katarína Szarka, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/CHdb/ ZLT/15	<b>Name:</b> Basic Laboratory Skills
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> During the semester the students will be issued two written tests each of maximum 30 points, while another amount of 40 points can be granted for his/her laboratory protocols. The final classification is obtained as the sum of points obtained for the written tests (60%) and from the classification of laboratory protocols (40%). For the final classification to be A one has to acquire 90-100% of the total points, for B 80-89%, for C 70-79%, for D 60-69% and for E 50-59%.	
<b>Results of education:</b> Upon completing the educational process the students acquire basic laboratory skills and become trained for handling the basic laboratory equipment. They become acquainted with basic laboratory procedures which they will be able to conduct by themselves with emphasis on adhering to laboratory safety regulations and rules.	
<b>Brief syllabus:</b> 1. Introduction. Laboratory regulations. 2. Safety and health Safety regulations and health protection in chemical laboratories, hygiene prescriptions, first aid in case of laboratory accident, fire safety. 3. Materials for laboratory use – glass, porcelain, rubber, cork, paper, metals, alloys and other materials. 4. Basic laboratory operations – measurement of mass, volume and density, dissolving, heating, cooling, precipitating, drying. 5. Cleaning and separation methods - decantation, centrifugation, crystallization, sublimation, distillation. 6. Filtration – classical and under low pressure. 7. Distillation under atmospheric pressure and vacuum distillation. 8. Solubility and solubility product. 9. Crystallization. 10. Sublimation. 11. Determination of density using a pycnometer. 12. Conductometry 13. Conclusion.	
<b>Literature:</b>	

**Odporúčaná literatúra:**

Fajnor V., a kol. ( 1992) : Laboratórna technika, názvoslovie a chemické výpočty. UK Bratislava, ISBN 80 223 0436 0

Sokolík J., a kol. (2012): Názvoslovie a príprava vybraných anorganických látok. UK Bratislava, ISBN 978 80 223 2913 2

Kiss Zs., (2004) : Összefoglaló feladatgyűjtemény kémiából - Megoldások. Budapest, Nemzeti Tankönyvkiadó, ISBN 963 19 5394 7

Kotočová A., Valigura D., (1993) : Všeobecná chémia - Návody na laboratórne cvičenia. Bratislava STU, ISBN 80 227 0560 8

Sík J., (1992): Kémiai számítások képletgyűjteménye. Budapest, Műszaki Könyvkiadó, ISBN 963 10 9419 7

**Language, knowledge of which is necessary to complete a course:****Notes:****Evaluation of subjects**

Total number of evaluated students: 40

A	B	C	D	E	FX
40.0	40.0	15.0	5.0	0.0	0.0

**Teacher:** Mgr. Katarína Szarka, PhD., Mgr. Alexandra Hengerics Szabó, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KMF/CITdb/ NJ/15		<b>Name:</b> Reading			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present					
<b>Number of credits:</b> 2					
<b>Recommended semester/trimester of study:</b> 1.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> Written test, evaluation in accordance with the study regulation of JSU					
<b>Results of education:</b> The student understands authentic German texts. He is able to interpret and to analyze them.					
<b>Brief syllabus:</b> Working with various types of texts.					
<b>Literature:</b> WESTHOFF, G.: Fertigkeit Lesen. Fernstudieneinheit 17. Berlin/München: Langenscheidt, 1997. Rôzne učebné texty (podľa aktuálnej tematiky) Texty z novín, časopisov (podľa aktuálnej tematiky)					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 72					
A	B	C	D	E	FX
44.44	25.0	13.89	11.11	5.56	0.0
<b>Teacher:</b>					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/Chdb/ OK1/19	<b>Name:</b> Conversation of Chemistry Disciplines in Slovak Language 1
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> 80% of presence on the seminars, working out the final seminar project and its presentation.	
<b>Results of education:</b> Through the subject student able to use slovak terminology of gyeneral chemistry.	
<b>Brief syllabus:</b> 1. Introduction to Chemistry – History of Chemistry 2. Basic Chemical Principles and Definitions. 3. Atomic Structure. 4. The quantum mechanical model of the atom 5. The periodic law and Periodic Table. 6. Chemical Bond, Classical Theory (Berzelius, Frankland) and Semi_Classical Theory of Chemical Bonds (Kössel and Lewis). 7. Theory of Molecular Orbitals. 8. Types of Chemical Bonds. 9. Chemical Reactions – rates of Chemical Reactions, Mechanism and rates, Rates Equations, Rates constant. 10. Energetics of Chemical Reactions (#Gr, #Hr, #Sr). 11. Catalysis and biocatalysis. 12. Properties of electrolytic solutions, acids and bases. 13. Basic Principles of Electrochemistry, electrolysis and electrochemical cells.	
<b>Literature:</b> Kotočová A., (1993): Všeobecná chémia. Bratislava, Slovenská technická univerzita, ISBN 80 227 0560 8 Gažo J. a kol., (1981): Všeobecná a anorganická chémia. Bratislava, ALFA	
<b>Language, knowledge of which is necessary to complete a course:</b>	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 11	

a	n
100.0	0.0
<b>Teacher:</b> Mgr. Andrea Vargová, PhD.	
<b>Date of last update:</b> 26.06.2023	
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.	

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/Chdb/ OK2/19	<b>Name:</b> Conversation of Chemistry Disciplines in Slovak Language 2
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> 80% of presence on the seminars, working out the final seminar project and its presentation.	
<b>Results of education:</b> Through the subject student is able to use slovak terminology of inorganic chemistry.	
<b>Brief syllabus:</b> 1. The periodic system of elements and the electron structure of their valence shells. 2. Compounds in general, lattice and bond types, characteristics and categories of compounds. 3. Hydrogen, bond types, occurrence, preparation, its compounds and isotopes. 4. General properties of metals (including transition metals). 5. Coordination compounds. 6. Alkali metals – elements of group I of the periodic system, bond types, compounds, the subgroup of copper. 7. Alkaline earth metals – elements of group II of the periodic system, bond types, compounds, the subgroup of zinc. 8. Elements of group III of the periodic system, bond types, compounds, the subgroup of scandium, hybridization types. 9. Elements of group IV of the periodic system, bond types, compounds, the subgroup of titanium. 10. Elements of group V of the periodic system, bond types, compounds, the subgroup of vanadium. 11. Elements of group VI of the periodic system, bond types, compounds, the subgroup of chromium. 12. Elements of group VII of the periodic system, bond types, compounds, the subgroup of manganese. 13. Elements of group VIII of the periodic system and their compounds.	
<b>Literature:</b> Krätsmár - Šmogrovič J. a kol., (2007): Všeobecná a anorganická chémia. Osveta, ISBN 80 806 3245 8 Fajnor V., (1998) : Všeobecná a anorganická chémia. - 1. vyd. – Bratislava, Univerzita Komenského - 266 s. - ISBN 80-223-1257-6 Gažo J., Kohout J., Serátor M., (1981) : Všeobecná a anorganická chémia. Bratislava, ALFA - 804 s.	

Lukeš I., (2009): Systematická anorganická chemie. - 1. vyd. – Praha, Nakladatelství Karolinum - 230 s. ISBN 978-80-246-1614-8.  
Zikmund M.,(1995): Anorganická chémia. Bratislava : Univerzita Komenského, ISBN 80-223-0919-2

**Language, knowledge of which is necessary to complete a course:**

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 8

a	n
100.0	0.0

**Teacher:** Mgr. Andrea Vargová, PhD.

**Date of last update:** 26.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/DLI db/ NJ/15	<b>Name:</b> Children literature
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final exam consists of written and oral part. The condition for granting credits is to obtain min. 60% of written part and an evaluation of A to E from oral part. The final evaluation takes into account the interim assessment, which includes interim analyzes, activity and participation.	
<b>Results of education:</b> Students get knowledge of the basic genres of children's literature. He is able to work with authentic texts. He understands the literary text as a medium of education, which plays a special role especially in the teaching of German as foreign language. The student should be capable to use creatively this kind of literature in teaching practice.	
<b>Brief syllabus:</b> The course concentrates on the definition of children's literature (with special attention to the findings and assumptions of German theorists), its basic features, the historical development of children's literature in a global perspective and in the special context of German speaking countries, the types and genres of children's literature (picture books, comics, fairy tales – folk tales and literary tales, legends, fantasy fiction for children, realistic fiction for children, adventure fiction, poetry, etc.). The course highlights the importance of children's literature in teaching German as a foreign language with special attention to reading comprehension and the usage of children's literature in teaching practice.	
<b>Literature:</b> WANGERIN, Wolfgang atd'.: Kinder und ihre Medien. 91-103, 110-116. s. In: NEUMANN, Karl (ed.): Kindsein. Zur Lebenssituation von Kindern in modernen Gesellschaften. 1. vyd. - Göttingen: Vandenhoeck & Ruprecht 1981. 218 s. - ISBN 3-525-33461-3. ( <a href="http://digi20.digitale-sammlungen.de/de/fs1/object/goToPage/bsb00051603.html?pageNo=91&amp;zoom=1.00">http://digi20.digitale-sammlungen.de/de/fs1/object/goToPage/bsb00051603.html?pageNo=91&amp;zoom=1.00</a> ) VOGDT, Ines-Bianca: Wunderhorn und Sprachgitter. Geschichte der intentionalen Kinderlyrik seit dem 18. Jahrhundert. 1. vyd. - München: Fink 1998. - 240 s. - ISBN 3-7705-3315-1. ( <a href="http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00041055_00001.html?zoom=1.00">http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00041055_00001.html?zoom=1.00</a> ) O'SULLIVAN, Emer: Zwischen Erziehung und Unterhaltung. Zur Verortung der Kinderliteratur von Robinson der Jüngere bis Harry Potter. 1. vyd. - München: Fink 2006. - 147-166 s. -	

ISBN 978-3-7705-4364-9. ( [http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00052201\\_00145.html?zoom=1.00](http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00052201_00145.html?zoom=1.00) )

O'SULLIVAN, Emer: Literaturbeziehungen. Repräsentationen fremder Nationen und Kulturen in der Literatur. In: FAULSTICH, Werner (ed.): Beziheungskulturen. 1. vyd. - München: Fink 2007. - 166-183 s. - ISBN 978-3-7705-4385-4. ( [http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00065944\\_00164.html?zoom=1.00](http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00065944_00164.html?zoom=1.00) )

LUSERKE-JAQUI, Matthias (ed.): Schule erzählt. Literarische Spiegelbilder im 19. und 20. Jahrhundert. 1. vyd. - Göttingen: Vandenhoeck & Ruprecht 1999. 160 s. - ISBN 3-525-34016-8. ( [http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00044538\\_00001.html?zoom=1.00](http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00044538_00001.html?zoom=1.00) )

**Language, knowledge of which is necessary to complete a course:**

German

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 29

A	B	C	D	E	FX
10.34	20.69	31.03	20.69	6.9	10.34

**Teacher:** Dr. habil. László V. Szabó, PhD.

**Date of last update:** 17.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ FONdb/NJ/15	<b>Name:</b> Phonetics and Phonology of German
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final exam consists of written and oral part. The condition for granting credits is to obtain min. 60% of written part and an evaluation of A to E from oral part.	
<b>Results of education:</b> The student acquires an extensive knowledge of the field of phonetics and phonology. He is able to acquire the basic specifications of the German pronunciation and differ these from the specific features of the Hungarian or Slovak. He has mastered the transcription. He has knowledge of the function of the speech organs. He is able to identify, classify and describe the German phonation.	
<b>Brief syllabus:</b> Phonetics and phonology and their status in the system of sciences. Object and tasks of phonetics and phonology. Phoneme, sound, allophone. Areas of phonetics. The physiological or articulatory phonetics. Elementary units of the segmental structure: the articulation of speech sounds. Vowels and consonants. According problems of designation. The (physiological) system of vowels because of their articulation features. Monophtonge and diphthongs. General rules of pronunciation of consonants. The (physiological) system of consonants because of their articulation features. The normalized Standardlautung in German: general rules of pronunciation of vowels. Associative system of sounds: coarticulation and assimilation. Components and communicative role of suprasegmental structure. Basic concepts of acoustic phonetics. Communicative role of components of suprasegmental structure.	
<b>Literature:</b> ADAMCOVÁ, L.: Phonetik der deutschen Sprache. Bratislava: UK, 1997. BRENNER, K./HUSZKA, B./WERK-MARINKÁS, Cs.: Deutsche Phonetik. Eine Einführung. Budapest/Veszprém: Bölcsész Konzorcium, 2006. RAUSCH, R./RAUSCH, I.: Deutsche Phonetik für Ausländer. 3. Aufl. Langenscheidt: München, 1993. UZONYI, Pál: Phonetik und Phonologie des Deutschen. Budapest: Akadémiai Kiadó, 1998. - 270. - ISBN 9630575647.	

HIRSCHFELD, Ursula: Phonetik sim sala bim : Ein Übungskurs für Deutschlernende. - 1. vyd. - Berlin: Langenscheidt, 1998. - 128 s. - ISBN 3468905408.

**Language, knowledge of which is necessary to complete a course:**

German

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 41

A	B	C	D	E	FX
26.83	17.07	12.2	12.2	24.39	7.32

**Teacher:** prof. Erzsébet Drahotová-Szabó, PhD.

**Date of last update:** 09.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KMF/ HOVdb/NJ/15		<b>Name:</b> Speaking			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present					
<b>Number of credits:</b> 2					
<b>Recommended semester/trimester of study:</b> 4.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> Active participation in the seminars, 2x presentation on selected topics					
<b>Results of education:</b> The student recognizes expression in different communication data. He is able to identify the structure of the oral statement and make independent statements.					
<b>Brief syllabus:</b> Speaking practice Pronunciation exercises Argumentation Discussion Presentation of own point					
<b>Literature:</b> LODEWICK, K.: DSH-Training. Texte aus Wissenschaft und Gesellschaft. 2. Aufl. Göttingen: Fabouda Verlag, 1997. WILLKOP, E.-M./WIEMER, C. u.a.: Auf neuen Wegen. Deutsch als Fremdsprache für die Mittelstufe und Oberstufe. München: Hueber Verlag, 2003. Rôzne učebné texty (podľa aktuálnej tematiky) Texty z novín, časopisov (podľa aktuálnej tematiky)					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 32					
A	B	C	D	E	FX
34.38	15.63	18.75	6.25	21.88	3.13
<b>Teacher:</b>					
<b>Date of last update:</b> 09.06.2023					

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/IKUdb/ NJ/15	<b>Name:</b> Intercultural Communication
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Presentation to a selected topic (30% of total), final test (30% of total), seminar paper (40% of total).	
<b>Results of education:</b> The students acquire basic theories and concepts from the field of intercultural communication and the basics of intercultural competence.	
<b>Brief syllabus:</b> The concept of culture and the communication process; Cultural specifics; Language and nonverbal communication, traditions and values, social groups and relations; Cultural standards, stereotypes and prejudices; Influence of the media on and attitudes; Multiculturalism - interculturalism - transculturality	
<b>Literature:</b> BOLTEN, J.: Interkulturelle Kompetenz. Bonn: Landeszentrale für politische Bildung Thüringen, 2007. ( <a href="http://www.ikkompetenz.thueringen.de/downloads/1210Bolten_Ik_Kompetenz_Vorversion_5Aufl.pdf">http://www.ikkompetenz.thueringen.de/downloads/1210Bolten_Ik_Kompetenz_Vorversion_5Aufl.pdf</a> ) HOSCH, Wolfram: Bilder in der Landeskunde: Langenscheidt, 2000. - 192. - ISBN 3468496605. FÖLDES, Csaba: Interkulturelle Kommunikation: Positionen zu Forschungsfragen, Methoden und Perspektiven. - 1. vyd. : Universitätsverlag Veszprém, 2007. - 63 s. - ISBN 978-963-9696-10-5.	
<b>Language, knowledge of which is necessary to complete a course:</b> German	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 27	

A	B	C	D	E	FX
14.81	14.81	29.63	7.41	22.22	11.11
<b>Teacher:</b>					
<b>Date of last update:</b> 17.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ITGdb/ NJ/15	<b>Name:</b> ICT for students of German studies
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Presentation (40%) and written paper (60%). Evaluation in accordance with the study regulation of JSU	
<b>Results of education:</b> Students have practical knowledge of the use of information technology in teaching of German. They are able to look into online resources for relevant information and adapt them to their own needs. They know the key formats, technologies and media and have knowledge in the field of e-learning.	
<b>Brief syllabus:</b> 1. Modern information technologies in the educational process 2. Specific information for Germanists 3. Recherching in Online Resources 4. Important formats and their peculiarities 5. Electronic Books 6. Electronic databases and their use 7. E-learning, policies, practices 8. Video and Audio in German lessons 9. Web 2.0 and its use in German lessons 10. Lesson planning based on ICT	
<b>Literature:</b> Häfele, H.: 101 e-Learning Seminarmethoden. Methoden und Strategien für die Online- und Blended-Learning-Seminarpraxis. Edition Training: 2012 Rey, G-D.: E-Learning. Theorien, Gestaltungsempfehlungen und Forschung. Huber, 2009 Meister, H./Shalaby, D.: E-Learning: Handbuch für den Fremdsprachenunterricht. Deutsch als Fremdsprache. Hueber, 2014 Hugger, K-U.: Digitale Lernwelten: Konzepte, Beispiele und Perspektiven. VS Verlag, 2010 Wagner, J./Heckmann, V.: Web 2.0 im Fremdsprachenunterricht: Ein Praxisbuch für Lehrende in Schule und Hochschule. VWH, 2012 Strasser, T.: Moodle im Fremdsprachenunterricht: Blended Learning als innovativer didaktischer Ansatz oder pädagogische Eintagsfliege? VWH, 2010	

Jankowski, R./von Osthoff, M./Zöller-Greer, P.: Virtuelles Klassenzimmer und Teleteaching für die Praxis: Do-it-Yourself-eTeaching im Unterrichtseinsatz. Composita, 2012					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 18					
A	B	C	D	E	FX
100.0	0.0	0.0	0.0	0.0	0.0
<b>Teacher:</b>					
<b>Date of last update:</b> 17.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KCH/KCH/ CHdb/BPO/15	<b>Name:</b> Bacalar Thesis and Its' Defens
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> <b>Recommended extent of course ( in hours ):</b> <b>Per week: For the study period:</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 5., 6..	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Submission of the final dissertation. The referee's and supervisor's positive reviews. The successful defense of the bacselor dissertation	
<b>Results of education:</b> The candidate will learn the rules of the preparation, and independently will make and submit the final dissertation	
<b>Brief syllabus:</b> 1. The type and administration of the dissertation 2. Structure of the dissertation 3. The arrangement of the chapters and formation the essay 4. Citations and bibliographic part, literature list 5. Introduction and importance of the selected topic 6. Formation of the hypothesis, the goal of the study and the objective 7. Methodology of the topics. The selection of the methods 8. Discussion and summary of the results. Interpretation and summary 9. Conclusion. Supplements 10. Submission of the dissertation, license agreement, statement of honour	
<b>Literature:</b> Smernica rektora Univerzity J. Selyeho Komárno o úprave, registrácii, sprístupnení a archivácii záverečných prác na Univerzite J. Selyeho. - Vždy aktuálne vydanie Smernice Katuščák D. (2008) : Ako písať záverečné a kvalifikačné práce. - 5. vyd. - Nitra : Enigma, 164 s. - ISBN 978 80 89 132 45 4 Albert S. (2001) : Písanie záverečnej práce. Košice, Technická univerzita – 47 s. - ISBN 80 709 9727 3 Albert S. (2007) : Dolgozatok írása. Komárno SJE, ISBN 978-80-89234-22-6 Odborná literatúra – podľa schválenej témy bakalárskej práce.	
<b>Language, knowledge of which is necessary to complete a course:</b>	
<b>Notes:</b>	

<b>Evaluation of subjects</b>					
Total number of evaluated students: 5					
A	B	C	D	E	FX
40.0	40.0	0.0	20.0	0.0	0.0
<b>Teacher:</b>					
<b>Date of last update:</b> 12.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ KUGdb/NJ/15	<b>Name:</b> Culture and History of German speaking countries
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Active participation in the seminars and final test. Condition of the grant of credits: at least 60% of the total score.	
<b>Results of education:</b> The students gain insight into the history of the German language area. They acquire basic factual knowledge and understand historical and cultural contexts.	
<b>Brief syllabus:</b> Germanic tribes, Frankish Empire German states in the late Middle Ages Great dynasties: Ottos, Staufer, Luxembourg Humanism and Renaissance, Reformation Counter-Reformation and the Thirty Years War Holy Roman Empire and the Habsburgs Revolution years Prussia; Unification of Germany Germany and Austria at the turn of the century, the First World War Weimar Republic, between years 1933 - 1939 Second World War ; time after war German States after 1989	
<b>Literature:</b> MAI, Manfred: Deutsche Geschichte. - 1. vyd. - Gelberg: Gulliver, 2006. - 196s. - ISBN 978-3-407-75524-7. MÜLLER, Reiner: Deutsche Geschichte in Quellen und Darstellung. Band 1-11. Stuttgart : Philipp Reclam jun., 2002. - 472. - ISBN 315017001x. Fragen an die deutsche Geschichte / Bonn Deutscher Bundestag Referat Öffentlichkeitsarbeit, Bonn Deutscher Bundestag Referat Öffentlichkeitsarbeit, Bonn Deutscher Bundestag Referat Öffentlichkeitsarbeit. - Köln: Sonderausgabe, 1990. - s. 502. - ISBN 392452159x.	
<b>Language, knowledge of which is necessary to complete a course:</b> German	

<b>Notes:</b>					
<b>Evaluation of subjects</b>					
Total number of evaluated students: 30					
A	B	C	D	E	FX
46.67	20.0	23.33	6.67	0.0	3.33
<b>Teacher:</b>					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ LEXdb/NJ/15	<b>Name:</b> Lexicology
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final exam consists of written and oral part. The condition for granting credits is to obtain min. 60% of written part and an evaluation of A to E from oral part.	
<b>Results of education:</b> The student has knowledge in the field of German vocabulary. He knows the basic categories, the structure, function and the basic principles of German word formation and is able to demonstrate this using the example of lexical units.	
<b>Brief syllabus:</b> Subject and object of lexicology The theories of lexicology The concept of motivation The theories of meaning Connotative and denotative meaning of lexemes The structure of word meaning The methods of description and illustration of the lexical meaning The prototype theory Lexical semantic meaning conditions The breakdown of vocabulary Changes in vocabulary	
<b>Literature:</b> HESSKY, R./KNIPF, E.: Lexikologie I-II. Budapest: Holnap Kiadó, 1998. LINKE, A./NUSSBAUMER, M./PORTMANN, P. R.: Studienbuch Linguistik. 5., erw. Aufl. Tübingen: Niemeyer, 2004. LUTZEIER, P. R.: Lexikologie. Ein Arbeitsbuch. Tübingen: Narr, 1995. SCHIPPAN, T.: Lexikologie der deutschen Gegenwartssprache. Tübingen: Niemeyer, 1992. SCHLAEFER, T.: Lexikologie und Lexikographie: Eine Einführung am Beispiel deutscher Wörterbücher. 2., durchges. Aufl. Berlin: Erich Schmidt, 2008.	
<b>Language, knowledge of which is necessary to complete a course:</b> German	

<b>Notes:</b>					
<b>Evaluation of subjects</b>					
Total number of evaluated students: 141					
A	B	C	D	E	FX
29.08	21.28	21.28	15.6	9.93	2.84
<b>Teacher:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc.					
<b>Date of last update:</b> 17.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KNJ/LIT1db/ NJ/15	<b>Name:</b> Literature of German speaking countries 1
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final exam consists of written and oral part. The condition for granting credits is to obtain min. 60% of written part and an evaluation of A to E from oral part.	
<b>Results of education:</b> The students will acquire knowledge about the different stages of development of German literature. They know the cultural, the political-historical and linguistic features of each era. You can recognize the typical literary genres and the most important works and authors.	
<b>Brief syllabus:</b> Enlightenment in the German-speaking countries. Historical and philosophical background. Immanuel Kant: "What is Enlightenment?" Literary theories of the Enlightenment: - Johann Christoph Gottsched; - Theoretical views of Johann Jakob Bodmer and Johann Jakob Breitinger; - Theoretical views of Gotthold Ephraim Lessing and Moses Mendelssohn; - Lessing and the drama of the Enlightenment (the idea of the National Theatre). Lessing's "Emilia Galotti", "Nathan the Wise" Friedrich Gottlieb Klopstock and Johann Gottfried Herder Sturm und Drang, "Intrigue and Love", "The Robbers" by Friedrich Schiller Johann Wolfgang Goethe: "Prometheus", "The Sorrows of Young Werther". Weimar Classicism, forms and genres, authors and works. Goethe's transition to the classics: "Roman Elegies", "Iphigenia in Tauris", "Faust"; Selected Poems. Literary cooperation between Goethe and Schiller Friedrich Schiller, "Don Carlos" Goethe: Wilhelm Meister's Apprenticeship as a development paradigm of the classic novel.	
<b>Literature:</b> JORGENSEN, Sven Aage: Geschichte der deutschen literatur. Aufklärung, Sturm und Drang, frühe Klassik. München : Beck, 1990. 670 s. ISBN-3 406 34573 5. 1. vyd NÜRNBERGER, Helmuth: Geschichte der Deutschen Literatur. München : Bayerischer Schulbuch Verlag GmbH, 2006. - 863s. - ISBN 978-3-7627-2531-2. 1. vyd.	

BAHR, E.: Geschichte der deutschen Literatur Von der Aufklärung bis zum Vormärz. 2. Bd. Tübingen: Francke Verlag, 1988. 531 s., ISBN 3-7720-1741-X.  
KAISER, Gerhard: Aufklärung Empfindsamkeit Sturm und Drang. Tübingen : Francke Verlag, 1991. 374 s. - ISBN 3-7720-1288-4. 4. vyd

**Language, knowledge of which is necessary to complete a course:**  
German

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 163

A	B	C	D	E	FX
9.82	14.72	26.38	12.88	28.22	7.98

**Teacher:** Dr. habil. László V. Szabó, PhD.

**Date of last update:** 09.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ LIT2db/NJ/15	<b>Name:</b> Literature of German speaking countries 2
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final exam consists of written and oral part. The condition for granting credits is to obtain min. 60% of written part and an evaluation of A to E from oral part.	
<b>Results of education:</b> The students will acquire knowledge about the different stages of development of German literature. They know the cultural, the political-historical and linguistic features of each era. You can recognize the typical literary genres and the most important works and authors.	
<b>Brief syllabus:</b> The social situation in the German-speaking countries after the French Revolution. The philosophy of the era and the establishment of German Studies as a scientific discipline, Periodisation, literary centers. The origins of women's emancipation. The Berlin salons, romantic poetics, theoreticians and theoretical work of early Romanticism. "Folk tales", literary centers of early Romanticism. Between Classicism and Romanticism. High and Late Romanticism. Between romanticism and realism, Restauration streams after 1815. Emancipation movements - Young Germany, Vormärz. Biedermeier, Vienna Volkstheater. Realism, social situation in Germany after 1848. Prose, lyrical works, dramatic works.	
<b>Literature:</b> SCHMITZ-EMANS, Monika: Einführung in die Literatur der Romantik. 3. vyd. Darmstadt: Wissenschaftliche Buchgesellschaft, 2009. 167 s. ;. - ISBN 978-3-534-23056-3. BORRIES, Erika & Ernst von: Deutsche Literaturgeschichte . Zwischen Klassik und Romantik: Hölderlin, Kleist, Jean Paul. 4. Bd. München: Deutscher Taschenbuch Verlag, 1993. 346 s. - ISBN 3-423-03344-4 BORRIES, Erika & Ernst von: Deutsche Literaturgeschichte. Romantik. 5. Bd. 2. vydl. München: Deutscher Taschenbuch Verlag 1999. 447 s. - ISBN 3-423-03345-2 CASSIRER, Ernst: Idee und Gestalt. Goethe - Schiller - Hölderlin – Kleist. Darmstadt: Wissenschaftliche Buchgesellschaft, 1971. 202 s BARK, Joachim: Biedermeier-Vormärz Bürgerlicher Realismus. Stuttgart: Ernst Klett, 1984. 173 s. - ISBN 3-12-347440-2 MARTINI, Fritz: Deutsche Literaturgeschichte : von den Anfängen bis zur Gegenwart. 19. vyd. Stuttgart: Alfred Kröner Verlag, 1991. 765 s. ;. - ISBN 3-520-19619-0	

GRAWE, Christian: Fontanes Novellen und Romane: Interpretationen. 1. vyd. - Stuttgart: Reclam 1991. - 304 s. - ISBN 978 3 15 008416 8

**Language, knowledge of which is necessary to complete a course:**

German

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 154

A	B	C	D	E	FX
14.94	18.18	24.68	19.48	17.53	5.19

**Teacher:** Dr. habil. László V. Szabó, PhD.

**Date of last update:** 09.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ LIT3db/NJ/15	<b>Name:</b> Literature of German speaking countries 3
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final exam consists of written and oral part. The condition for granting credits is to obtain min. 60% of written part and an evaluation of A to E from oral part.	
<b>Results of education:</b> The students will acquire knowledge about the different stages of development of German literature. They know the cultural, the political-historical and linguistic features of each era. You can recognize the typical literary genres and the most important works and authors.	
<b>Brief syllabus:</b> 1. Naturalism, theoretical foundations and media (historical, social and cultural background, positivism, the theory of environment of Hippolytus Taine, Auguste Comte and his law of the three stages, positivist literature of Wilhelm Scherer, the impact of new media. Photography, Phonograph , cinematography. 2. Naturalism (Emile Zola, experimental novel, Henrik Ibsen, eras and centers of German naturalism, poetic naturalism. 4-5. Literature of the Fin de Siècle: a) Contexts of modernization and urbanization, film and theatre, psychology and psychoanalysis, "modern" individualism b) Aspects of literature, "disintegration" of the subject, Young Wien, decadence 6. Impressionism 7. Women as writer, emancipation of women (Ricarda Huch, Else Lasker-Schüler, Lou Andreas-Salomé) 8-9. Expressionism: a) The concept of avant-garde, expressionist groups and magazines, key Concepts b) Key figures, poetics of Expressionism 10. German Literature in Prague 11-12. The phenomenon called Kafka (stories and fragments of novels)	
<b>Literature:</b> LEISS, Ingo/Stadler, Hermann: Deutsche Literaturgeschichte. Wege in die Moderne 1890-1918. 8. Bd, München: Deutscher Taschenbuch Verlag, 1997. 442 s. - ISBN 3-423-03348-7	

ZEMAN, Herbert: Literaturgeschichte Österreichs. Von den Anfängen im Mittelalter bis zur Gegenwart. Graz: Akademische Druck. U. V., 1996. 604 s  
MARTINI, Fritz: Deutsche Literaturgeschichte. Von den Anfängen bis zur Gegenwart. 19. vyd. Stuttgart: Alfred Kröner Verlag, 1991. 765 s. - ISBN 3-520-19619-0  
VAJDA, Károly: Diesseits, Jenseits, Abseits. Zur Apokalyptik des Expressionismus. In: Jahrbuch der ungarischen Germanistik 2013. Budapest-Bonn, s. 108-129. ISSN 1217-0216  
VAJDA, Károly: Literaturontologische Überlegungen zu Arthur Schnitzlers "Der Sohn", 2014. In: Signaturen des Geschehens: Ereignisse zwischen Öffentlichkeit und Latenz. - Bielefeld : Transcript. - ISBN 978-3-8376-2606-3, P. 69-89.

**Language, knowledge of which is necessary to complete a course:**

German

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 155

A	B	C	D	E	FX
8.39	24.52	25.81	25.81	6.45	9.03

**Teacher:** Dr. habil. László V. Szabó, PhD.

**Date of last update:** 09.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ LKObd/NJ/15	<b>Name:</b> Comparative Literature
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> The final test takes place in the form of a written work (analysis of a selected work) and an oral exam. The conditions for the granting of credits is to obtain min. 65% points from all activities. The final evaluation also includes active participation in the classroom.	
<b>Results of education:</b> The students are able to analyze and to interpret literary texts in inter-literary and intercultural contexts. They have knowledge about the problems of literary contacts between Hungary and Germany.	
<b>Brief syllabus:</b> 1. Introduction to the phenomena of comparative literature 2. Literature as part of Cultural Studies 3. German literature and its reception in Hungary 4. The Hungarian literature and its reception in the German-speaking countries 5. German-Hungarian inter-literary and intercultural interaction 6-8. The current German-language literature, Hungarian literature and their mutual relations 9. Notion of identity in literature 10. Bilingual writers, bilingualism in literary works 11-12. Minority identities in literary text 13. Identification and identity in a multicultural context	
<b>Literature:</b> ĀURISĀN, Dionýz: Ősszehasonlító irodalomkutatás. Budapest: Gondolat 1977, ISBN 963 280 482 1 FRIED, István: Kelet- és Közép-Európa között. Irodalmi párhuzamok és szembesítések a kelet-közép-európai irodalmak köréből. Budapest: Gondolat 1986., ISBN 963 281 663 3 SZEGEDY-MASZÁK, Mihály: "Minta a szőnyegen". A műértelmezés esélyei. Budapest: Balassi Kiadó 1995, ISBN 963 506 057 2 GÖRLING, Reinhold: Heterotopia. Lektüren einer interkulturellen Literaturwissenschaft. München: Fink 1997, ISBN 3-7705-3177-9 ( <a href="http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00041043_00001.html?zoom=1.00">http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00041043_00001.html?zoom=1.00</a> )	

<p>JÄGER, Ludwig (Hrsg.): Germanistik in der Mediengesellschaft. München: Fink 1994, ISBN 3-7705-2981-2 ( <a href="http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00040898_00001.html?zoom=1.00">http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00040898_00001.html?zoom=1.00</a> )</p>					
<p><b>Language, knowledge of which is necessary to complete a course:</b> German</p>					
<p><b>Notes:</b></p>					
<p><b>Evaluation of subjects</b> Total number of evaluated students: 25</p>					
A	B	C	D	E	FX
12.0	20.0	32.0	12.0	16.0	8.0
<p><b>Teacher:</b> Dr. habil. László V. Szabó, PhD.</p>					
<p><b>Date of last update:</b> 17.05.2023</p>					
<p><b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.</p>					



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KNJ/ MORdb/NJ/15	<b>Name:</b> Morphology of German Language
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final exam consists of written and oral part. The condition for granting credits is to obtain min. 60% of written part and an evaluation of A to E from oral part.	
<b>Results of education:</b> The student has knowledge in the field of linguistic disciplines: terminology, methods, paradigms, linguistic faculties. He knows the basic linguistic concepts, with particular regard to the key issues of the morphology of the German.	
<b>Brief syllabus:</b> The term morphology and their status in the system of sciences Fundamentals of morphology: morpheme and its types. Morphological methods. Synthetic and analytical forms. Speech of Germans, their classification and problems of speech classes. Classification of nouns according to morphological, semantic and syntactic aspects. Noun declension. Grammatical categories of the noun: number, gender, case. The noun and its distributions: valence, determinants and quantifiers. Pronouns. Exophoric, anaphoric and cataphoric pronouns. Declension of pronouns. Form system and use of adjectives. Form system and use of numerals. Form of the verb system. Conjugation of the verb. Grammatical categories of the verb: number, tense, person, mood. Indicative and passive. Type of action verbs. Functional verbal structures. Sentence construction ability of verbs. Valence and distribution of verbs. Form system and use unflektierbarer speech. Form system and use of the adverbial and particles. Form system and use of prepositions. Form system and use of conjunctions.	
<b>Literature:</b> HELBIG, G./BUSCHA, J.: Deutsche Grammatik. Ein Handbuch für den Ausländerunterricht. Verfügbare Auflagen.	

HENTSCHEL, E./WEYDT, H.: Handbuch der deutschen Grammatik. 3. Aufl. Berlin/New York: De Gruyter, 2003.  
PONGÓ, S.: Grundriss der deutschen Grammatik – kontrastiv. Komárno: UJS, 2008.  
SOMMERFELDT, K.-E./STARKE, G.: Einführung in die Grammatik der deutschen Gegenwartssprache. 2., neu bearb., Aufl. Tübingen: Niemeyer, 1992.

**Language, knowledge of which is necessary to complete a course:**

German

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 159

A	B	C	D	E	FX
12.58	14.47	15.09	12.58	41.51	3.77

**Teacher:** prof. Dr. Erzsébet Komlósiné Knipf, DSc.

**Date of last update:** 09.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KMF/ MS1db/NJ/15		<b>Name:</b> Proseminar: German Grammar 1			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present					
<b>Number of credits:</b> 3					
<b>Recommended semester/trimester of study:</b> 1.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> 2x written test ( á 30 points) + written examination (60 points)					
<b>Results of education:</b> The students acquire the individual grammatical resources and therefore the ability to self-expression (oral and written expression). Based on the acquired knowledge they are able to understand the regularities of German grammar.					
<b>Brief syllabus:</b> Parts of speech - the verb					
<b>Literature:</b> HELBIG, G./BUSCHA, J.: Deutsche Übungsgrammatik. Leipzig: Langenscheidt, 1998. HOBERG, R. és U./szerk. FÖLDES, Cs.: Német nyelvtan. Budapest: Duden-Akadémiai Kiadó, 2002. KOCSÁNY, P./LÁSZLÓ, S.: Die Wortklassen des Deutschen. Budapest: Nemzeti Tankönyvkiadó, 1991. PONGÓ, S.: Grundriss der deutschen Grammatik – kontrastiv. Komárno: UJS, 2008.					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 172					
A	B	C	D	E	FX
20.35	15.7	18.02	19.19	23.26	3.49
<b>Teacher:</b>					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KMF/ MS2db/NJ/15		<b>Name:</b> Proseminar: German Grammar 2			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present					
<b>Number of credits:</b> 3					
<b>Recommended semester/trimester of study:</b> 2.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> 2x written test (á 30 points) + written examination (60 points)					
<b>Results of education:</b> The students acquire the individual grammatical resources and therefore the ability to self-expression (oral and written expression). Based on the acquired knowledge they are able to understand the regularities of German grammar.					
<b>Brief syllabus:</b> Part of Speech - Nouns					
<b>Literature:</b> HELBIG, G./BUSCHA, J.: Deutsche Übungsgrammatik. Leipzig: Langenscheidt, 1998. HOBERG, R. és U./szerk. FÖLDES, Cs.: Német nyelvtan. Budapest: Duden-Akadémiai Kiadó, 2002. KOCSÁNY, P./LÁSZLÓ, S.: Die Wortklassen des Deutschen. Budapest: Nemzeti Tankönyvkiadó, 1991. PONGÓ, S.: Grundriss der deutschen Grammatik – kontrastiv. Komárno: UJS, 2008.					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 157					
A	B	C	D	E	FX
18.47	18.47	21.66	18.47	21.66	1.27
<b>Teacher:</b> Dr. phil. Mgr. Attila Mészáros					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KNJ/MS3db/ NJ/15		<b>Name:</b> Proseminar: German Grammar 3			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present					
<b>Number of credits:</b> 3					
<b>Recommended semester/trimester of study:</b> 3.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> 2x written test ( á 30 points) + written examination (60 points)					
<b>Results of education:</b> 2x schriftliche Tests ( á 30 Punkte) + schriftlicher Abschlusstest (60 Punkte)					
<b>Brief syllabus:</b> Other Parts of speech - adjective, numerals et al					
<b>Literature:</b> HELBIG, G./BUSCHA, J.: Deutsche Übungsgrammatik. Leipzig: Langenscheidt, 1998. HOBERG, R. és U./szerk. FÖLDES, Cs.: Német nyelvtan. Budapest: Duden-Akadémiai Kiadó, 2002. KOCSÁNY, P./LÁSZLÓ, S.: Die Wortklassen des Deutschen. Budapest: Nemzeti Tankönyvkiadó, 1991. PONGÓ, S.: Grundriss der deutschen Grammatik – kontrastiv. Komárno: UJS, 2008.					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 163					
A	B	C	D	E	FX
12.27	13.5	18.4	19.02	31.9	4.91
<b>Teacher:</b>					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KMF/ MS4db/NJ/15		<b>Name:</b> Proseminar: German Grammar 4			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present					
<b>Number of credits:</b> 3					
<b>Recommended semester/trimester of study:</b> 4.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> 2x written test (á 30 points) + written examination (60 points)					
<b>Results of education:</b> The students acquire knowledge in the area of German syntax. They know the terminology and methods of parsing. They are able to carry out a complex analysis of the German sentence.					
<b>Brief syllabus:</b> Methods of parsing Sentence parts Complex sentence analysis Comparison of traditional and modern syntax					
<b>Literature:</b> HELBIG, G./BUSCHA, J.: Deutsche Grammatik. 15., durchges. Auflage. Leipzig/Berlin/ München/Wien/Zürich/New York: Langenscheidt, 1993. PONGÓ, S.: Grundriss der deutschen Grammatik – kontrastiv. Komárno: UJS, 2008. SZIGETI, I.: Lehrbuch der deutschen Syntax. Piliscsaba: Katholische Péter-Pázmány-Universität, 2002. TROŠOK, R.: Syntaktische Analysen. Nitra: UKF, 2002.					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 154					
A	B	C	D	E	FX
14.29	23.38	14.94	16.88	26.62	3.9
<b>Teacher:</b>					
<b>Date of last update:</b> 09.06.2023					

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ OBPdb/NJ/15	<b>Name:</b> Bachelor thesis and defence
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> <b>Recommended extent of course ( in hours ):</b> <b>Per week: For the study period:</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 1., 2., 3., 4., 5., 6..	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Accomplished thesis, positive evaluations from the supervisor and from the opponent. Successful defence of the thesis.	
<b>Results of education:</b> The student studies and conforms to the requirements for thesis writing, and then becomes able to work independently on a topic and to accomplish writing a thesis.	
<b>Brief syllabus:</b> 1. Types and administration of theses. 2. Thesis structure. 3. Formal organization of the parts of a thesis. 4. Citations, bibliographical entries, literature used. 5. Research related to the thesis topic. 6. Working hypothesis, the formulation of goals and tasks. 7. Research methodology. 8. Analysis and processing of results, discussion. 9. Conclusion and appendices. 10. Submitting the thesis, licence agreement, declaration of originality.	
<b>Literature:</b> Smernica rektora Univerzity J. Selyeho Komárno o úprave, registrácii, sprístupnení a archivácii záverečných prác na Univerzite J. Selyeho ESSELBORN-KRUMBIEGEL, H.: Von der Idee zum Text. Eine Anleitung zum wissenschaftlichen Schreiben. 3., überarb. Aufl. Paderborn: Ferdinand Schöningh, 2008. LANGER, I./VON THUN, F. S./TAUSCH, R.: Sich verständlich ausdrücken. 8. Aufl. München/Basel: Ernst Reinhardt Verlag, 2006. MEYER-KELLER, E.: Arbeitstechniken. Literaturwissenschaft. 4. Aufl. München: Wilhelm Fink, 1994. POENICKE, K.: Wie verfaßt man wissenschaftliche Arbeiten? Ein Leitfaden vom ersten Studiensemester bis zur Promotion. 2., neu bearb. Aufl. Mannheim: Dudenverlag, 1988. ROTHMANN, K.: Anleitung zur Abfassung literaturwissenschaftlicher Arbeiten. Arbeitstexte für den Unterricht. Stuttgart: Reclam, 1979.	
<b>Language, knowledge of which is necessary to complete a course:</b> German	
<b>Notes:</b>	



**Evaluation of subjects**

Total number of evaluated students: 762

A	B	C	D	E	FX
32.02	25.46	21.39	10.89	9.19	1.05

**Teacher:****Date of last update:** 17.05.2023**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/PISdb/ NJ/15	<b>Name:</b> Writing
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> 4x written work + active participation in the courses	
<b>Results of education:</b> The student is familiar with the spelling of the German. He can interpret and analyze oral and written texts. He knows the process for the development of writing skills. He is able to prepare a presentation on a selected topic.	
<b>Brief syllabus:</b> The writing process Texts in university studies scientific writing LSP and Specialized style Spelling	
<b>Literature:</b> BOÓCZ-BARNA, K./KÓCZIÁN, N./LEITNER, B.: Schreiben. Budapest: Nemzeti Tankönyvkiadó, 1998. ESSELBORN-KRUMBIEGEL, H.: Von der Idee zum Text. Eine Anleitung zum wissenschaftlichen Schreiben. 3., überarb. Aufl. Paderborn: Ferdinand Schöningh, 2008. KAST, B.: Fertigkeit Schreiben. Fernstudieneinheit 12. München: Goethe-Institut, 1999. LANGER, I./VON THUN, F. S./TAUSCH, R.: Sich verständlich ausdrücken. 8. Aufl. München/Basel: Ernst Reinhardt Verlag, 2006. POENICKE, K.: Wie verfaßt man wissenschaftliche Arbeiten? Ein Leitfaden vom ersten Studiensemester bis zur Promotion. 2., neu bearb. Aufl. Mannheim: Dudenverlag, 1988.	
<b>Language, knowledge of which is necessary to complete a course:</b> German	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 149	

A	B	C	D	E	FX
46.98	30.2	8.05	7.38	5.37	2.01
<b>Teacher:</b>					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KMF/ POCdb/NJ/15		<b>Name:</b> Listening			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present					
<b>Number of credits:</b> 2					
<b>Recommended semester/trimester of study:</b> 2.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> Condition for the granting of credits: a minimum of 6 tests during the semester and achieve at least 70% of the total score					
<b>Results of education:</b> The student understands authentic texts. He uses the techniques and strategies while listening to texts. He is able to understand lectures and seminars in German. He may respond in writing and orally.					
<b>Brief syllabus:</b> Strategies of listening Techniques of listening Listening exercises					
<b>Literature:</b> LODEWICK, K.: DSH-Training. Texte aus Wissenschaft und Gesellschaft. 2. Aufl. Göttingen: Fabouda Verlag, 2000. DÖMÖK, Sz.: Hören. Budapest: Nemzeti Tankönyvkiadó, 2002. BAUSCH, K. R., CHRIST, H., HÜLL, W., KRUMM, H.-J.: Handbuch Fremdsprachenunterricht. Tübingen: Francke Verlag 1989, S. 201-204, ISBN 3-7720-1788-6. DAHLHAUS, B.: Fertigkeit Hören. Fernstudieneinheit 5. Goethe-Institut München: Langenscheidt 1994, ISBN 3-468-49675-3. Materiály z televízie a z rozhlasu					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 73					
A	B	C	D	E	FX
38.36	31.51	12.33	4.11	13.7	0.0

<b>Teacher:</b>
<b>Date of last update:</b> 09.06.2023
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/PPX/ EDU/20	<b>Name:</b> Pedagogické praktikum - Používanie aplikácie EduPage
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b>	
<b>Results of education:</b>	
<b>Brief syllabus:</b>	
<b>Literature:</b>	
<b>Language, knowledge of which is necessary to complete a course:</b>	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 43	
a	n
97.67	2.33
<b>Teacher:</b>	
<b>Date of last update:</b> 25.06.2023	
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.	

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ REAdb/NJ/15	<b>Name:</b> Regional Studies on German speaking countries
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Presentation and final test. Condition of the grant of credits: at least 60% of the total score.	
<b>Results of education:</b> The student has basic knowledge in the field of history, economics, education and culture of German-speaking countries.	
<b>Brief syllabus:</b> 1. Germany after 1989. Unification of Germany 2. Berlin 3. Germany in the European Union 4. Germany as a destination of immigration 5. Provinces 6. The school system in Germany, academic mobility 7. Germany and German-Slovak / Hungarian relations 8. Austria in the 20th century 9. The Austrian German 10. Life in Austria 11. Austria and his importance for Slovakia and Hungary 12. The policy of Switzerland, cantons 13. Typically Swiss ...	
<b>Literature:</b> Tatsachen über Deutschland. URL: <a href="http://nemet.ektf.hu/files/konyvek/tatsachen_ueber_deutschland.pdf">http://nemet.ektf.hu/files/konyvek/tatsachen_ueber_deutschland.pdf</a> Andersen, Uwe/Woyke Wichard (Hrsg): Handwörterbuch des politischen Systems der Bundesrepublik Deutschland. Bonn : Bundeszentrale für politische Bildung, 2000. ( <a href="http://www.bpb.de/nachschlagen/lexika/handwoerterbuch-politisches-system/">http://www.bpb.de/nachschlagen/lexika/handwoerterbuch-politisches-system/</a> ) Multimediale Österreich-Chronik 1900–2000 (Österreichische Mediathek) ( <a href="http://www.akustische-chronik.at/1958-1959/">http://www.akustische-chronik.at/1958-1959/</a> )	
<b>Language, knowledge of which is necessary to complete a course:</b> German	

<b>Notes:</b>					
<b>Evaluation of subjects</b>					
Total number of evaluated students: 31					
A	B	C	D	E	FX
67.74	12.9	16.13	0.0	3.23	0.0
<b>Teacher:</b> Dr. phil. Mgr. Attila Mészáros					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KMF/ SJS1db/NJ/15		<b>Name:</b> Special Language Seminar 1			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present					
<b>Number of credits:</b> 2					
<b>Recommended semester/trimester of study:</b> 5.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> 4x written test (á 20 points). Evaluation in accordance with the study regulation of JSU					
<b>Results of education:</b> The students intensify knowledge in the field of language skills.					
<b>Brief syllabus:</b> Modal verbs Using Past and Present perfect Declination of adjectives Adjectives Separable and inseparable prefixes Indirect question					
<b>Literature:</b> HELBIG, G./BUSCHA, J.: Deutsche Übungsgrammatik. Leipzig: Langenscheidt, 1976. HOBERG, R. és U./szerk. FÖLDES, Cs.: Német nyelvtan. Budapest: Duden-Akadémiai Kiadó, 2002. KOCSÁNY, P./LÁSZLÓ, S.: Die Wortklassen des Deutschen. Budapest: Nemzeti Tankönyvkiadó, 1991. PONGÓ, S.: Grundriss der deutschen Grammatik – kontrastiv. Komárno: UJS, 2008. SCHULZ, H./SUNDERMEYER, W.: Deutsche Sprachlehre für Ausländer. Grammatik und Übungsbuch. Bearbeitet von Bernhard Thies. München, 1960.					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 29					
A	B	C	D	E	FX
44.83	10.34	10.34	24.14	3.45	6.9

<b>Teacher:</b>
<b>Date of last update:</b> 17.05.2023
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KMF/ SJS2db/NJ/15		<b>Name:</b> Special Language Seminar 2			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present					
<b>Number of credits:</b> 2					
<b>Recommended semester/trimester of study:</b> 6.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b> 4x schriftlicher Test (à 20 Punkte). Bewertung in Einklang mit der Studienregelung der JSU					
<b>Results of education:</b> The students intensify knowledge in the field of language skills.					
<b>Brief syllabus:</b> Modal verbs Using Past and Present perfect Declination of adjectives Adjectives Separable and inseparable prefixes Indirect question					
<b>Literature:</b> HELBIG, G./BUSCHA, J.: Deutsche Übungsgrammatik. Leipzig: Langenscheidt, 1976. HOBERG, R. és U./szerk. FÖLDES, Cs.: Német nyelvtan. Budapest: Duden-Akadémiai Kiadó, 2002. KOCSÁNY, P./LÁSZLÓ, S.: Die Wortklassen des Deutschen. Budapest: Nemzeti Tankönyvkiadó, 1991. PONGÓ, S.: Grundriss der deutschen Grammatik – kontrastiv. Komárno: UJS, 2008. SCHULZ, H./SUNDERMEYER, W.: Deutsche Sprachlehre für Ausländer. Grammatik und Übungsbuch. Bearbeitet von Bernhard Thies. München, 1960.					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 25					
A	B	C	D	E	FX
52.0	16.0	12.0	8.0	8.0	4.0

<b>Teacher:</b>
<b>Date of last update:</b> 17.05.2023
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ SYNdb/NJ/15	<b>Name:</b> German Syntax
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final exam consists of written and oral part. The condition for granting credits is to obtain min. 60% of written part and an evaluation of A to E from oral part.	
<b>Results of education:</b> The student has knowledge in the field of linguistic disciplines: terminology, methods, paradigms, linguistic faculties. He knows the basic linguistic concepts, with particular regard to the key issues of the syntax of the German language.	
<b>Brief syllabus:</b> The relationship between morphology and syntax - Morphosyntax Particularities and linguistic approach to traditional syntax Base definitions, types and syntactic usage modalities of parts of a sentence, sentence types and models of composite sentence: subordination and juxtaposition, style and syntactical usage modalities of subordinate clauses Modern Syntax theories: Germanic, European and Anglo-Saxon tradition of research Text grammar - Text Syntax Some syntactic peculiarities of spoken German	
<b>Literature:</b> HELBIG, G./BUSCHA, J.: Deutsche Grammatik. 15., durchges. Auflage. Leipzig/Berlin/München/Wien/Zürich/New York: Langenscheidt, 1993. PONGÓ, S.: Grundriss der deutschen Grammatik – kontrastiv. Komárno: UJS, 2008. SZIGETI, I.: Lehrbuch der deutschen Syntax. Piliscsaba: Katholische Péter-Pázmány-Universität, 2002. PITTNER, Karin, BERMAN, Judith: Deutsche Syntax. Ein Arbeitsbuch. - 1. vyd. - Tübingen: GNV, 2007. - 200s. - ISBN 978 3 8233 6278 4. TROŠOK, Roman: Syntax = Eine Einführung in die Dependenzsyntax des Deutschen - 1. vyd. - Nitra: Vysoká škola pedagogická, 1995. - 125s. - ISBN 80-8050-007-X. ENGEL, Ulrich: Syntax der deutschen Gegenwartssprache. - 1. vyd. - Berlin: Erich Schmidt Verlag, 1994. - 315s. - ISBN 3 503 03094 8. ALTMANN, Hans, HAHNEMANN, Susan: Syntax fürs Exams. - 1. vyd. - Göttingen: Vandenhoeck & Ruprecht, 2007. - 226s. - ISBN 978 3 525 26500 0.	

<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 153					
A	B	C	D	E	FX
20.26	11.11	13.73	21.57	28.1	5.23
<b>Teacher:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc.					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/SZSdb/ NJ/15	<b>Name:</b> German Language and Literature
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> <b>Recommended extent of course ( in hours ):</b> <b>Per week: For the study period:</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 1., 2., 3., 4., 5., 6..	
<b>Level of study:</b> I.	
<b>Prerequisites:</b> KMF/FONdb/NJ/15 and KMF/MS1db/NJ/15 and KMF/ULGdb/NJ/15 and KMF/USLdb/NJ/15 and KNJ/LIT1db/NJ/15 and KMF/MS2db/NJ/15 and KMF/KUGdb/NJ/15 and KMF/LIT2db/NJ/15 and KNJ/MORdb/NJ/15 and KNJ/MS3db/NJ/15 and KMF/REAdb/NJ/15 and KMF/SYNdb/NJ/15 and KMF/LIT3db/NJ/15 and KMF/MS4db/NJ/15 and KMF/DLIdb/NJ/15 and KMF/ZDIdb/NJ/15 and KMF/UVPdb/NJ/15 and KMF/ZTRdb/NJ/15 and KMF/IKUdb/NJ/15 and KMF/LEXdb/NJ/15 and KMF/LKOdb/NJ/15	
<b>Conditions for passing the subject:</b> Students have to obtain the necessary number of credits from compulsory and elective subjects required by the study plan of the programme.	
<b>Results of education:</b>	
<b>Brief syllabus:</b> Sprachwissenschaft <ol style="list-style-type: none"> <li>1. Grundfragen und Zweige der Sprachwissenschaft; Richtungen der diachron und der synchron angelegten Sprachwissenschaft</li> <li>2. Klassifizierung der Sprachen nach dem diachronen Aspekt: Sprachfamilien; Klassifizierung der Sprachen nach dem synchronen Aspekt: Sprachtypen</li> <li>3. Die Sprache als Zeichensystem – Arten der Zeichen, Zeichenmodelle; die Sprache als Kommunikationsmittel – Kommunikationsmodelle</li> <li>4. Phonetik und Phonologie – Grundbegriffe (Sprechlaut, Phonem, Allophone); Teilgebiete der Phonetik; Phonemanalyse (Kommutationstest)</li> <li>5. Morphologie – Grundbegriffe (Morph, Morphem, Allomorphe); Arten der Morpheme; Wortarteneinteilungen im Deutschen</li> <li>6. Das Verb nach dem semantischen, nach dem morphologischen und nach dem syntaktischen Aspekt; die grammatischen Kategorien des Verbs (Person, Numerus, Tempus, Modus und Genus)</li> <li>7. Das Substantiv nach dem semantischen, nach dem morphologischen und nach dem syntaktischen Aspekt; die grammatischen Kategorien des Substantivs (Genus, Numerus und Kasus)</li> <li>8. Das Adjektiv nach dem semantischen, nach dem morphologischen und nach dem syntaktischen Aspekt; Deklination und Komparation</li> </ol>	

9. Die Pronomina, die Numeralia, die Adverbien, die Modalwörter, die Satzäquivalente und die Konjunktionen nach dem semantischen, nach dem morphologischen und nach dem syntaktischen Aspekt
10. Lexikologie – Aufbau und Veränderung des Wortschatzes; fachsprachliche Kommunikation
11. Syntax – Grundbegriffe (Satz-Definitionen); die traditionelle Satzanalyse (Satzglieder, Syntagmen); die Arten der Sätze nach ihrer Komplexität
12. Grundlagen der Übersetzungswissenschaft; interkulturelle Kommunikation

#### Literaturwissenschaft

1. Poetik (literarische Kunstarten und Gattungen, Grundzüge, Entwicklungstendenzen)
2. Die Aufklärung in der deutschen Literatur (Winckelmann, Gottsched, Bodmer und Breitinger)
3. Lessing und sein Konzept des bürgerlichen Trauerspiels
4. Versuch einer Reorientierung (Wieland, Klopstock, Haman, Sturm und Drang)
5. Deutsche Klassik (Weimarer Klassik)
6. Deutsche Romantik (Epochen, wichtigste Vertreter und Werke)
7. Im literarischen Niemandsland zwischen Klassik und Romantik (Hölderlin, Kleist, Jean Paul)
8. Das Zeitalter der Epigonen: Biedermeier und Vormärz
9. Realismus
10. Naturalismus
11. Impressionismus, Literatur des Fin de Siècle, Expressionismus
12. Deutsche Literatur aus Prag: Rilke und Kafka

#### Methodik und Didaktik

1. Die Grammatik-Übersetzungsmethode (GÜM)
2. Behavioristische Verfahren
3. Kognitivistische Verfahren
4. Konstruktivistische Verfahren
5. Kommunikative Didaktik und alternative Methoden
6. Grundbegriffe der DaF-Didaktik

#### **Literature:**

Indicated in the information sheets of the compulsory courses of the study programme.

#### **Language, knowledge of which is necessary to complete a course:**

German

#### **Notes:**

#### **Evaluation of subjects**

Total number of evaluated students: 20

A	B	C	D	E	FX
20.0	20.0	30.0	10.0	15.0	5.0

#### **Teacher:**

**Date of last update:** 17.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ ANA/15	<b>Name:</b> Preparation and Analysis of listen
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Students will attend 5-5 hours of each approbation at training schools. The course will be passed if 4 “listening reports” will be submitted (2 of each approbation).	
<b>Results of education:</b> During the pedagogical training students will follow and analyze the education process, they will learn how to apply the theoretical knowledge and the methodology of teaching and will become professionals being able to provide teaching, being able to record and analyze the realities of the profession. The students will be able for reflexion and self-reflexion.	
<b>Brief syllabus:</b> Attending training school classes. Preparing the attendance. Definition of goals, methodology and specialization. Self-observation of the educational process, didactical methods, pedagogical communication, psychological aspects. End phase of the attendance. Analyzation of records, reflexing and interpreting of the observed educational issues.	
<b>Literature:</b> Albert Sándor: Általános didaktika. Komárno : Selye János Egyetem, 2006. 226. ISBN 8089234070 Barabási Tünde: A tanítói tudás összetevői és fejlesztésük : Az elmélet és gyakorlat integrációja a magyarországi és romániai magyar tanítóképzési rendszerben. 1. vyd. Kolozsvár : Kolozsvári Egyetemi Kiadó, 2008. 151 s. ISBN 9789736107030 Cangelosi S. James. Strategie řízení třídy : Jak získat a udržet spolupráci žáků při výuce. 2. vyd. Praha : Portál, 1996. 300 s. ISBN 8071780839 Falus Iván: Didaktika. Budapest : Nemzeti Tankönyvkiadó, 2003. 552 s. ISBN 9631952967 Falus Iván. A tanárrá válás folyamata. - 1. vyd. - Budapest : Gondolat, 2007. - 245 s. - ISBN 978 963 9610 97 2 Falus Iván et all. A pedagógusok pedagógiája. - Budapest : Nemzeti Tankönyvkiadó, 2001. - 355 s. - ISBN 963191805x. Kalhous Zdeněk: Školní didaktika. 2. vyd. Praha : Portál, 2009. 448 s. ISBN 9788073675714 Kovátsné-Németh Mária. Fenntarthatóság, pedagógia, kutatás. - 1. vyd. - Győr : Nyugat-Magyarországi Egyetem Apáczai Csere János Kar, 2007. 227 s. ISBN 9789639364851	

Kosová Beata. Vysokoškolské vzdelávanie učiteľov : Vývoj, analýza, perspektívy. - 1. vyd. - Banská Bystrica : Pedagogická fakulta Univerzity Mateja Bela, 2012. 143 s. ISBN 9788055703534

Nagy József. Kompetencia alapú kritériumorientált PEDAGÓGIA. 1. vyd. Szeged : Mozaik Kiadó, 2007. 383 s. ISBN 978 963 697 541 8

Roeders Paul, Gefferth Éva. A hatékony tanulás titka : A hatékony tanítás és tanulás dinamikája. 1. vyd. : Trefort Kiadó, 2007. 215 s. ISBN 9789634464532

Petlák, Erich. Všeobecná didaktika. 1. vyd. : IRIS, 2004. 316 s. ISBN 8089018645

Pukánszky Béla. Iskola és pedagógusképzés. 1. vyd. Budapest : Gondolat Kiadó, 2014. 182 s. ISBN 9789636932282

Pasch Marvin, Gardner Trevor et all. Od vzdělávacího programu k vyučovací hodině : Jak pracovat s kurikulem. 1. vyd. Praha : Portál, s.r.o., 1998. 416 s. ISBN 8073670542

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 379

a	n
98.94	1.06

**Teacher:** Dr. habil. PaedDr. Kinga Horváth, PhD., PaedDr. Beáta Kiss

**Date of last update:** 18.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ APK/15	<b>Name:</b> Alternative pedagogical concepts
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> One written test during a term for 50 points, another 50 points could be earned for continuous in-class activities. At least 50 points – 50% of all possible points - has to be earned to pass the class. Evolution: A mark - 90-100%; B mark 80-89%; C mark 70-79%; D mark 60-69%, E mark 50-59%.	
<b>Results of education:</b> The student will learn the existing alternative trends and ways of development in the field of pedagogy both at home and abroad. This way the student will be able to identify, analyze, demonstrate and evaluate those alternative trends.	
<b>Brief syllabus:</b> The establishment of alternative schools in the first half of the 20th century (Waldorf, Jena-plan, Dalton, Freinet, Montessori). Establishment of alternative schools in the second half of the 20th century (client-centered approach according to Rogers, ITV, Zsolnay school, open school, project education, global education, etc.). Modeling of the reform school – outlining the prognosis.	
<b>Literature:</b> Németh András. A reformpedagógia múltja és jelene . - Budapest : Nemzeti Tankönyvkiadó, 2003. - 167 s. - ISBN 963 19 2190 5. Bodoni Ágnes. Reformpedagógia : Pedagógusi kompetenciák fejlesztése reform- és alternatív pedagógiai módszerek segítségével. - 1. vyd. - Kolozsvár : Ábel Kiadó, 2012. - 127 s. - ISBN 978-973-114-150-3. Németh András, Ehrenhard Skiera. Reformpedagógia és az iskola reformja. - 1. vyd. - Budapest : Nemzeti Tankönyv, 1999. - 345 s. - ISBN 963 19 0168 8. Németh András, Pirka Veronika. Az életreform és reformpedagógia-recepció és intézményesülési folyamatok a 20. század első felében. - 1. vyd. - Budapest : Gondolat Kiadó, 2013. - 409 s. - ISBN 978 963 693 471 2. Kovátsné-Németh Mária. Reformpedagógiai koncepciók, alternatív megoldások. - Komárno : Selye János Egyetem, 2007. - 330 s. - ISBN 9788089234349. Zelina Miron. Alternativne školstvo : alternativne školy, alternativna pedagogika, alternativne pedagogické koncepcie a smery. - 1. vyd. - Bratislava : IRIS, 2000. - 257 s. - ISBN 80-88778-98-0. Prucha Jan. Alternativní školy a inovace ve vzdělávání. Portál, 2004. - 144 s. - ISBN 8071789771. Pukánszky Béla. Iskola és pedagógusképzés. - 1. vyd. - Budapest : Gondolat Kiadó, 2014. - 182 s. - ISBN 9789636932282.	

Pukánszky Béla. Két évszázad gyermekei : A tizenkilencedik-huszedik század gyermekkorának története. - 1. vyd. - Budapest : Eötvös József Könyvkiadó, 2003. - 308 s. - ISBN 963 9316 65 2.

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak Language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 463

A	B	C	D	E	FX
53.78	25.49	15.12	4.97	0.65	0.0

**Teacher:** prof. Dr. Béla István Pukánszky, DSc.

**Date of last update:** 18.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ BDZ/15	<b>Name:</b> Biology child and school health
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%.	
<b>Results of education:</b> Students acquire basic knowledge about the human body - body composition, human ontogenesis, developmental specificities of organ systems and the basics of school hygiene.	
<b>Brief syllabus:</b> Morphological and functional characteristics of the human body and physical ontogeny of human, which is analyze from prenatal period to adulthood with an emphasis on teen age and young adulthood. Developmental specificities of the different organ systems. School hygiene.	
<b>Literature:</b> Dylevský, I.: Somatológia. Bratislava : OSVETA, 2000. - 439 s. - ISBN 80-8063-127-1 Feneis, H.: Anatomický obrazový slovník. Stuttgart : Georg Thieme Verlag, 1993. - 455s. - ISBN 80 7169 197 6 Mader, S. S.: Human biology. Wm. C. Brown Publishers, USA, Third edition 1992. 500 s. - ISBN 0-697-12333-2 McCracken, T.O.: Háromdimenziós anatómiai atlasz. Budapest : Scolar Kiadó, 2000. - 237 s. - ISBN 978-963-9193-99-4 Nagy, M.: Humánbiológia, Lilium Aurum, Dunaszerdahely, 2006, ISBN 80-8062-283-3. Netter, F. H.: Humán anatómiai atlasz. Budapest : Medicina Könyvkiadó, 2004. - 562 s. ISBN 963 242 848 X POSPÍŠIL, M.: Biológia člověka I. Přírodovědecká fakulta UK Praha, 1998, 340s. ISBN 80-223-1579-6 Szentágothai, J.: Funkcionális anatómia I.-III. Budapest : Medicina Könyvkiadó, 2006. - 710, 600, 800. - ISBN 963 242 565 0 Šmarda, J. a kol.: Biologie pro psychology a pedagogy. Portál, Praha, 2004.	
<b>Language, knowledge of which is necessary to complete a course:</b> Slovak or Hungarian	

<b>Notes:</b>					
<b>Evaluation of subjects</b>					
Total number of evaluated students: 386					
A	B	C	D	E	FX
10.88	14.51	24.09	19.69	25.39	5.44
<b>Teacher:</b> Dr. habil. PaedDr. Melinda Nagy, PhD., Dr. habil. Csaba Miklós Szinetár, CSc.					
<b>Date of last update:</b> 25.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ DID/15	<b>Name:</b> General didactics
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%.	
<b>Results of education:</b> The students will get the basic didactical knowledges – subject and methodological profile of the didactics, extension of knowledges, skills of teaching professionals, planning and organizational skills, controls and evaluation.	
<b>Brief syllabus:</b> Historical and current status of the didactics.Modernization of the teaching material.Basic educational documents.Processual aspects of the education.Principles of the education.Tools for education.Categories of the education goals.Teaching styles.Differentiation.Evaluation.Theoretical models and conceptual forms of education and evaluation.	
<b>Literature:</b> Albert Sándor. Általános didaktika. - Komárno : Selye János Egyetem, 2006. - 226. - ISBN 80-89234-07-0. Albert Sándor. Didaktika. Liliium Aurum, 2005. - 250 s. - ISBN 8080622523. Falus Iván. Didaktika. - Budapest : Nemzeti Tankönyvkiadó, 2003. - 552 s. - ISBN 9631952967. Nagy Sándor. Didaktika. - Budapest : Tankönyvkiadó, 1969. - 239 s. - ISBN 0012790. Kalhous Zdeněk. Školní didaktika. - 2. vyd. - Praha : Portál, 2009. - 448 s. - ISBN 978-80-7367-571-4. Petlák, Erich. Všeobecná didaktika. - 1. vyd. : IRIS, 2004. - 316 s. - ISBN 80-89018-64-5. Komenský Ján Ámos. Výber myšlienok z diela Veľká didaktika. - Prešov : Metodické centrum Prešov, 1992. - 23 s. - ISBN 8085410273. Barabási Tünde. A tanítói tudás összetevői és fejlesztésük : Az elmélet és gyakorlat integrációja a magyarországi és romániai magyar tanítóképzési rendszerben. - 1. vyd. - Kolozsvár : Kolozsvári Egyetemi Kiadó, 2008. - 151 s. - ISBN 978-973-610-703-0. Nagy József. Kompetencia alapú kritériumorientált PEDAGÓGIA. - 1. vyd. - Szeged : Mozaik Kiadó, 2007. - 383 s. - ISBN 978 963 697 541 8.	

Falus Iván et all. A pedagógusok pedagógiája. - Budapest : Nemzeti Tankönyvkiadó, 2001. - 355 s. - ISBN 963191805x.

Falus Iván. A tanárrá válás folyamata. - 1. vyd. - Budapest : Gondolat, 2007. - 245 s. - ISBN 978 963 9610 97 2.

Kovátsné-Németh Mária. Fenntarthatóság, pedagógia, kutatás. - 1. vyd. - Győr : Nyugat-Magyarországi Egyetem Apáczai Csere János Kar, 2007. - 227 s. - ISBN 978-963-9364-85-1.

Roeders Paul, Gefferth Éva. A hatékony tanulás titka : A hatékony tanítás és tanulás dinamikája. - 1. vyd. : Trefort Kiadó, 2007. - 215 s. - ISBN 978-963-446-453-2.

Kosová Beata. Vysokoškolské vzdelávanie učiteľov : Vývoj, analýza, perspektívy. - 1. vyd. - Banská Bystrica : Pedagogická fakulta Univerzity Mateja Bela, 2012. - 143 s. - ISBN 978-80-557-0353-4.

Cangelosi S. James. Strategie řízení třídy : Jak získat a udržet spolupráci žáků při výuce. - 2. vyd. - Praha : Portál, 1996. - 300 s. - ISBN 80-7178-083-9.

Pasch Marvin, Gardner Trevor et all. Od vzdělávacího programu k vyučovací hodině : Jak pracovat s kurikulem. - 1. vyd. - Praha : Portál, s.r.o., 1998. - 416 s. - ISBN 80-7367-054-2.

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak Language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 1218

A	B	C	D	E	FX
11.66	17.57	16.58	17.32	26.35	10.51

**Teacher:** prof. Dr. Péter Tóth, PhD., Dr. habil. PaedDr. Kinga Horváth, PhD.

**Date of last update:** 25.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ FVV/15	<b>Name:</b> Philosophy of Education
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> One written test during a term for 50 points, another 50 points could be earned for continuous in-class activities. At least 50 points – 50% of all possible points - has to be earned to pass the class. A mark - 90-100%; B mark 80-89%; C mark 70-79%; D mark 60-69%, E mark 50-59%.	
<b>Results of education:</b> The student will receive an overview of the scope of current educational theory. The student will be introduced to current problems, up-to-date theories and knowledge, so that the student will be able to recognize the theoretical concepts in the educational practice. The student will be able to find the relationship between the trends of educational philosophy, education theories and education concepts. The student will be able to evaluate the educational phenomena based on philosophy, ideology and theory.	
<b>Brief syllabus:</b> Basics of perennialism, essentialism, progressivism, behaviorism, and existentialism. Educational theory: intellectual, personalistic, social, academic, cognitive-psychological, social-cognitive, technological. Educational concepts: problem solving, project education, cooperative education, mastery learning, global education, and constructivism.	
<b>Literature:</b> Angelusz Erzsébet. Filozófia, antropológia, nevelés. - Budapest : Akadémiai Kiadó, 1984. - 104 s. - ISBN 963 05 3404 5. Halasy-Nagy József. A filozófia. - Budapest : Pantheon Kiadás, 1991. - 408 s. - ISBN 963 05 5929 3. Mészáros András. A felső-magyarországi iskolai filozófia lexikona. - Pozsony : Kalligram, 2003. - 288 s. - ISBN 8071495409. Pukánszky Béla. Iskola és pedagógusképzés. - 1. vyd. - Budapest : Gondolat Kiadó, 2014. - 182 s. - ISBN 9789636932282. Pukánszky Béla. A gyermekkor története. - 1. vyd. - Budapest : Műszaki Könyvkiadó, 2001. - 201s. - ISBN 963 16 2782 9. Pukánszky Béla. Két évszázad gyermekei : A tizenkilencedik-huszedik század gyermekkorának története. - 1. vyd. - Budapest : Eötvös József Könyvkiadó, 2003. - 308 s. - ISBN 963 9316 65 2.	

Pukánszky Béla. Pedagógiai eszmetörténet. - 1. vyd. - Budapest : Gondolat Kiadó, 2013. - 168 s. - ISBN 978-963-693-228-2.  
Vajda Zsuzsanna, Kósa Éva. Neveléslélektan. - 1. vyd. - Budapest : Osiris Kiadó, 2005. - 564 s. - ISBN 963 389 728 9. - ISSN 1218-9855.

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 1009

A	B	C	D	E	FX
27.75	28.84	27.45	11.79	3.77	0.4

**Teacher:** Gyöngyi Gál, PhD.

**Date of last update:** 25.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ IKT/15	<b>Name:</b> ICT-based
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Making 2 projects during the semester , each for 25 points and the final presentation of the methodology of a selected lesson for 50 points. In order to pass the course the student needs to collect at least 50% of the maximum points. The scale of evaluation is the following: A – 90 -100%, B – 80 -89%, C – 70 -79%, D – 60 - 69%, E – 50 -59%.	
<b>Results of education:</b> By the completion of the course, students deepen their professional competence in the field of information and communication technologies. They will be able to locate, evaluate and use information so that they become autonomous, independent and lifelong learners. They will have the ability to locate, evaluate, use and communicate information in all their various forms, such as the integration of books, computer, the media and technology, ethics, critical thinking, information and communication skills.	
<b>Brief syllabus:</b> <ul style="list-style-type: none"> <li>• Basic concepts of work with computers (objects, files, types, maps, addresses )</li> <li>• Basics of Word (copy protection, basic items, formatting)</li> <li>• Working with pictures, WordArt, ClipArt - special text effects</li> <li>• Basics of graphical environment Paint (copy protection, basic controls)</li> <li>• Introduction to digital technology, principles of operation, working with the media</li> <li>• the use of digital and multimedia devices in the educational process</li> <li>• Creating lessons from selected objects, integrated learning, practical use of certain information for the preparation of materials in teaching.</li> <li>• The Internet - Definitions</li> <li>• Browser, criteria for finding, downloading images and texts from the Internet</li> <li>• E-mail: e-mail, creating your own e-mail addresses, basic work, connecting documents</li> </ul>	
<b>Literature:</b> Baka Magdolna, Koczka Ferenc: Informatika - szövegszerkesztés, Eger : EKTF Líceum Kiadó, 1997. 170 s. Csórián Sándor: Információ és kommunikáció. Budapest : Kossuth Könyvkiadó, 2003. 119. ISBN 9630944103 Czifra Juraj at all.: Informačné a komunikačné technológie v praxi I. Komárno : Selye János Egyetem, 2007. 450 s. ISBN 9788089234417 Szököl István: Modulárny systém výučby informatiky. Komárno : UJS, 2010. 100s. ISBN 9788089234974	

Stoffa Veronika: Az informatika alapjai I. Apáczai közalapítvány, 2007. 268 s. ISBN 9788089234295  
Wyatt L. Allen: Az internet alapjai. Budapest : Kossuth Könyvkiadó, 1996. 352. ISBN 9630938383x

**Language, knowledge of which is necessary to complete a course:**  
Hungarian or Slovak Language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 523

A	B	C	D	E	FX
54.88	20.84	12.43	5.16	2.87	3.82

**Teacher:** Mgr. Dávid Paksi

**Date of last update:** 25.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ INV/15	<b>Name:</b> intercultural education
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> One written test during a term for 50 points, another 50 points could be earned for continuous in-class activities (presentation of casuistics). At least 50 points – 50% of all possible points - has to be earned to pass the class. A mark - 90-100%; B mark 80-89%; C mark 70-79%; D mark 60-69%, E mark 50-59%.	
<b>Results of education:</b> By completing the course students will gain knowledge on the essence of ethnic processes and ethnic minorities of Slovakia, furthermore gain skills in practical applying acquired theories in educational process.	
<b>Brief syllabus:</b> Basic terminology: ethnicity, nation, nationality, ethnic minorities, multiculturalism, inter-cultural competence, etc. Inter-ethnic and inter-cultural relations. Ethnic symbols, stereotypes. Ethnic history of Slovakia. History of ethnic minorities in Slovakia, with particular regard to Hungarians. Concrete examples on Hungarian-Slovak, Hungarian-German, Hungarian-Rusin inter-ethnic relationships. The problem of the Rome minority in Slovakia and Central Europe. Practical opportunities of evolving inter-cultural competencies (meeting other cultures, respecting otherness, tolerance).	
<b>Literature:</b> Ács Zoltán: Nemzetiségek a történelmi Magyarországon. Budapest: Kossuth Könyvkiadó 1986. Botík, Ján: Chorváti na Slovensku. Bratislava: Slovenské národné múzeum 1996. Forray R. Katalin szerk.: Ismeretek a romológia alapképzési szakhoz. Pécs: Pécsi Tudományegyetem 2006. <a href="http://mek.oszk.hu/04800/04867/04867.pdf">http://mek.oszk.hu/04800/04867/04867.pdf</a> Gabal, Ivan: Etnické menšiny ve střední Evropě. Praha 1999. Gallová Kriglerová, Eva–Kadlečíková, Jana–Lajčáková Jarmila: Migranti. Multikulturalizmus a kultúrna integrácia migrantov na Slovensku. Nový pohľad na staré problémy. Bratislava: CVEK 2009. Gecse Annabella: Az etnikai és társadalmi átrendeződés folyamata egy gömöri falu 20. századi életében. Komárom–Somorja: Fórum Kisebbségkutató Intézet 2007 /Interethnica10./ Gyurgyík László: A szlovákiai magyarság népesedési folyamatai a 20. században (1918-tól 2001-ig). Komárom: Selye János Egyetem Tanárképző Kara 2013 / Monographiae Comaromienses 10./ Horváthová, Margaréta: Nemci na Slovensku. Etnokultúrne tradície z aspektu osídlenia, remesiel a odievania. Komárno–Dunajská Streda: Fórum inštitút–	

Spoločenskovedný ústav–Vydavateľstvo Lilium Aurum 2002 /Interethnica 4./ L. Juhász Ilona: „Fába róva, földbe ütve...” A kopjafák/emlékoszlopok mint a szimbolikus térfoglalás eszközei a szlovákiai magyaroknál. Komárom–Dunaszerdahely: Fórum Kisebbségkutató Intézet–Lilium Aurum Könyvkiadó 2005 /Interethnica 8./ Kiss Gabriella: Multikulturalizmus és oktatás. Debrecen: Kossuth Egyetemi Kiadó 2001. Liszka József: Bevezetés a néprajzba. A magyar néprajz/ európai etnológia alapjai. Dunaszerdahely: Lilium Aurum 2006. Liszka József szerk.: Interetnikus és interkulturális kapcsolatok Dél-Szlovákiában. Komárom: Selye János Egyetem Tanárképző Kara 2009 /Monographiae Comaromienses 1./ Liszka József: Populáris kultúra. Somorja: Fórum Kisebbségkutató Intézet 2010 /Magyarok Szlovákiában 6./ Magyar néprajzi lexikon 1–5. Budapest: Akadémiai Kiadó 1977–1982. Paládi-Kovács Attila szerk.: A nemzetiségek néprajzi felfedezői. Budapest: Akadémiai Kiadó 2006. Sopoliga, Miroslav: Ukrajinci na Slovensku. Etnokultúrne tradície z aspektu osídlenia, ľudovej architektúry a bývania. Komárno–Dunajská Streda: Fórum inštitút – Spoločenskovedný ústav–Vydavateľstvo Lilium Aurum 2002 /Interethnica 2./ Tradičná ľudová kultúra Slovenska slovom a obrazom. Elektronická encyklopédia (<http://www.ludovakultura.sk/index.php?id=11>) Vajda Barnabás szerk.: Államhatár és identitás–Komárom/Komárno. Komárom: Selye János Egyetem Tanárképző Kara 2011 /Monographiae Comaromienses 3./ Varjú Katalin: „Pénteken délig nyitva van az ég!” Somorja–Dunaszerdahely: Fórum Kisebbségkutató Intézet–Lilium Aurum Könyvkiadó 2003 / Interethnica 6.

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak Language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 404

A	B	C	D	E	FX
56.44	19.55	12.87	5.2	4.95	0.99

**Teacher:** PaedDr. Terézia Strédl, PhD., István Jobbágy, PhD.

**Date of last update:** 25.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ IPE/15	<b>Name:</b> Inclusive Education
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> One written test during a term for 50 points, another 50 points could be earned for continuous in-class activities (presentation of casuistics). At least 50 points – 50% of all possible points - has to be earned to pass the class. A mark - 90-100%; B mark 80-89%; C mark 70-79%; D mark 60-69%, E mark 50-59%.	
<b>Results of education:</b> Students becomes competent in differential approach to students/pupils with special education needs and also will be able to identify the problems and difficulties of learning. The students will get wide and complex picture about the work of special education teacher and school psychologist, about stimulation programmes, therapies and about the supportive care generally.	
<b>Brief syllabus:</b> Mission of the special education – education of disabled pupils. Sensory disabled children and their education. Physically disabled children and their education. Disabilities on communication skills of children. Emotionally disturbed children and possibilities within their education. Segregation, integration and inclusion of disabled children. Special education system and special education consultation services. Therapies, corrections and reeducation as tools for the optimisation of the education process for children with specific developmental learning disorders.	
<b>Literature:</b> Gordosné Szabó Anna: Bevezetés a gyógypedagógiába. 7. vyd. Budapest : Nemzeti Tankönyvkiadó. 2000. 116 s. Gordosné Szabó Anna: Gyógyító pedagógia = Nevelés és terápia. 1. vyd. Budapest : Medicina Könyvkiadó, 2004. 587 s. ISBN 963 242 757 2 Illyés Gyuláné: Gyógypedagógiai pszichológia. Budapest : Akadémiai Kiadó, 1971. 465 s. ISBN 0007635 Illyés Gyuláné: Speciálnopedagógická psychológia. 1. vyd. Bratislava : Slovenské Pedagogické Nakladateľstvo. 1978. 589 s. Mesterházi Zsuzsa: A nehezen tanuló gyermekek iskolai nevelése. 1. vyd. : Eötvös Lóránd Tudományegyetem Bárczi Gusztáv Gyógypedagógiai Kar. 1998. 348 s. ISBN 9637151126 Strédl Terézia: Inkluzív pedagógia avagy a gyógypedagógiáról másképp. 1. vyd. Komárno : Univerzita J. Selyeho. 2013. 148 s. ISBN 9788081220890 Vašek Štefan: Pedagogika viacnásobne postihnutých. 1. vyd. Bratislava : Sapiencia. 1999. 296 s. ISBN	

8096718045 Vašek Štefan: Špeciálno pedagogická diagnostika. 4. vyd. : Sapiaientia s.r.o, 2004.168 s. ISBN 8096911201					
<b>Language, knowledge of which is necessary to complete a course:</b> Hungarian or Slovak Language					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 1002					
A	B	C	D	E	FX
28.74	24.35	28.74	13.37	3.89	0.9
<b>Teacher:</b> Mgr. Anita Tóth-Bakos, PhD.					
<b>Date of last update:</b> 25.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					



## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ LAD/15	<b>Name:</b> School legislation and documentation
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%	
<b>Results of education:</b> The student will be able to: - list the number, title and content of legislation concerning education - describe public education management and its level, - list concretely the changes in public education and interpret their legal consequences, - understand the typical features of public education management and local government principles in education and school management, - analyze the relevant regulations, -classify and categorize the relevant regulations, - evaluate the processes in public education.	
<b>Brief syllabus:</b> Sorting legal and pedagogical and organizational standards applied in education. The interpretation of the Constitution in terms of its application in the field of education, training and education. Government Program Declaration after November 1989 in terms of orientation to school education. Projects oriented towards the long term development of education in Slovakia. The process of creating laws, decrees and other documents forming the legal framework of education and training. The Education Act and the ensuing regulations. Act on school facilities and the subsequent regulations. The Higher Education Act and the ensuing regulations. Legal solutions of the qualification and further training of teaching staff of schools. The questions of managing "non-state" schools and school facilities.	
<b>Literature:</b> The Constitution of the Slovak Republic 245/2008 The Law on Upbringing and Education (School Act) and on amendments to certain laws Other relevant laws and regulations.	
<b>Language, knowledge of which is necessary to complete a course:</b> Hungarian or Slovak Language	
<b>Notes:</b>	
<b>Evaluation of subjects</b>	

Total number of evaluated students: 808					
A	B	C	D	E	FX
61.26	22.03	9.53	3.84	3.34	0.0
<b>Teacher:</b>					
<b>Date of last update:</b> 25.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ MPE/15	<b>Name:</b> Multimedia education
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Making a presentation, wheres lides showing thetheoretical and practicalskills of multimediaeducation, (max 50. points) A – 90 -100%, B – 80 -89%, C – 70 -79%, D – 60 - 69%, E – 50 -59%.	
<b>Results of education:</b> The student learns touse themultimedia mediamethods, tools in the pedagogicalpractice. And also developcriticalthinking skills and information.	
<b>Brief syllabus:</b> Communicationtypes, forms. background of stereotypes and conventionsof mobile screencontents. Photo texts, basiccodes, text writing and readingon mobile. The socialfunction of themedia. The mediacategorization. The mediausagehabits, modes, language. The theoretical and practicalknowledge of mediapedagogy. International practice. The computer basedlearning. Electronicmedia, video, and computer use. Criticalthinking: themassmedia and communication, manipulation, informationsociety. The analysis of themutimediainteractions.	
<b>Literature:</b> Komenczi Bertalan: Információ és társadalom. Eger : EKF Liceum. 2002. 200 s. ISBN 0269771 Karvalics Z. László: Neumann Jánostól az Internetig. Budapest : Napvilág, 1999. 140. ISBN 9639082228 Z. Karvalics L.: Információs társadalom (a technikától az emberig). Műegyetemi Kiadó BME TTK Budapest. 1995 Stoffová Veronika: Education for information and knowledge based society = Vzdelávanie pre informačnú a vedomostnú spoločnosť. Brno : Univerzita J. Selyeho Komárno, 2012. 245 s. ISBN 9788081220647 Stoffová Veronika: Počítač univerzálny didaktický prostriedok. Nitra, 2004. 173Ss. ISBN 80 8050 765 1 Tapscott Don: Digitális gyermekkor. Budapest : Kossuth Könyvkiadó, 2001. 383 s. ISBN 9630943042 Zrinszky László: Tájékozódás, tanulás, tudás. Budapest : Usiris Könyvkiadó, 2007. 240 s. ISBN 978 963 9706 14 9 MEDIÁLNI PEDAGOGIKA V TEORII A PRAXI - Asociace pro ...	

[www.medialnipedagogika.cz/.../Schorb-Sloboda\\_Teorie-med-ped\\_in\\_Medialni-pedagogika-v-teorii-a-praxi.pdf](http://www.medialnipedagogika.cz/.../Schorb-Sloboda_Teorie-med-ped_in_Medialni-pedagogika-v-teorii-a-praxi.pdf)

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak Language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 358

A	B	C	D	E	FX
44.97	25.7	14.53	6.7	5.87	2.23

**Teacher:** Mgr. Nikolas Katona

**Date of last update:** 18.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ PKO/15	<b>Name:</b> Educational communication
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Evolution: A – 90 -100%, B – 80 -89%, C – 70 -79%, D – 60 - 69%, E – 50 -59%.	
<b>Results of education:</b> Student will get theoretical and practical basic skills within the social and pedagogical communication. During practices student will learn verbal and non-verbal skills used within the social communication, will train standard pedagogical situations - such as introduction of a new student, praise of a student, communication with parents. Student will be able to use non-verbal and paralinguistic means of expressions within these situations. Student will be able to analyze the school classes according to aspects of pedagogical communication.	
<b>Brief syllabus:</b> Introduction to communication. Definition of communication, social communication and terms. People and communication. Individual communication skills. Verbal communication. Words and their interpretation. Paralinguistic means of expression. Practicing of verbal skills. Non-verbal communication. Means of expression of non-verbal communication. Emphatic and assertive communication, behaviour and its importance in the communication. Basic characteristics of pedagogical communication. Educational goals and pedagogical communication. Organisational forms and didactical methods in accordance with communication. Main characteristics of teacher's communication. Monological and dialogical communication forms. Verbal behaviour of students. Cooperation between teachers and students. How does the teacher motivate? The question of the teacher. Teacher's instructions. Evaluation. Teacher's explanation. Solving of educational conflicts. Regulation of student's communication. Non-verbal communication during the class. Paralinguistic communication. Body-communication in education. Communication barriers. Expression of expectations.	
<b>Literature:</b> Buda Béla. A közvetlen emberi kommunikáció szabályszerűségei. Budapest : Tömegkommunikációs Kutatóközpont, 1988. 296 s. ISBN 963 333 043 2 Gavora Peter. Akí sú moji žiaci? . 3. vyd. Nitra : Enigma, 2011. 222 s. ISBN 9788089132911 Nelešovská Alena. Pedagogická komunikace v teorii a praxi. 1. vyd. : Grada, 2005. 175s. ISBN 8024707381	

Mareš Jiří. Sociální a pedagogická komunikace ve škole. 1. vyd. Praha : Statní Pedagogické Nakladatelství, 1989. 165s. ISBN 8004218547  
Strédl Terézia. Kommunikáció és konfliktuskezelés. 1. vyd. Révkomárom : Szakképző és Felnőttképzési Intézet, 2009. 71 s. ISBN 9788097001124

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak Language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 1101

A	B	C	D	E	FX
66.39	14.26	11.44	4.27	2.72	0.91

**Teacher:**

**Date of last update:** 25.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KPD/SZdb/ PRV/15/15		<b>Name:</b> Project Education			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present					
<b>Number of credits:</b> 1					
<b>Recommended semester/trimester of study:</b> 5.					
<b>Level of study:</b> I.					
<b>Prerequisites:</b>					
<b>Conditions for passing the subject:</b>					
<b>Results of education:</b>					
<b>Brief syllabus:</b>					
<b>Literature:</b>					
<b>Language, knowledge of which is necessary to complete a course:</b>					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 82					
A	B	C	D	E	FX
56.1	21.95	13.41	2.44	3.66	2.44
<b>Teacher:</b> Dr. habil. Erika Kopp, PhD.					
<b>Date of last update:</b> 18.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ PX1/15	<b>Name:</b> Teaching Practice
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week: For the study period:</b> 20s <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Graduate student observation in the range of 5-5 hours of both the scope of its subject specialization. Student completes a training school lectures and practical presentation of the school's documentation or school facility. Student in due time the Head of teaching experience shall submit report on teaching practice.	
<b>Results of education:</b> Students gain knowledge in the following topics: papers school or school facility, pedagogical documentation and school facilities, teaching methods, curricula, course teaching outline lesson and preparation for the lesson, the possibility of active work with pupils, criteria, methods and forms of assessment	
<b>Brief syllabus:</b> Observation in the range of 5-5 hours from both objects subject specialization. A training school conducted lectures and practical presentation of the school's documentation or school facility. National and school educational program. Class book and record classification	
<b>Literature:</b> ISCED2 ISCED3	
<b>Language, knowledge of which is necessary to complete a course:</b> Hungarian or Slovak Language	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 491	
a	n
98.37	1.63
<b>Teacher:</b> PaedDr. Tamás Török, PhD., prof. Dr. Béla István Pukánszky, DSc.	
<b>Date of last update:</b> 18.05.2023	



**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ ROR/15	<b>Name:</b> gender equality
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%.	
<b>Results of education:</b> The student will learn about the concept of social gender in social, psychological, and biological context. The student will be able to identify gender prejudice in education and develop preventive methods for women and men (girls and boys). The student will be able to recognize the stereotype system within the education, and its negative effects. The student will be able to apply the necessary methodology for ensuring social gender identity in the school environment.	
<b>Brief syllabus:</b> Gender - gender studies - definition: gender, sex, gender stereotypes, gender sensitization in education, both direct and indirect discrimination, emancipation, feminism. The social position of women. The principles of gender. The cultural and subcultural interpretation of social gender. The society and its role in social gender equality. Education and self-education. Equal opportunities. Education according to the social gender perspective - gender socialization theory, feminine pedagogy, sensitizing education of social gender. The gender in the education process. Inequalities in school. The gender aspects of family education. The role and potential of gender communication.	
<b>Literature:</b> Bútorová Zora: Násilie páchané na ženách ako problém verejnej politiky. Bratislava : IVO Inštitút pre verejné otázky, 2005. 132 s. ISBN 80 88935 78 4 Bútorová Zora: She and He in Slovakia Gender and Age in the Period of Transition. Bratislava : Institute for Public Affairs, 2008. 342 s. ISBN 978808934514 Pukánszky Béla: A nőnevelés története. 1. vyd. Budapest : Gondolat Kiadó, 2013. 228 s. ISBN 9789636932220 Pukánszky Béla: A gyermekkor története. 1. vyd. Budapest : Műszaki Könyvkiadó, 2001. 201s. ISBN 963 16 2782 9 Pukánszky Béla: Két évszázad gyermekei : A tizenkilencedik-huszedik század gyermekkorának története. 1. vyd. Budapest : Eötvös József Könyvkiadó, 2003. 308 s. ISBN 963 9316 65 2	

Vajda Zsuzsanna, Kósa Éva. Neveléslélektan. Budapest : Osiris Kiadó, 2005. 564 s. ISBN 963 389 728 9

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak Language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 357

A	B	C	D	E	FX
52.94	24.93	14.29	5.32	2.52	0.0

**Teacher:** prof. Dr. Béla István Pukánszky, DSc.

**Date of last update:** 25.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ SCP/15	<b>Name:</b> social psychology
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 3.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%.	
<b>Results of education:</b> Students will get wide picture about the interdisciplinary psychological trends being in context with the educational process such as group cohesion, attribution theory and sociometry, both on theoretical and practical level.	
<b>Brief syllabus:</b> Psychology and its interdisciplinarity in coherence with the science of nowadays. Social psychology and its 4 fields: individual, social relationships, group, crowd – characteristics, attributes. Social perception, social communication, social groups. Moreno and the sociometry. School environment and aspects of optimisation.	
<b>Literature:</b> Aronson Elliot: A társas lény. 1. vyd. Budapest : Akadémiai Kiadó, 2011. 504 s. ISBN 978 963 05 86283 Aronson Elliot: Columbine után : Az iskolai erőszak szociálpszichológiája. 1. vyd. Budapest : Ab Ovo Kiadó. 2009. 191 s. ISBN 978-963-9378-72-8. Boroš Július: Základy sociálnej psychológie : (pre študujúcich humánne, sociálne a ekonomické vedy) 1. vyd. : IRIS, 2001. 227 s. ISBN 8089018203 Csepeli György: A meghatározatlan állat : Szociálpszichológia kezdőknek és haladóknak. 1. vyd. Budapest : Jászöveg Műhely Kiadó, 2005. 324 s. ISBN 963 7052 25 9 Csepeli György: A szociálpszichológia vázlata. Budapest : Jászöveg Műhely Könyvkiadó. 2001. 160 s. ISBN 963 048 678 4 Goleman Daniel: Társas intelligencia = Az emberi kapcsolatok új tudománya. 3. vyd. Budapest. 506 s. ISBN 9789633100349	
<b>Language, knowledge of which is necessary to complete a course:</b> Hungarian or Slovak Language	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 1158	

A	B	C	D	E	FX
17.01	21.07	29.02	20.38	12.26	0.26
<b>Teacher:</b> PaedDr. Terézia Strédl, PhD.					
<b>Date of last update:</b> 25.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ SKS/15	<b>Name:</b> School - school climate
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Duringthesemester, studentswillproduce a five-pagestudyorprepare a sociogramduringthepedagogicalpractice, whichpresents the front of thegroup. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%	
<b>Results of education:</b> The students learnmapping the factors of school-enviromental factors, learn about satisfactionmeasurement, cooperation. The student infored about the socialenvironment of micro- and meso-schoolenviroment.	
<b>Brief syllabus:</b> Socialization and socialspace. International trends and socialsegments. The schoolworld:theoreticaltrendsinschool, historicaloverview, types, schoolclimate, system and structure. Manifest, latent, fulfilled andunfulfilledneeds. Hidden curriculum. The microenvironment of school. Locality and theschool. Extra-curriculartasks. Leisureas a thirdsocializingarea. The school'smacroenvironment. Family, teachers, professionals, society - trends and tendencies. Schoolint he postmodern / IT world.	
<b>Literature:</b> Csoma Gyula: Elviszik-e a kutyák az iskolát? Móra Könyvkiadó : Budapest. 1983 Hvozdík Stanislav: Vybrané kapitoly zo školskej psychológie I. Prešov : FF P. Katedra psychológie. 1999. 402 s. ISBN 80 88922 038 Gajdošová Eva: Školský psychológ = a jeho vstup do humanizácie našich škôl. 1. vyd. Bratislava : PRÍRODA a. s. 1998. 190 s. ISBN 80 0701029 7 Nagy Ádám: Családon és iskolán túl - a harmadlagos szocializációs közeg és az ifjúságügy mint önálló terület elméleti alapjai. Excenter füzetek 3. Budapest : Excenter Kutatóközpont. 2010(www.excenter.eu., www.iufjúságügy.hu) Nagy Ádám: Ifjúságügy - ifjúsági munka és az ifjúság. Excenter füzetek 5. Budapest : Excenter Kutatóközpont. 2010. www.excenter.eu Székely Levente: Virtuális ifjúsági munka és az e-ifjúság. Excenter füzetek 5. Budapest : Excenter Kutatóközpont. 2010. www.excenter.eu Trencsényi László: Hetedik nekifutás az értékek útvesztőjében. Budapesti Nevelő. 2009/2. <a href="http://preview.fppti.hu/data/cms54391/2009.2.szam.teljes%29.pdf">http:// preview.fppti.hu/data/cms54391/2009.2.szam.teljes%29.pdf</a>	

Turek, Ivan: Moderné trendy vo výučbe na vysokých školách. 1. vyd. Komárno : Univerzita J. Selyeho. 2006. 496s. ISBN 80 89234135  
Zelina Miron: Stratégie a metódy rozvoja osobnosti. Bratislava : Iris, 1994. 162s. ISBN 80 96701347

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak Language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 314

A	B	C	D	E	FX
28.66	22.93	14.65	9.87	19.11	4.78

**Teacher:** Dr. habil. Erika Kopp, PhD.

**Date of last update:** 25.06.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ SMP/15	<b>Name:</b> School management and school policy
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test – 100 points. Students can solve exercise given b the teacher during the class. Each exercise can earn 5 points (up to 230 points in total). If the amount of earned points is acceptable for the student to pass the class, there is no need to attend the final test. The class could be pass after earning of at least 50 % of earnable points. The marking is A – 90 -100%, B – 80 -89%, C – 70 -79%, D – 60 - 69%, E – 50 -59%	
<b>Results of education:</b> Student will get knowledges about the complex operational school management, quality management, managing styles, school marketing and will be able to apply these knowledges in practical way in accordance with the Slovak legislation.	
<b>Brief syllabus:</b> The functions of the school. The essence of school management in a democratic society. Adaptability of the school management system. The roles of the government and governmental institutions at school management. The main purposes of the school management. Concepts and management theories. School management. School managing models and its specialties. The basic management roles. Educational programs as a part of the school management. Internal rules leading to optimal operation. Managing styles. The personality and communicational skills of the manager. School marketing end the current needs of schools. The climate and culture within the school in case of producing nd applying of educational programs.	
<b>Literature:</b> Halász Gábor. A közoktatási rendszerek irányítása. Okker kiadó. 94 s. - ISBN 0009672. Halász Gábor. Az oktatás az Európai Únióban = Tanulás és együttműködés. - 1. vyd. - Budapest : Új Mandátum Könyvkiadó, 2012. - 376 s. - ISBN 978 963 287 053 3. Halász Gábor. Az oktatás minősége és az önkormányzati oktatásirányítás : Okker kiadó, 1996. - 364 s. - ISBN 9637315403. Halász Gábor. Az oktatási rendszer. - 1. vyd. - Budapest : Műszaki Könyvkiadó, 2001. - 215s. - ISBN 963-16-2769-1. Horváthová, Kinga, Manniová, Jolana. Úvod do školského manažmentu. - 1. vyd. - Ivanka pri Dunaji : AXIMA, 2008. - 179 s. - ISBN 978 80 969178 6 0.	



Školský manažment v nových spoločenských podmienkach (pre riadiacich pedagogických zamestnancov) = Zborník z medzinárodnej vedeckej konferencie / Kinga Horváthová. - 1. vyd. - Bratislava : Katedra pedagogiky Pedagogickej fakulty UK v Bratislave, 2008. - 182 s. - ISBN 978-80-969178-8-4.

Horváthová, Kinga. Kontrola a hodnotenie v školskom manažmente. - 1. vyd. - Bratislava : Wolters Kluwer, 2010. - 106 s. - ISBN 978-80-8078-329-7.

Albert Sándor. Iskolavezetés. - 1. vyd. - Selye János Egyetem : Komárom, 2007. - 82 s. - ISBN 978-80-89234-27-1.

Albert Sándor. Minőségfejlesztés az iskolában. - Komárno : Selye János Egyetem, 2006. - 130. - ISBN 8089234127.

Albert Sándor. Önértékelés és minőségbiztosítás az iskolában. - 1. vyd. - Pécs : Comenius Kft., 2009. - 108 s. - ISBN 978 963 9687 26 4.

Kosová Beata. Transformačné premeny Slovenského školstva po roku 1989. - 1. vyd. - Banská Bystrica : Pedagogická fakulta Univerzity Mateja Bela, 2011. - 168 s. - ISBN 978-80-557-0275-9.

**Language, knowledge of which is necessary to complete a course:**

Hungarian or Slovak Language

**Notes:**

**Evaluation of subjects**

Total number of evaluated students: 635

A	B	C	D	E	FX
21.26	14.33	17.32	20.94	24.57	1.57

**Teacher:** Dr. habil. PaedDr. Kinga Horváth, PhD.

**Date of last update:** 18.05.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ SPP/15	<b>Name:</b> School prevention programs
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 4.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test end PPT. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%	
<b>Results of education:</b> The students acquire competences about prevention, healthdevelopment in schoolenvironment. The topic of the presentationis the prevention for students. Atthe end of the semester there is a testassessesacquired knowledge.	
<b>Brief syllabus:</b> Defininghealth. Biological, psychological, emotional, mental and socialhealth. Riskybehavior. General, selective and indicatedprevention. Primary, secondary, tertiaryprevention. Dependencies and types. The schoolriskfactors. The healthylifestyle. Calorie-balance. Mentalhealthconditions. School-basedpreventionprograms. Relaxation. Presentation and tapasztalatsere.	
<b>Literature:</b> Bagdy Emőke: Személyiségfejlesztő módszerek az iskolában. Budapest : Nemzeti Tankönyvkiadó. 2002. 308 s. ISBN 9631922359. Bagdy Emőke. Pszichofitness. Budapest :ANIMULA, 2003.104 s. ISBN 9634080502 Buda Béla: A mentálhigiéné szemléleti és gyakorlati kérdései. Budapest : ANIMULA. 2002. 384 s. ISBN 963 05 2412 Labáth Vladimír: Expoprogram. Bratislava : Psychodiagnostika. 1991. 198 s. Metodické pokyny. www.statpedu. sk	
<b>Language, knowledge of which is necessary to complete a course:</b> Hungarian or Slovak Language	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 873	

A	B	C	D	E	FX
35.05	24.86	19.13	7.45	13.06	0.46
<b>Teacher:</b> PaedDr. Beáta Kiss					
<b>Date of last update:</b> 25.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University					
<b>Name of the faculty:</b> Faculty of Education					
<b>Code:</b> KPD/SZdb/ SVZ/15		<b>Name:</b> Socio-Scientific and pedagogical-psychological basis of teaching			
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> <b>Recommended extent of course ( in hours ):</b> <b>Per week: For the study period:</b> <b>Methods of study:</b> present					
<b>Number of credits:</b> 2					
<b>Recommended semester/trimester of study:</b>					
<b>Level of study:</b> I.					
<b>Prerequisites:</b> KPD/SZdb/VDP/15 and KPD/SZdb/ZVP/15 and KPD/SZdb/TEV/15 and KPD/SZdb/VPS/15 and KPD/SZdb/DID/15 and KPD/SZdb/SCP/15 and KPD/SZdb/FVV/15 and KPD/SZdb/LAD/15 and KPD/SZdb/ANA/15 and KPD/SZdb/PX1/15 and KPD/SZdb/SMP/15 and KPD/SZdb/APK/15					
<b>Conditions for passing the subject:</b> The student's answer verbal subjects which are of pedagogical and psychological foundations that evaluated examination committee. Evolution: A – 90 -100%, B – 80 -89%, C – 70 -79%, D – 60 - 69%, E – 50 -59%.					
<b>Results of education:</b> Graduated from the Department Teaching academic subjects through common sociálnovedného, pedagogical and psychological basis for teachers to acquire knowledge of the problems of educational sciences and social and legislative context of education and training and the basics of digital, psychological and special pedagogical literacy teacher.					
<b>Brief syllabus:</b> x					
<b>Literature:</b> The compulsory and elective subjects is given subject data sheets.					
<b>Language, knowledge of which is necessary to complete a course:</b> Hungarian or Slovak Language					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 261					
A	B	C	D	E	FX
26.05	23.37	22.61	15.71	10.34	1.92
<b>Teacher:</b>					
<b>Date of last update:</b> 25.05.2023					

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ TEV/15	<b>Name:</b> Theory of education
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%	
<b>Results of education:</b> The main goal of the subject is to transfer knowledge to the students about the mission of education, trends, to learning theoretical concepts in a historical context and the acquisition of basic skills of pedagogical thinking.	
<b>Brief syllabus:</b> Education tasks and aims. Reflexív- science theories before. Pragmatic-behavioral theory. Cognitive - behavioral theory. Humanistic theory-accrual of persona. Multimedia information-theory.	
<b>Literature:</b> Bábosik István. Nevelélmélet. - Budapest : Osiris Kiadó, 2004. - 615 s. - ISBN 963389655x. Budai Ágnes. Nevelélmélet gyakorlatközelben : A Majzik-jelenség. - 1. vyd. - Budapest : Műszaki Könyvkiadó, 2005. - 115s. - ISBN 963 16 4041 8. Péter Lilla. Nevelélméleti alapkérdések. - 1. vyd. - Kolozsvár : Kolozsvári Egyetemi Kiadó, 2008. - 203 s. - ISBN 978-973-610-738-2. Zelina Miron. Teórie výchovy alebo Hľadanie dobra. - 2. vyd. - Bratislava : SPN, 2010. - 232 s. - ISBN 978-80-10-01884-0. Pukánszky Béla. Iskola és pedagógusképzés. - 1. vyd. - Budapest : Gondolat Kiadó, 2014. - 182 s. - ISBN 9789636932282. Pukánszky Béla. A gyermekkor története. - 1. vyd. - Budapest : Műszaki Könyvkiadó, 2001. - 201s. - ISBN 963 16 2782 9. Pukánszky Béla. Két évszázad gyermekei : A tizenkilencedik-huszedik század gyermekkorának története. - 1. vyd. - Budapest : Eötvös József Könyvkiadó, 2003. - 308 s. - ISBN 963 9316 65 2. Vajda Zsuzsanna, Kósa Éva. Neveléslélektan. - 1. vyd. - Budapest : Osiris Kiadó, 2005. - 564 s. - ISBN 963 389 728 9. - ISSN 1218-9855.	
<b>Language, knowledge of which is necessary to complete a course:</b>	

Hungarian or Slovak Language					
<b>Notes:</b>					
<b>Evaluation of subjects</b>					
Total number of evaluated students: 749					
A	B	C	D	E	FX
27.64	25.77	21.23	14.69	9.61	1.07
<b>Teacher:</b> prof. Dr. Attila Józsefné Katalin Ambrus, DSc.					
<b>Date of last update:</b> 25.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ VDP/15	<b>Name:</b> General education and history education
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week: 2 For the study period: 26</b> <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%.	
<b>Results of education:</b> The students will receive a brief overview of the history of education, taxonomy, pedagogical concepts, and the laws of pedagogy.	
<b>Brief syllabus:</b> Introduction to the history of pedagogy. Education in ancient Greece, Egypt, Athens, and Sparta. Democritos, Socrates, Plato, Aristotle. Hellenic era, Roman Empire. Education in feudalism, the early Middle Ages. Comenius, Locke, Rousseau, Pestalozzi, Tesedík, Lehotsky,. The history of education in Slovakia. The new education movement. Educational theories. The approach of Bertrand. Pragmatic-behavioral, cognitive-scientific, humanistic, and personalist trends. Pedagogical models, their analysis and importance in today's educational practice. Patterns of educational situations. The practical application of educational theory. Compilation of evaluation scales, introduction of the "rating". Monitoring methodology and its analysis in the classroom.	
<b>Literature:</b> Slávka Hlásna, Kinga Horváthová, Martin Mucha, Renáta Tóthová. Úvod do pedagogiky / - 1. vyd. - Nitra : ENIGMA, 2006. - 356 s. - ISBN 80-89132-29-4. Švecová Valéria. Základy pedagogiky. Technická univerzita v Košiciach, 1998. - 124 s. - ISBN 80-7099-323-5. Prucha Jan. Moderní pedagogika. - 4. vyd. - Praha : Portál, 2009. - 481 s. - ISBN 978-80-7367-503-5. Zelina, Miron. Teórie výchovy alebo Hľadanie dobra. - 2. vyd. - Bratislava : SPN, 2010. - 232 s. - ISBN 978-80-10-01884-0. Kasper Tomáš, Kasperová, Dana. Dějiny pedagogiky. - 1. vyd. - Praha : Grada Publishing, 2010. - 224 s. - ISBN 978-80-247-2429-4. Pukánszky Béla. A magyar iskolatörténet és pedagógusképzés paradigmái. - 1. vyd. - Komárno : Univerzita J. Selyeho, 2014. - 119 s. - ISBN 978-80-8122-096-8.	
<b>Language, knowledge of which is necessary to complete a course:</b> Hungarian or Slovak Language	



<b>Notes:</b>					
<b>Evaluation of subjects</b>					
Total number of evaluated students: 1275					
A	B	C	D	E	FX
33.57	31.84	22.75	8.08	3.76	0.0
<b>Teacher:</b> prof. Dr. Béla István Pukánszky, DSc., prof. Dr. Attila Józsefné Katalin Ambrus, DSc.					
<b>Date of last update:</b> 25.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ VPS/15	<b>Name:</b> Developmental psychology
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 2.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%.	
<b>Results of education:</b> Student will learn fylogenetetic and ontogenetic development patterns, the characteristics of the developments periods focused to students.	
<b>Brief syllabus:</b> History and main trends of developmetnalpsychology. Developmentalperiodizationas per differentauthors (L. Nagy, S. Freud, Erikson, J. Piaget) and itscomparation. Psychicaldevelopmetnindifferentages: prenatal, natal, postnatal, pre-schoolage, schoolage, teenage, adolescence. Adult life periods: early, middle and matureadult, senior life and death. Developmentspecifics a ser theircharacteristics: optimal, slowed, late, pathological and disharmonical.	
<b>Literature:</b> Atkinson L. Rita: Pszichológia. Budapest : Osiris Kiadó, 2005. 852 s. ISBN 9633897130. Bordás Sándor, Forró Zsuzsa, Németh Margit, Stredl Terézia: Pszichológiai jegyzetek. 3. vyd. Komárom : Valeur s.r.o. 2009. 320s. ISBN 9788089234851 Cole Michael: Fejlődéslélektan. Budapest : Osiris Kiadó, 2003. 810 s. ISBN 9633894735 Erényi Tibor at all.: Freud, avagy a modern individuum felfedezése. Budapest : Napvilág, 1997. 98. ISBN 9639082015 Mérei Ferenc - Binet V. Ágnes: Gyermeklélektan. Budapest : Medicina Könyvkiadó, 2006. 303 s. ISBN 963 226 027 9 Inhelder Barbel, Jean Piaget: A gyermek logikájától az ifjú logikájáig : A formális műveleti struktúrák kialakulása. Budapest : Akadémiai Kiadó. 1984. 336 s. ISBN 963 05 3642 0. Zelina Miron: Stratégie a metódy rozvoja osobnosti : Metódy výchovy. 2. vyd. Bratislava : Iris. 1996. 234 s. ISBN 8096701347	
<b>Language, knowledge of which is necessary to complete a course:</b> Hungarian or Slovak Language	
<b>Notes:</b>	

<b>Evaluation of subjects</b>					
Total number of evaluated students: 1204					
A	B	C	D	E	FX
14.7	17.11	25.58	30.56	10.96	1.08
<b>Teacher:</b> PaedDr. Terézia Strédl, PhD.					
<b>Date of last update:</b> 25.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/SZdb/ ZVP/15	<b>Name:</b> Fundamentals of General Psychology
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Final test. Condition for successful completion of this course is to obtain at least 50% of the maximum possible assessment of the subject. Evaluation: A - 90 -100%, B - 80% -89 C - -79% 70, D - 60-69%, E - 50 -59%.	
<b>Results of education:</b> The goal is to clarify the basic theoretical knowledge of general psychology and to bring psychology as a scientific discipline in terms of its historical development, research and theories. Mastering this knowledge is necessary not only for the management of other psychological disciplines, but also for understanding the functioning mechanisms of the human psyche. Student after completion of the course: can define individual psychological concepts such as memory, thinking, language, etc., knows the functioning mechanisms of cognitive, emotional and motivational processes, identifies various psychological approaches examining the psyche of the individual, their specifics and can apply his knowledge to solve practical problems in various areas of social life, but especially in educational practice.	
<b>Brief syllabus:</b> 1. Introduction 2. Main goals and methodology 3. Nature and nurture, neuropsychology 4. Sensation and perception 5. Thinking 6. Language and communication 7. Memory 8. Learning 9. Emotions 10. IQ and creativity 12. Motivation 12. Personality 13. Coping	
<b>Literature:</b> Atkinson L. Rita: Pszichológia. Budapest : Osiris Kiadó, 2005. 852 s. ISBN 9633897130. Bordás Sándor, Forró Zsuzsa, Németh Margit, Stredl Terézia: Pszichológiai jegyzetek. 3. vyd. Komárom : Valeur s.r.o., 2009. 320s. ISBN 9788089234851 Bugán A., Pléh Cs: Fejezetek a pszichológia alapterületeiből. Budapest : ELTE Eötvös Kiadó, 2000. 408 s. ISBN 9634633838 Pléh Csaba: A lélektan története. 2. vyd. Budapest : Osiris Kiadó, 2010. 652 s. ISBN 978 963 276 0520 Pléh Csaba, Boross Ottilia: Akadémiai lexikonok - Pszichológia : A pszichológia legfontosabb fogalmai magyar és angol nyelven. 1. vyd. Budapest : Akadémiai Kiadó, 2010. 403 s. ISBN 978 963 8658 0	

<b>Language, knowledge of which is necessary to complete a course:</b> Hungarian or Slovak Language					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 1439					
A	B	C	D	E	FX
10.35	16.68	20.92	21.06	25.5	5.49
<b>Teacher:</b> Mgr. Anita Tóth-Bakos, PhD.					
<b>Date of last update:</b> 25.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ ULGdb/NJ/15	<b>Name:</b> Introduction into Linguistics
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 4	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> The final test takes place in the form of a written test, the conditions for the granting of credits is to obtain min. 65% points from the test.	
<b>Results of education:</b> The student knows the basic terms and concepts of Germanic linguistics. He has knowledge of the individual linguistic levels and linguistic disciplines.	
<b>Brief syllabus:</b> 1. The languages of the earth 2. The genetic relationships between languages 3. The typological relations between languages 4. Terminology and branches of linguistics, the levels of language 5. What is language? 6. Fundamentals of Communication Theory 7. Functional models of language 8. Speech acts and speech acts 9. Basic concepts and issues of phonetics 10. Basic concepts and problems of phonology 11. Basic concepts and issues of morphology 12. Basic concepts and issues of syntax 13. Basic concepts and issues of semantics and pragmatics 14. Basic concepts and problems of lexicology 15. Basic concepts and problems of text linguistics	
<b>Literature:</b> BUSCH, A./STENSCHKE, O.: Germanistische Linguistik. Tübingen: Gunter Narr, 2007. BUßMANN, H.: Lexikon der Sprachwissenschaft. Stuttgart: Kröner, 2002. ERNST, P. (Hrsg.): Einführung in die synchrone Sprachwissenschaft. Wien: WUV, 1999. FORGÁCS, E.: Einführung in die Sprachwissenschaft. Ein Studienbuch für ungarische Germanistikstudenten. Szeged: JGYF, 2004 HOFFMANN, L. (Hrsg.): Sprachwissenschaft. Ein Reader. 2., verbesserte Aufl. Berlin/New York: De Gruyter, 2000.	

LINKE, A./NUSSBAUMER, M./PORTMANN, P. R.: Studienbuch Linguistik. 2. Aufl. Tübingen: Niemeyer, 1994. MÜLLER, H. M.: Arbeitsbuch Linguistik. Paderborn: Schöningh, 2002.					
<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 163					
A	B	C	D	E	FX
33.74	15.95	26.38	12.27	11.04	0.61
<b>Teacher:</b> Dr. phil. Mgr. Attila Mészáros					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/USLdb/ NJ/15	<b>Name:</b> Introduction into Literature Science
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 3	
<b>Recommended semester/trimester of study:</b> 1.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> The final test takes place in the form of a written test, the conditions for the granting of credits is to obtain min. 65% points from the test.	
<b>Results of education:</b> The students acquire the terminology and methodology, which is necessary for the understanding of literary texts. They know the basic issues and methods of literary studies. They are able to apply the acquired theoretical knowledge in the practical analysis of literary texts.	
<b>Brief syllabus:</b> Literature and Literary Studies: identifying two basic concepts; Poetics and rhetoric; Stylistics and metrics; Problems of literary types and genres. Epic: epic genres and their characteristics, the analysis of the epic text, lyric: metrics, rhyme, rhythm, lyric genres and their characteristics, the analysis of poetry, drama: the elements of drama, dramatic genres and their characteristics, dramatic effect Methods of literature, rhetorical figures, basics of narrative texts, the concept of immanent interpretation and evaluation of literary canon.	
<b>Literature:</b> HORN, András: Theorie der literarischen Gattungen. - 1. vyd. - Würzburg: Königshausen & Neumann 1998. - 210s. - ISBN 3-8260-1544-4. STANZEL, Franz Karl: Theorie des Erzählens. - 1. vyd. - München: UTB Vandenhoeck, 1991. - 340 s. - ISBN 3-525-03208-0. ASMUTH, Bernhard: Einführung in die Dramenanalyse. - 7. vyd. - Stuttgart: Metzler 2009. - 235s. - ISBN 978 3 476 17188 7. MEYER-KRENTLER, Eckhardt: Arbeitstechniken Literaturwissenschaft. - 1. vyd. - München: Wilhelm Fink Verlag 1994. - 143s. - ISBN 3-8252-1582-2. VOGT, Jochen: Einladung zur Literaturwissenschaft. - 4. vyd. - München: Fink 2002. - 288. - ISBN 3825220729. LUSERKE, Matthias (ed.): Literaturwissenschaft – Kulturwissenschaft. Positionen, Themen, Perspektiven. - 1. vyd. - Opladen: Westdeutscher Verlag 1996 - 215 s. - ISBN 3-531-22171-X. ( <a href="http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00041419_00001.html?zoom=1.00">http://digi20.digitale-sammlungen.de/de/fs1/object/display/bsb00041419_00001.html?zoom=1.00</a> )	



<b>Language, knowledge of which is necessary to complete a course:</b> German					
<b>Notes:</b>					
<b>Evaluation of subjects</b> Total number of evaluated students: 173					
A	B	C	D	E	FX
10.4	17.34	26.01	21.97	18.5	5.78
<b>Teacher:</b> Dr. habil. László V. Szabó, PhD.					
<b>Date of last update:</b> 09.06.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ UVPdb/NJ/15	<b>Name:</b> Introduction into Scientific research
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 <b>For the study period:</b> 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Development of a portfolio (term paper, 60%) + a glossary at the given topic (40%)	
<b>Results of education:</b> The students acquire basic knowledge about the content, method and formal aspects of scientific work. They understand the basics of scientific ethics. They can work with the material, interpreting the material and process further. They can apply the acquired knowledge when writing their own work.	
<b>Brief syllabus:</b> Study of theoretical literature Creation of terminology Creating corpora Creation of excerpts Empirical Methods. Statistics. Structural work Evaluation of the results obtained Conclusions	
<b>Literature:</b> ESSELBORN-KRUMBIEGEL, H.: Von der Idee zum Text. Eine Anleitung zum wissenschaftlichen Schreiben. 3., überarb. Aufl. Paderborn: Ferdinand Schöningh, 2008. LANGER, I./VON THUN, F. S./TAUSCH, R.: Sich verständlich ausdrücken. 8. Aufl. München/ Basel: Ernst Reinhardt Verlag, 2006. MEYER-KELLER, E.: Arbeitstechniken. Literaturwissenschaft. 4. Aufl. München: Wilhelm Fink, 1994. POENICKE, K.: Wie verfaßt man wissenschaftliche Arbeiten? Ein Leitfaden vom ersten Studiensemester bis zur Promotion. 2., neu bearb. Aufl. Mannheim: Dudenverlag, 1988. ROTHMANN, K.: Anleitung zur Abfassung literaturwissenschaftlicher Arbeiten. Arbeitstexte für den Unterricht. Stuttgart: Reclam, 1979.	
<b>Language, knowledge of which is necessary to complete a course:</b> German	

<b>Notes:</b>					
<b>Evaluation of subjects</b>					
Total number of evaluated students: 25					
A	B	C	D	E	FX
88.0	4.0	0.0	0.0	8.0	0.0
<b>Teacher:</b>					
<b>Date of last update:</b> 17.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KPD/Uzb/ PPC1a/23	<b>Name:</b> Supporting pedagogical practice 1
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Practical <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 20 <b>For the study period:</b> 260 <b>Methods of study:</b> present	
<b>Number of credits:</b> 1	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> The conditions for completing the course: <ul style="list-style-type: none"> <li>- active participation of the student in a teaching practice in a primary school (primary school) or a secondary school (secondary school),</li> <li>- participation of the student in assigned tasks and involvement in analysis and discussion during the teaching practice,</li> <li>- submission of a completed and validated PPC1 completion report,</li> <li>- Completion of observation sheets from the teaching practice in primary or secondary school: lesson observation records,</li> <li>- Student reflection on PPC1a.</li> </ul> Evaluation of the submitted documents (max. 50 points): <ul style="list-style-type: none"> <li>o Content page 35 points,</li> <li>o formal aspect 15 points.</li> </ul> Total student workload: 1 credit = 30 hours <ul style="list-style-type: none"> <li>- 20 hours of participation in the teaching practice (contact hours): of which 10 hours of hospitalization and 10 hours of analysis; 2 hours of introductory meeting; 8 hours of preparation of observation sheets and reflection.</li> </ul> Final assessment: <ul style="list-style-type: none"> <li>- passed = 50 - 100% (25 - 50 points)</li> <li>- not passed = 49 - 0% (0 - 24 points)</li> </ul>	
<b>Results of education:</b> <b>Knowledge:</b> <ul style="list-style-type: none"> <li>- The student is competent to observe lessons in elementary and middle school.</li> <li>- The student is able to document observed lessons in grade 2 elementary and middle school.</li> <li>- The student is able to navigate some school documents.</li> </ul> <b>Skills:</b> <ul style="list-style-type: none"> <li>- The student is able to identify diverse manifestations of structural elements of personality, psychological processes of the student in the teaching process and in social interactions.</li> <li>- The student will describe the didactic aids, communication technologies and means used in the teaching process and the possibilities of applying computers, interactive whiteboards, the</li> </ul>	

Internet, specific teaching programmes and software, dynamic systems and interactive teaching materials and portals in teaching subjects at the 2nd level of primary and secondary school.  
- It identifies teachers' teaching and communication styles and professional skills.

Competencies:

- The student is able to conceive his/her own work practices for effective observation.
- Takes a position on observed phenomena based on prior theoretical knowledge.
- Understands the relationship between the principles of teaching and the consequences - the effectiveness of learning.

### **Brief syllabus:**

Basic attributes of observation.

Observation and evaluation of the interior and exterior of a training primary and secondary school.

Recognition and work with pedagogical documentation of the classroom.

Observation of lessons in a 2nd grade elementary school and an SHS.

Analysis of observed lessons together with the trainee teacher.

Documenting the progress of each lesson observed.

Structure of observation sheets.

Completion of observation sheets.

### **Literature:**

Štátny vzdelávací program pre 2. stupeň základnej školy v Slovenskej republike ISCED 2 – nižšie sekundárne vzdelávanie. [https://www.statpedu.sk/files/articles/dokumenty/statny-vzdelavaci-program/isced2\\_spu\\_uprava.pdf](https://www.statpedu.sk/files/articles/dokumenty/statny-vzdelavaci-program/isced2_spu_uprava.pdf)

Štátny vzdelávací program pre gymnázia v Slovenskej republike

ISCED 3A – Vyššie sekundárne vzdelávanie. [https://www.statpedu.sk/files/articles/dokumenty/statny-vzdelavaci-program/isced3\\_spu\\_uprava.pdf](https://www.statpedu.sk/files/articles/dokumenty/statny-vzdelavaci-program/isced3_spu_uprava.pdf)

Zákon č. 245/2008 Z. z. – Zákon o výchove a vzdelávaní (školský zákon) a o zmene a doplnení niektorých zákonov. Bratislava : MŠ SR, 2008 (respektíve aktuálny školský zákon).

Aktuálny vnútorný predpis UJS: Zásady realizácie pedagogickej praxe na Pedagogickej fakulte UJS

Gadušová, Z. a kol.: Mentor Training : Ostrava : Ostravská univerzita, 2021. - online, 268 s. - ISBN 978-80-7599-294-9.

### **Language, knowledge of which is necessary to complete a course:**

hungarian, slovak

### **Notes:**

### **Evaluation of subjects**

Total number of evaluated students: 98

A	B	C	D	E	FX
100.0	0.0	0.0	0.0	0.0	0.0

**Teacher:** Mgr. Katarína Szarka, PhD., PaedDr. Tamás Török, PhD.

**Date of last update:** 29.11.2023

**Approved by:** prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ZDI db/ NJ/15	<b>Name:</b> Introduction into Didactics of German as Foreign Language
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 2 <b>For the study period:</b> 26 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 5.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Written final test	
<b>Results of education:</b> The student has knowledge of the historical development of teaching methods. He identifies their characteristics. He can assess the advantages and disadvantages of each method. He knows the basic concepts and principles and can apply the methods in the teaching process.	
<b>Brief syllabus:</b> Methods and procedures as Categories Basic questions of didactics and methodology of foreign language teaching The grammar-translation method The direct method. Building a module using the direct method. The audio-lingual method. The audio-visual method. Teaching techniques of audio-visual method. The mediating method The development of communicative didactics. The intercultural approach Universal life experiences as a bridge for intercultural issues planning of GFL teaching. The pragmatic-functional and cross-cultural concept.	
<b>Literature:</b> BAUSCH., K.R. (Hrsg.): Handbuch Fremdsprachenunterricht. Stuttgart: UTB, 2003. DALHAUS, B.: Fertigkeit Hörverstehen. München: Langenscheidt, 1995. HELBIG, G./GÖTZE, L./HENRICI, G./KRUMM, H-J. (Hrsg.): Deutsch als Fremdsprache. Berlin (HSK; 19.1-2), 2001. KAST, B.: Fertigkeit Schreiben. München: Langenscheidt, 1995. NEUF-MÜNKEL, G./ROLAND, R.: Fertigkeit Sprechen. München: Langenscheidt, 1994. NEUNER, G./HUNFELD, H.: Methoden des fremdsprachlichen Deutschunterrichts. München: Langenscheidt, 1993. WESTHOFF, G.: Fertigkeit Lesen. München: Langenscheidt, 1997.	
<b>Language, knowledge of which is necessary to complete a course:</b> German	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 27	

A	B	C	D	E	FX
48.15	18.52	7.41	14.81	11.11	0.0
<b>Teacher:</b> Mgr. Erzsébet Szabó, PhD.					
<b>Date of last update:</b> 17.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					

## INFORMATION SHEET

<b>Name of the university:</b> J. Selye University	
<b>Name of the faculty:</b> Faculty of Education	
<b>Code:</b> KMF/ZTRdb/ NJ/15	<b>Name:</b> Translating Studies
<b>Types, range and methods of educational activities:</b> <b>Form of study:</b> Lecture / Seminar <b>Recommended extent of course ( in hours ):</b> <b>Per week:</b> 1 / 1 <b>For the study period:</b> 13 / 13 <b>Methods of study:</b> present	
<b>Number of credits:</b> 2	
<b>Recommended semester/trimester of study:</b> 6.	
<b>Level of study:</b> I.	
<b>Prerequisites:</b>	
<b>Conditions for passing the subject:</b> Condition of the granting of credit points: Development of a translation portfolio and an HU-DE / SK-DE glossary to a particular theme (40% of total points); Translation of a popularizing / technical text (DE-HU / DE-SK) (60% of total points).	
<b>Results of education:</b> The students acquire the basics of Translation Studies. They know the basic terms and the basic theories. They are able to apply the translational techniques in practice. They are able to identify cultural specifics of the source text and to transfer this adequately into the target language.	
<b>Brief syllabus:</b> Introduction to the problems of translation - the translation process, the basic concepts. Translation as an intercultural process. Typology of literary genres, text functions. Translation techniques. Research work, work with different types of dictionaries and online resources.	
<b>Literature:</b> AMMON, U.; BICKEL, H.; EBNER, J. (Hrsg.): Variantenwörterbuch des Deutschen. Berlin: Gruyter Verlag, 2004. DUBOVÁ, J.: Übersetzen nichtliterarischer Texte (URL: <a href="http://www.knj.upol.cz/materialy/preklad.doc">http://www.knj.upol.cz/materialy/preklad.doc</a> ) SNELL-HORNBY, M; HÖNIG, H. G.; KUßMAUL, P.; SCHMITT, P. A. (Hrsg.): Handbuch Translation. Unveränderter Nachdruck der 2. Auflage. Tübingen: Stauffenburg Verlag, 2006.	
<b>Language, knowledge of which is necessary to complete a course:</b> German	
<b>Notes:</b>	
<b>Evaluation of subjects</b> Total number of evaluated students: 27	



A	B	C	D	E	FX
44.44	14.81	22.22	7.41	11.11	0.0
<b>Teacher:</b> Dr. phil. Mgr. Attila Mészáros					
<b>Date of last update:</b> 17.05.2023					
<b>Approved by:</b> prof. Dr. Erzsébet Komlósiné Knipf, DSc., prof. Dr. Béla István Pukánszky, DSc., doc. RNDr. Róbert Gyepes, PhD.					